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# Spectroscopic Estimation of Pioglitazone Hydrochloride

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**Abstract** - A simple, sensitive, accurate, precise, reproducible and cost effective UV spectroscopic method has been developed for the estimation of Pioglitazone hydrochloride in bulk and tablet dosage form. Pioglitazone hydrochloride shows maximum absorption at 269 nm with molar absorptivity of  $9.6013 \times 10^4$  l/mol.cm. Beer's law was obeyed in the concentration range of 10-70 µg/ml. The method was validated for linearity, precision, accuracy, sensitivity and specificity. The data obtained was treated with the statistical approach. The proposed method was found to be accurate and precise for estimation of Pioglitazone hydrochloride in bulk and tablet dosage form.

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## I. INTRODUCTION

Analysis of the drug is the most important aspect of formulation development. A suitable method is essential for the estimation of bulk drug, of the drug in formulation, in dissolution studies and in biological samples.

Chemically Pioglitazone hydrochloride (PH) is 5-(4-[2-(5-ethylpyridin-2-yl)ethoxy]benzyl)thiazolidine-2,4-Dione. (Figure 1) [1] It belongs to the Class II of the BCS system and is practically insoluble in water but soluble in organic solvents like methanol, dimethyl sulfoxide and dimethylformamide. [1] It selectively stimulates the nuclear peroxisome proliferator-activated gamma receptor (PPAR-γ) and to a lesser extent peroxisome proliferator-activated alpha receptor (PPAR-α). It is primarily used in the treatment of diabetic conditions in combination or singly. It modulates the transcription of the insulin-sensitive genes involved in the control of glucose and lipid metabolism in the muscle, adipose tissue and the liver. As a result, it reduces insulin resistance in the liver and peripheral tissues, increases the expense of insulin-dependent

glucose, decreases withdrawal of glucose from the liver, and reduces quantity of glucose. [2]

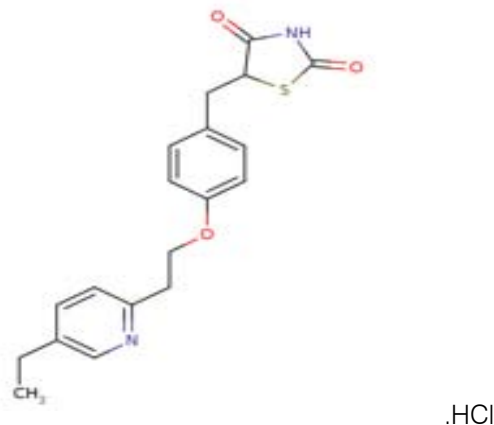


Figure 1: Chemical Structure of Pioglitazone hydrochloride

According to literature review, a HPLC method for the estimation of Pioglitazone hydrochloride is available. [3] The method is relatively complex and expensive. The UV method for estimation of Pioglitazone hydrochloride in methanol: water: hydrochloric acid (250:250:1) system [4] and in 0.2 M sulphuric acid [5] have been reported. However, quantitative estimation of PH in other media has not been reported. This is essential in drug release study. The objective of the study was to develop a simple, accurate, precise, cost effective and reproducible UV method for estimation of PH in 0.1N hydrochloric acid as per ICH guidelines. [6]

## II. MATERIALS AND METHODS

Shimadzu UV/Visible double beam spectrophotometer and a Jasco V-630 instrument with 1 cm matched quartz cells were used for the spectral measurement. Shimadzu AX200 analytical balance was used for the weighing purpose. The reference standard of PH was obtained as a gift sample from Aarti Drugs, Thane (India) with 99.8% assay value. PH tablets (Piomed, 15 mg) were obtained from the market and utilized for the study. All other chemicals were of analytical grade.

### a) Selection of The Media

The criterion for selection of the medium was the solubility and the stability, i.e. PH should be soluble

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and stable in the medium for sufficient time during the study. Methanol: water: hydrochloric acid (250:250:1) was used in the reported method. Dissolution studies recommend 0.1N hydrochloric acid. Hence 0.1N hydrochloric acid was selected as the analytical medium for the present work.

#### b) Preparation of Standard Solution

Standard solution of PH was prepared by dissolving 100 mg of drug in 100 ml of 0.1N hydrochloric acid (Solution A, 1000  $\mu\text{g/ml}$ ). Further 10 ml of the solution A was diluted to 100 ml with 0.1N hydrochloric acid (Solution B, 100  $\mu\text{g/ml}$ ). Solution B was used as the standard stock solution.

#### c) Preparation of Calibration Curve

Aliquots of 1 ml to 7 ml of the standard solution B were transferred into a series of calibrated 10 ml standard volumetric flasks and the final volume was made up using 0.1N hydrochloric acid. The solutions were scanned in the range of 200-400 nm against blank

(0.1N hydrochloric acid). The absorption maximum was found to be at 269 nm. (Figure 2) The absorbance of the solutions were measured at 269 nm against the blank (Table 1) and the calibration curve was constructed. (Figure 3)

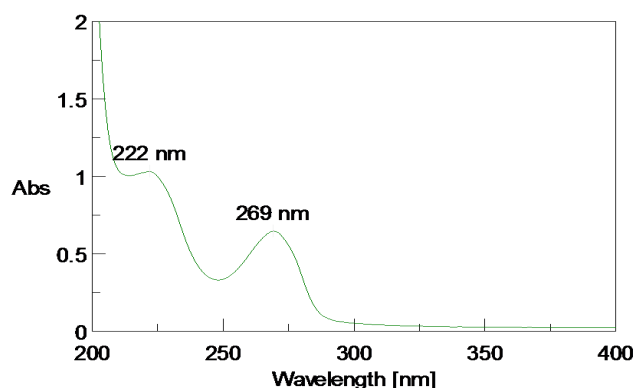


Figure 2 : Pioglitazone hydrochloride absorbance spectrum

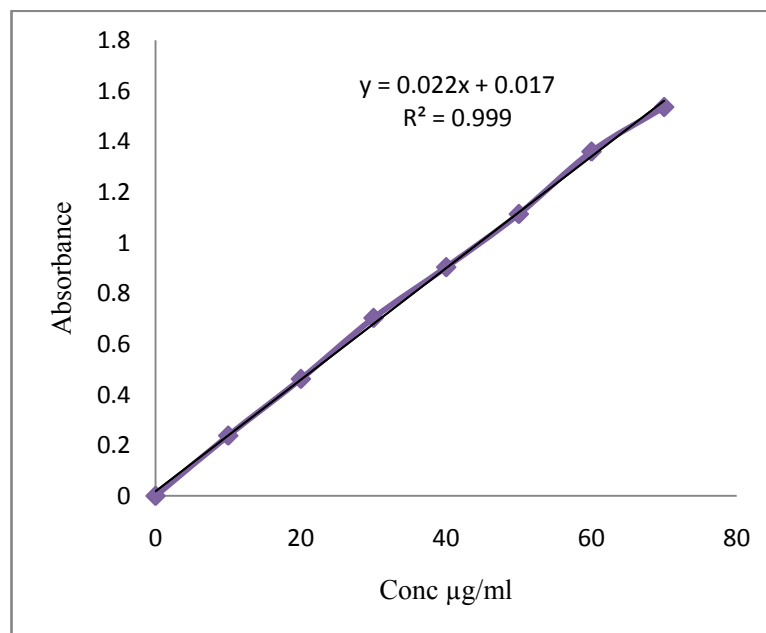


Figure 3 : Calibration Curve of Pioglitazone hydrochloride

Table 1: Calibration Curve of Pioglitazone hydrochloride

Sr. no	Concentration ( $\mu\text{g/ml}$ )	Absorbance	Standard deviation
1	0	0	0
2	10	0.2380	$\pm 0.003551$
3	20	0.4620	$\pm 0.003404$
4	30	0.7385	$\pm 0.003593$
5	40	0.9034	$\pm 0.002524$
6	50	1.1134	$\pm 0.000917$
7	60	1.3600	$\pm 0.001000$
8	70	1.5359	$\pm 0.002571$

The optical characteristics were summarized. (Table 2)

*Table 2* : Validation Parameters

Sr no	Parameter	Result
1.	Absorption maxima	269 nm
2.	Linearity range	10-70 $\mu\text{g/ml}$
3.	Standard Regression Equation	$y = 0.022x + 0.017$
4.	Correlation coefficient ( $r^2$ )	0.999
5.	Molar Absorptivity	$9.6013 \times 10^4 \text{ l/mol.cm}$
6.	A (1%, 1 cm)	244.328 dl/gm/cm
7.	Accuracy (% recovery $\pm$ S.D)	$99.3233 \pm 0.7026$
8.	Specificity	A 20 $\mu\text{g/ml}$ of drug in 0.1 N HCl at UV detection wavelength of 269 nm shows an absorbance value of $0.4620 \pm 0.003404$
9.	LOD ( $\mu\text{g/ml}$ )	0.03
10.	LOQ ( $\mu\text{g/ml}$ )	0.10

#### d) Preparation of Sample Solution

The proposed method was applied to marketed PH tablets (Piomed, 15 mg). Twenty tablets of PH were weighed and powdered in a glass mortar. Powder equivalent to 100 mg of the drug was weighed accurately and transferred to a 100 ml standard volumetric flask. It was dissolved in about 50 ml of 0.1N hydrochloric acid and the volume was made up with 0.1N hydrochloric acid so that the concentration was 1000  $\mu\text{g/ml}$  (Solution P). Ten ml of the solution P was transferred to a 100 ml standard volumetric flask and the volume was adjusted with 0.1N hydrochloric acid (Solution Q). The solution was filtered through Whatmann filter paper no. 41. The filtrate was diluted suitably with 0.1N hydrochloric acid to obtain a sample solution (20  $\mu\text{g/ml}$ ). The absorbance of the sample solution was measured at 269 nm and the amount of PH was determined from the calibration curve. The method was studied for accuracy and precision.

### III. RESULTS AND DISCUSSION

#### a) Linearity

Pioglitazone hydrochloride exhibited maximum absorption at 269 nm and obeyed Beer's Law in the range of 10-70  $\mu\text{g/ml}$ . [8, 10] Linear regression of absorbance Vs concentration yielded equation  $y = 0.022x + 0.017$  with a correlation coefficient of 0.999.

#### b) Accuracy

To determine the suitability and reproducibility of the proposed method, recovery studies were carried

out by adding known amount of standard PH (80%, 100%, and 120%) to the tablet solution P and analyzing the mixtures by the proposed method. Three samples were prepared for each recovery level. The percentage recovery of PH was found to be  $99.3233 \pm 0.7026$  (Table 3) indicating that there is no interference by the excipients in the method. According to ICH guidelines, an acceptance criterion for the % recovery is 98-102%.

*Table 3* : Accuracy by recovery method

Ingredient	Pioglitazone hydrochloride		
Tablet amount ( $\mu\text{g/ml}$ )	20	20	20
Level of addition (%)	80	100	120
Amount added ( $\mu\text{g/ml}$ )	16	20	24
Amount recovered ( $\mu\text{g/ml}$ )	35.748	39.574	44.1288
% Recovery	99.3000	98.9350	100.2927
Average % recovery	$99.3233 \pm 0.7026$		

#### c) Precision

Precision of the method was demonstrated by intra-day and inter-day variation studies. For intra-day precision, six sample solutions of Pioglitazone hydrochloride of same concentration (20  $\mu\text{g/ml}$ ) were analyzed three times in a day. The result is indicated by % RSD in Table 4.

During the intermediate precision (inter-day precision), six sample solutions of the same concentration (20  $\mu\text{g/ml}$ ) were analyzed on three consecutive days and by two different analysts in same laboratory. The results are indicated by % RSD in Table 5 and 6.

For intra- day precision, the % drug content and the relative standard deviation (RSD) were found to be  $99.958 \pm 0.7874$ ,  $99.928 \pm 1.104$ ,  $99.297 \pm 1.114$  and 1.0087 respectively; whereas for inter- day precision, the % drug content and the relative standard deviation (RSD) values  $99.59722 \pm 0.4722$ ,  $100.7488 \pm 0.4522$ ,  $100.4226 \pm 0.5617$  and 0.4940 respectively. When the analyst was changed the RSD values were 0.48225 and 0.4662. According to ICH guidelines, an acceptance criterion for the precision is  $\text{RSD} \leq 2\%$ .

**Table 4 :** Intra- day Precision for Pioglitazone hydrochloride

Sample Number	Analysis of Pioglitazone hydrochloride as percent of drug content		
	10:00 am	2:00 pm	6:00 pm
1	101.214	98.979	98.569
2	99.458	99.568	100.598
3	99.587	99.259	99.454
4	100.254	101.871	100.598
5	98.979	99.298	97.995
6	100.256	100.598	98.568
Mean $\pm$ SD	99.958 $\pm$ 0.7874	99.928 $\pm$ 1.104	99.297 $\pm$ 1.114
Average $\pm$ RSD	99.7276 $\pm$ 1.00875 1.0087		

**Table 5 :** Inter-day precision for Pioglitazone hydrochloride

Sample number	Analysis of Pioglitazone hydrochloride as percent of labeled content		
	DAY-1	DAY-2	DAY-3
1	99.8467	100.725	100.053
2	100.0230	101.146	99.8792
3	99.4538	99.9875	99.9103
4	98.9985	100.5473	101.163
5	99.1356	100.856	100.5409
6	100.1257	101.231	100.9892
Mean $\pm$ SD	99.59722 $\pm$ 0.4722	100.7488 $\pm$ 0.4522	100.4226 $\pm$ 0.5617
Average $\pm$ RSD	100.2562 $\pm$ 0.4940 0.4940		

**Table 6 :** Determination of Precision, different analysts (Intra-day precision)

Sample number	Analysis of Pioglitazone hydrochloride as percent of labeled amount	
	Analyst-I	Analyst-II
1	100.2346	99.1035
2	100.9812	99.1418
3	99.8754	98.7460
4	100.0213	99.2435
5	99.5381	97.9924
6	100.1509	98.6356
Mean	100.1335	98.8149
Std. Deviation	0.48225	0.4662

**d) Robustness**

When the analysis was carried on two different instruments, the RSD values were 0.5297 and 0.5213.

**Table 7 :** Determination of Robustness (Different Instruments)

Sample number	Analysis of Pioglitazone hydrochloride as percent of labeled content	
	Shimadzu	Jasco
1	99.184	100.231
2	98.793	101.104
3	99.862	100.863
4	100.021	99.982
5	98.795	101.016
6	99.568	99.989
Mean	99.3705	100.5308
Std Deviation	0.5297	0.5213

**e) Limit Of Detection (LOD) And Limit of Quantitation (LOQ)**

The LOD and LOQ of PH were determined by using standard deviation of the response and the slope approach as defined in the ICH Guidelines [6]. The LOD and LOQ were found to be 0.03 $\mu$ g/ml and 0.1 $\mu$ g/ml respectively. The proposed method showed molar absorptivity of  $9.6013 \times 10^4$  l/mol.cm. (Table 2)

## IV. CONCLUSION

The developed method was found to be simple, accurate, precise, reproducible and can be used for dissolution studies and routine quality control analysis of PH in bulk and in tablet form.

## V. ACKNOWLEDGEMENTS

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