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Hydroelectric Dam Penstocks

Enhancing Composite Performance

Highlights

Friction Coefficient Evolution

Theoretical vs. Numerical Methods

Discovering Thoughts, Inventing Future

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Enhancing Composite Performance: Hydrothermally Treated Wood Reinforcement in Recycled Polypropylene

By Andressa dos Santos, Raphael Leonardo Bulla, Laís Weber Aguiar,
Murilo Pereira Moises, Eduardo Radovanovic & Silvia Luciana Favaro

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Abstract- The low thermal stability of cellulose presents unique technological challenges to the formulation of wood and plastic composites that are compatible and processable. For this, hydrothermal modification is a well-established technology for improving dimensional stability and durability of wood's components, in addition to providing better interaction with the polymer. This study produced polymer composites in which hydrothermally treated wood waste fibers (WT) reinforce a recycled polypropylene (RPP) matrix. Wood waste fibers were selected by grain size and distribution, treated hydrothermally, and characterized by SEM, ATR-FTIR, and water sorption. Composites were produced varying the reaction time of treatment hydrothermal (from 30 to 180 minutes), granulometric size (from 425 to 1400 μm) and percentage of WT (from 10 to 20%), following a 2^3 full-factorial experimental design, by extrusion with internal recirculation and the mechanical test specimens were modulated by injection. Tensile, flexion, IZOD impact and water sorption tests were statistically analyzed. Reaction time was the most statistically significant factor.

Keywords: *fibers, mechanical properties, polymer, plastic composite, statistics.*

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ENHANCING COMPOSITE PERFORMANCE HYDROTHERMALLY TREATED WOOD REINFORCEMENT IN RECYCLED POLYPROPYLENE

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Enhancing Composite Performance: Hydrothermally Treated Wood Reinforcement in Recycled Polypropylene

Andressa dos Santos ^α, Raphael Leonardo Bulla ^σ, Laís Weber Aguiar ^ρ, Murilo Pereira Moises ^ω, Eduardo Radovanovic [¥] & Silvia Luciana Favaro [§]

Abstract- The low thermal stability of cellulose presents unique technological challenges to the formulation of wood and plastic composites that are compatible and processable. For this, hydrothermal modification is a well-established technology for improving dimensional stability and durability of wood's components, in addition to providing better interaction with the polymer. This study produced polymer composites in which hydrothermally treated wood waste fibers (WT) reinforce a recycled polypropylene (RPP) matrix. Wood waste fibers were selected by grain size and distribution, treated hydrothermally, and characterized by SEM, ATR-FTIR, and water sorption. Composites were produced varying the reaction time of treatment hydrothermal (from 30 to 180 minutes), granulometric size (from 425 to 1400 µm) and percentage of WT (from 10 to 20%), following a 2³ full-factorial experimental design, by extrusion with internal recirculation and the mechanical test specimens were modulated by injection. Tensile, flexion, IZOD impact and water sorption tests were statistically analyzed. Reaction time was the most statistically significant factor. Composites of wood waste fibers treated for 30 min and containing 20% of WT presented better mechanical properties than expected. However, the preservation of the lamellar fibers during the reaction time allowed for better adherence to the polymer, and the insertion of a greater quantity of fibers in the material provided greater rigidity in the composite. In general, the results obtained gives properties of stability and resistance to damage of composites containing hydrothermally treated wood fibers.

Keywords: fibers, mechanical properties, polymer, plastic composite, statistics.

1. INTRODUCTION

Environmental preservation is a crucial area in academia and industry, as both sectors can benefit from processes that recycle and reuse waste. ^[1,2] Proper management of waste disposal and reuse protects the environment and allows sustainable product development ^[3]. In accordance with Abiplast, the global average consumption of plastics has reached 35 Kg/inhabitant in the last two years ^[4]. The increase in

plastic consumption creates a significant challenge for proper waste disposal.

Most plastics are non-biodegradable polymers. Because of their low cost and versatility, plastics are consumed increasingly in urban areas. Plastic materials used in food packaging have specific thermophysical characteristics and are classified as thermoplastics or thermosetting polymers ^[5,6].

Thermoplastic materials are usually to form composites; each polymer has unique advantages and disadvantages. Polypropylene (PP) is a versatile semi-crystalline thermoplastic for general use. Its principal characteristics are high resistance to flexural ruptures, high resistance to chemicals and solvents, good electrical properties, good thermal stability, a low specific weight (0,905 g/cm³), and low cost ^[5,7]. Thermoplastic polymers are fused via heating and solidified by cooling in a reversibly process. They are soft and pliable due to weak van der Waals forces, allowing reversibility ^[8].

Polymeric materials are prized as matrices that conform at low temperatures and pressure; however, they possess lower mechanical resistance than metallic and ceramic materials [9]. In the face of the global environmental scenario, the appeal for alternative reinforcement fibers has been growing. Natural fibers have revealed several economic and sustainable development advantages, mainly due to their characteristics, which include excellent mechanical properties, low cost, low density, low abrasiveness, ease of processing, abundance and biodegradability^[10]. Fibers obtained from *Agave sisalana* (Sisal)^[11], *Sansevieria cylindrica* (Saint George's Spear)^[12], *Agave tequilana* (Blue Agave)^[13], rice husk ^[14], sugarcane bagasse ^[15] and wood fiber^[16] have shown promising results as reinforcement in composites, motivating the search for new species that show high potential as reinforcement.

Wooden residues, especially powder and bran, are primarily transported to rural areas, where they are used as ground cover or animal bedding. However, the storage and transportation of these materials is dangerous, as they are highly flammable and explosive ^[17]. Wood comprises cellulose, hemicellulose, lignin,

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and extractives in various proportions. The cellulose and hemicellulose components contain hydroxyl groups, making wood a hygroscopic material, which can expand the cellular wall and cause dimensional swelling ^[18]. Efforts have been made to improve the dimensional stability of wood and, consequently, that of the final products. However, is that the compounding of polymer matrix with the wood fiber often leads to weak mechanical properties of the composites, which are tensile, impact strengths and elongation at break. The poor mechanical properties of composites can be attributed to the low compatibility between the polar hydrophilic wood fiber and the non-polar hydrophobic polypropylene, with weak interfacial adhesion and the low dispersion of wood fiber in the polypropylene matrix due to strong interactions between of the fiber, resulting from hydrogen bonding ^[16].

In industry, thermal treatment is used most often to modify the characteristics of wood in order to improve dimensional stability ^[15]. This technique involves exposing wood to different factors, such as temperature, time, pressure, and work conditions, enhance the quality of wood for specific applications. Thermal treatment changes wood's physical, chemical, and mechanical properties due to the degradation of its principal chemical components ^[20].

CARVALHO (2015) studied the hydrothermal treatment of wood panels and observed the degradation of some chemical constituents, specifically mannans, xylans, and arabinanas, which caused acidification and loss of mass; this resulted in a consequent decrease in swelling, but did not negatively influence the quality of the panel [18]. Hydrothermal treatment stands out among the various treatments for fiber surfaces since it does not modify the chemical composition of the fibers. Hydrothermal treatment employs different reactions (extraction, hydrolysis, carbonization, liquefaction) at various temperatures, between 100–374°C ^[21]. Hydrothermal treatments do not require acid; consequently, reactors are not required to be highly resistant to corrosion, reducing the cost of this process^[22].

Herein, we describe the incorporation of hydrothermally treated wood fiber into wood and plastic composites. The treated wood fibers replace polymer matrices in recycled polypropylene to improve the composites' mechanical properties for their use in residential and commercial construction.

II. EXPERIMENTAL

a) Materials

Solid wood waste shavings (WW) extracted from *Pinus* and *Eucalyptus* woods, provided by *Madereira Altônia* located in Maringá, Paraná, Brazil, were collected, dried in an oven at 60°C for 24 hours, and triturated in a TROPP electric crusher (model TRF

750) with motor speed of 60Hz/3600rpm, equipped with 20 hammers and 2 knives. For the matrix phase, recycled polypropylene (RPP) was supplied by *Plaspet Reciclagem Maringa LTDA ME* from Maringá, Paraná, Brazil.

b) Characterization of Recycled Polypropylene

Recycled polypropylene sample, compression molding to 350mm diameter and 190mm thickness, was characterized by X-ray diffraction (XRD) using a Shimadzu XRD-7000 X-ray diffractometer equipped with a Cu K α radiation source ($\lambda = 1.5406 \text{ \AA}$) over the 2θ range 5–60° at a scan rate of 1°/min. Differential Scanning Calorimetry (DSC) was recorded on a DSC Q20 thermal analyzer from TA Instruments under a nitrogen flow atmosphere at 50 mL/min, with a heating rate of 10°C/min, over the range 40–220°C, used approximately 10 mg of the rPP slivers.

c) Hydrothermal Treatment and Characterization of Wood Waste

Wood waste was hydrothermally treated (WT) at 180°C in a Teflon-coated stainless steel autoclave (100 mL) using 3g of WW and 80 mL of deionized water. After the reaction time (30, 60, 120, 180, and 240 minutes) elapsed, the sample was dried at 100°C for 24h.

Wood waste was analyzed before and after hydrothermal treatment by Scanning Electron Microscopy (SEM) using an FEI QUANTA 250 with 5000x magnification. Attenuated Total Reflectance - Fourier Transform Infrared Spectroscopy (ATR-FTIR) was performed on a Bruker Vertex 70v FTIR spectrometer fitted with a Platinum ATR single reflection diamond ATR module. Spectra were collected from 4000 to 600 cm⁻¹, with 4 cm⁻¹ resolution and 32 scans. Free hydroxyl groups (Free-OH) and lignin degradation were determined for the peak ratios /3340/1029 and /1734/1029, respectively [23]. Water Sorption (WS) was performed at 75.3 \pm 0.2% relative humidity with sodium chloride, following the ASTM-E-104 standard, for 78h at 25°C. Tangential swelling (SW) measurements were obtained with an electronic digital caliper to within 300 mm; the tangential thickness of the WS test specimens was measured before and after treatment. The statistical influences of Free-OH, lignin degradation, WA, and TS in wood wastes before and after hydrothermal treatment were analyzed using one-way analysis of variance (ANOVA), followed by Tukey's test at a significance level 5%.

d) Composite RPP/WT Experiments

Treated wood waste was mixed with a recycled polypropylene matrix according to the compositions in 23 full-factorial experimental designs with a central point (Table 1); reaction time, granulometric size distribution, and WT percentage were varied. All tests were performed in duplicate, and the response variance of each assay was used to estimate the overall variance of

individual responses. The statistical data used in response to the mechanical properties were obtained

from IZOD traction and impact tests using the Design-Expert® software.

Table 1: Factors and levels used in 23 full-factorial experimental designs with a central point

Factor	Name	Units	Type	Level (-1)	Level (0)	Level (+1)
A	Reaction time	minutes	Numeric	30	105	180
B	Granulometric	µm	Numeric	425–610	710-850	850–1400
C	WT percentage	%	Numeric	10	15	20

e) Preparation of Composites

The samples were processed in a Thermo Scientific HAAKE MiniLab II Rheomex CTM 5 twin-screw extruder at 190°C at 60 rpm for 5 min with internal recirculation. Mechanical test specimens were processed in an injection machine (Thermo Scientific HAAKE MiniJet II) with a cannon temperature of 210°C, mold temperature of 40°C, injection pressure of 450 bar, injection time of 15s, repression pressure of 300 bar, repression time of 30s.

f) Characterization of Composites

Tensile strength tests were conducted following ASTM D638 using an EMIC DL10000 machine at 20mm/min crosshead speed, using specimens (dogbone shape) with dimensions 16.15 mm x 3.25 mm x 3.25 mm. Flexural strength was measured according to ASTM D790 - 00 on a universal Lloyd machine (Instruments LR 10K plus model) using a crosshead speed of 20 mm/min. Specimens were prepared with dimensions of 84.00 mm x 12.66 mm x 3.25 mm. Izod impact strength measurements were conducted according to ASTM D256 - 02A using CEAST equipment (Resil Impactor Junior) with rectangular samples measuring 84.00 mm x 12.65 mm x 3.25 mm that were fractured by a test pendulum with a load (impact action) of 2.75 J. Water absorption was measured following ASTM D570; before testing, the specimens (20.00 mm

diameter and 5.00 mm thickness) were dried at 60°C for 24h, then immersed in distilled water for three weeks at $23 \pm 2^\circ\text{C}$. All properties were tested using five samples for each group.

III. RESULTS AND DISCUSSION

a) Characterization of Recycled Polypropylene

The X-ray diffraction patterns obtained for RPP are shown in Figure 1a. Peaks observed at $2\theta = 14.1, 16.8, 18.5,$ and 21.4° correspond to the (110), (040), (130), and (131) planes of isotactic polypropylene, respectively ^[24–26]. A peak confirms the presence of syndiotactic polypropylene at 25.4° ^[24]. High-density polyethylene (HDPE) exhibits lower intensity peaks at $23.7, 29.4,$ and 36.1° , which correspond to the (200), (210), and (020) reflection planes in the typical orthorhombic unit cell structure ^[27,28]. Compatibility of the recycled polymers increases the amorphous halo, indicating good dispersion of the components throughout the amorphous phase and decreasing the crystallinity ^[25]. Polypropylene and high-density polyethylene are considered semi-crystalline thermoplastics, as they have crystalline (ordered) and amorphous (disordered) domains. The recycled polymers tend to become fragile materials, reducing their deformation at rupture and impact resistance ^[29].

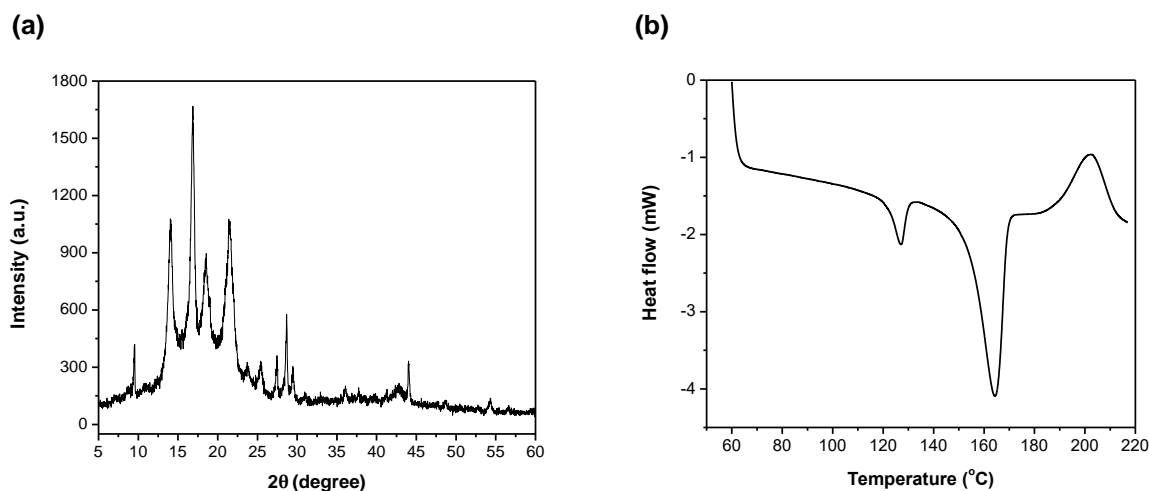


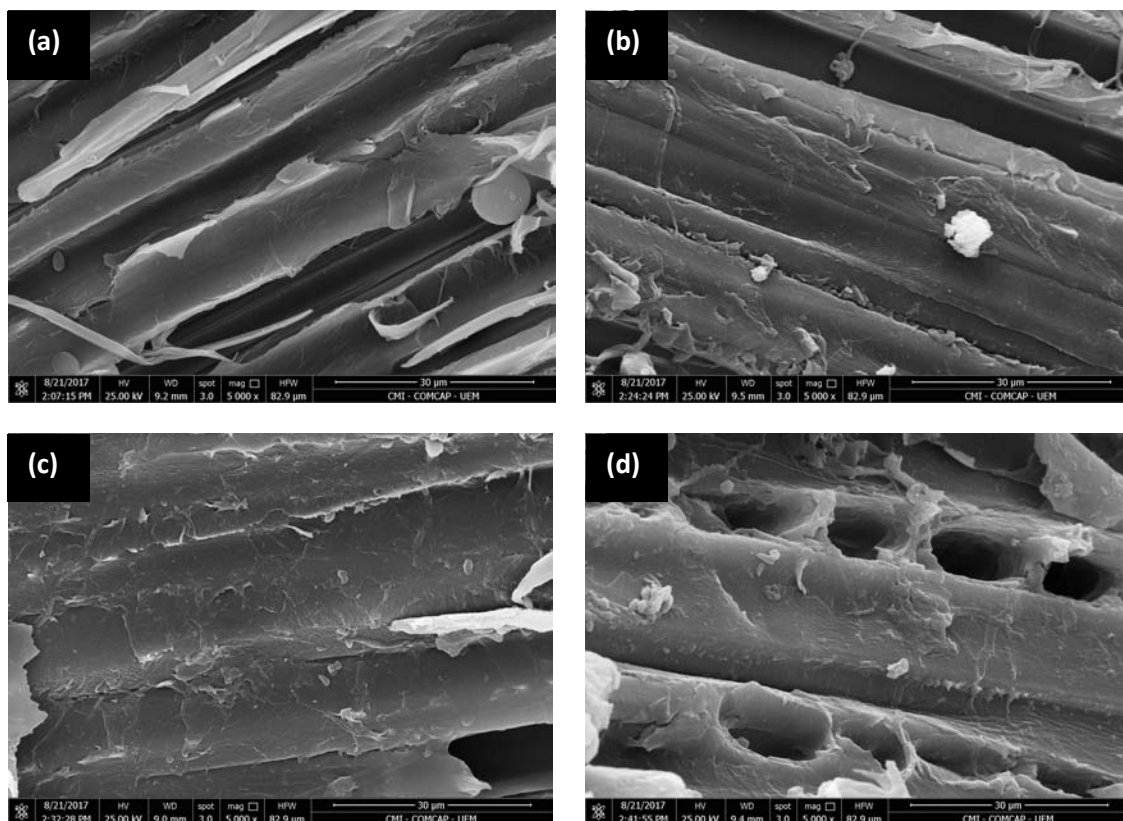
Figure 1: (a) X-ray diffractogram (b) DSC curve of the RPP

The DSC curve of RPP, shown in Figure 1b, exhibits two melting temperature (T_m) peaks at approximately 127°C and 164°C, corresponding to high-density polyethylene and polypropylene, respectively; and one exothermic transition at approximately 202°C, related to recycling additive compounds [24,28,30,31]. Generally, post-consumer polypropylene contains contaminants; in this case, the sample contains polyethylene [2]. Contamination is a byproduct of density-based separation used at recycling facilities since polypropylene and polyethylene have very similar densities [24]. The two endothermic transitions on the thermograph indicate that PP and HDPE co-exist and imply decreased compatibility between the polymers [31]. The exothermic event suggests the presence of a recycling additive, a compatibilizer agent, which serves as a polymer surfactant with low surface tension that promotes interfacial adhesion between different phases in a polyblend [32]. Polymers when melted, peak endothermic, in the manufacture of composites directly affect crystallinity, as a consequence of their mechanical, thermal and optical properties [33]. This first-order thermal event is necessary for the fusion and incorporation of reinforcement, such as fibers. The exothermic peak is the thermal event in which recrystallization or curing of the polymer occurs,

promoting the formation of a crystalline nucleus in the sample [33].

b) Characterization of Wood Waste

Scanning electron micrographs of the untreated and hydrothermally treated wood waste fibers are presented in Figure 2. The control fiber demonstrates a well-defined wood structure, with continuous fibers exhibiting a large, ordered surface along its length. During hydrothermal treatment, the surface area increases due to gaps formed in the lignocellulose structure on the surface of the fibers. This results in a vitreous and brittle appearance with depolymerization and solubilization from lignin and hemicellulose [31,32,34]. Throughout the reaction, sample deformation becomes visible with the formation of small holes in the surface [32]. The ray cells along the tangential direction of the wood retain their shape, but small cracks appear in the middle of the lamella [35]. Large cavities or cracks are observed along the fibers transverse section [34]. The formation of intercellular spaces can be attributed to the disruption of ray cells with thin cell walls by gases released as extractives degrade during the drying phase; this process is hampered by the lignins present in Eucalyptus wood [34].



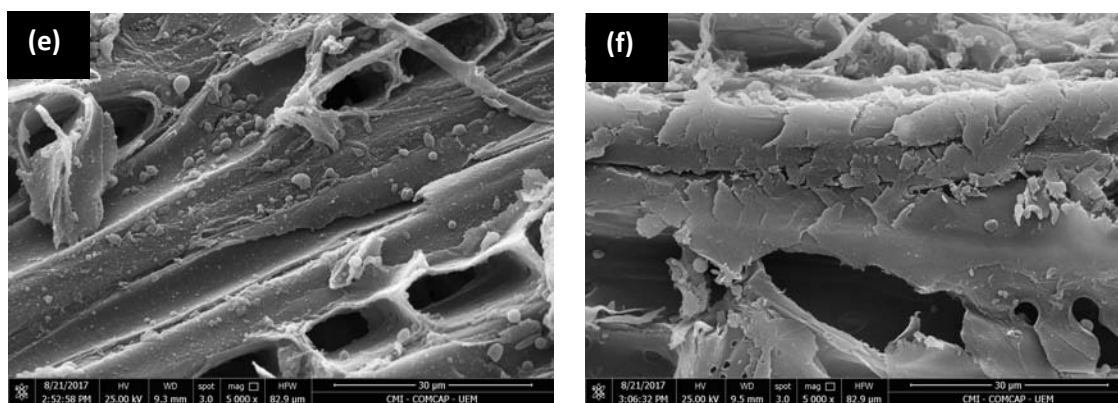


Figure 2: SEM images of wood waste non-treated, (a) control, and treated, (b) WT30, (c) WT60, (d) WT120, (e) WT180, and (f) WT240

Infrared spectra of the untreated and hydrothermally treated wood fibers display characteristic absorption signals for lignin and cellulose structures, as shown in Figure 3. The major absorption bands are associated with the vibrational stretch of the OH bond at 3336 cm^{-1} [34,35]. Lignin is confirmed by the absorption

signal at 1734 cm^{-1} , corresponding to C=O ketone vibrational stretches [35,36]. The reference absorption band at 1031 cm^{-1} corresponds to C-O cellulose binding, a highly stable structure to thermal and chemical treatments [23].

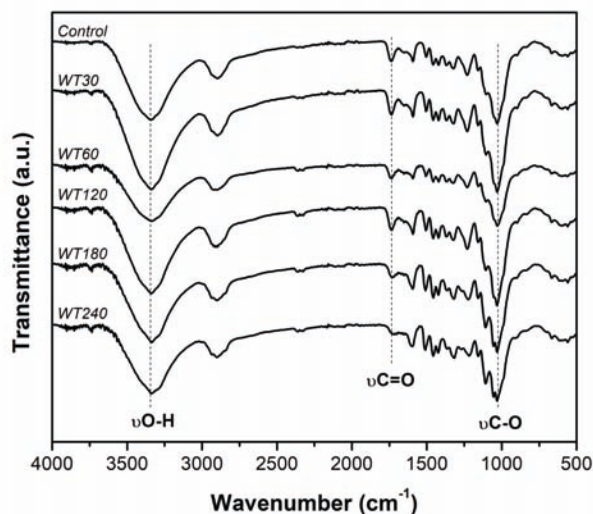


Figure 3: ATR-FTIR spectra of wood waste non-treated (Control) and treated (WT) at different times

ATR-FTIR analysis was used to study the relationship between stable absorption bands and bands modified by hydrothermal treatment. The assignments of the absorption bands observed in the infrared spectra of the main functional groups are presented in detail in Table 2. After hydrothermal treatment, the I3338/1031 ratio tends to decrease, confirming the loss of free hydroxyl groups (-OH free) on the WT surface. In general, wood fibers subjected to longer reaction times (over 120 min) exhibited a reduction in hydrophilic character. The decrease in I1734/1031 also corresponds to a decrease in the lignin ketone groups. This decrease may result from the decomposition of lignin at prolonged reaction times (over 180 min) since lignin is known to have more

excellent thermal stability for hydrothermal processes than cellulose [37]. The results demonstrate increased hydrophobic character and lignin degradation with increasing reaction time. These fiber modifications facilitate a more comprehensive study of the affinity between wood fibers and recycled polymer.

Table 2: Chemical and physical properties of wood waste

Sample	Peak (I _{3336/1021})	ratio Peak ratio (I _{1734/1031})	Water sorption* (%)	Tangential swelling* (%)
Control	0.956	0.242	10.617 ± 0.098 ^a	74.6 ± 4.6 ^a
WT30	0.976	0.240	10.188 ± 0.067 ^b	56.6 ± 2.5 ^b
WT60	0.925	0.225	9.685 ± 0.128 ^c	42.2 ± 3.8 ^c
WT120	0.877	0.236	8.993 ± 0.138 ^d	37.4 ± 1.8 ^c
WT180	0.891	0.158	7.407 ± 0.072 ^e	18.8 ± 3.9 ^d
WT240	0.906	0.121	6.700 ± 0.034 ^f	7.0 ± 2.2 ^e

*Similar letters on the column are not significantly different ($p > 0.05$)

Water absorption and tangential swelling of wood fibers (shown in Table 2) reached equilibrium after 22h. Decreased water absorption percentage was observed with increasing treatment time; this corroborates spectral data indicating a decrease in the number of free hydroxyl groups on the surface of the wood fibers. The same trend was also observed with tangential swelling: decreased water absorption resulted in reduced sample swelling. This results from increased hydrophobicity of the wood fibers due to partial degradation of the cellulose and lignin molecules after hydrothermal treatment ^[19].

c) Statistical Analysis of Composite RPP/WT

Responses for the strain (Y1), yield strength (Y2), Young's Modulus (Y3), impact strength (Y4), flexural strength (Y5), and water absorption (Y6) are shown in Table 3. A 2³ full-factorial design with four central points and one replicate, generating 20 total runs, was executed, and the factors for these responses were analyzed using statistical analysis Design Expert® software.

Table 3: Factors levels and responses for complete factorial design with four central points

Run	Factor			Response					
	A	B	C	Strain (MPa)	Yield Strength (Mpa)	Young's Modulus (Mpa)	Impact Strength (J/m)	Flexural Strength (MPa)	Water Absorption (%)
1	(+1)	(-1)	(+1)	24.315	23.625	936.154	45.736	38.216	0.920
2	(-1)	(-1)	(+1)	24.627	24.059	932.944	32.306	37.882	0.957
3	(-1)	(+1)	(-1)	22.727	21.275	652.237	35.746	33.072	0.465
4	(+1)	(-1)	(-1)	23.552	22.468	681.088	34.056	33.887	0.487
5	(-1)	(-1)	(-1)	24.339	22.830	711.690	46.101	35.680	0.475
6	(-1)	(+1)	(+1)	23.537	23.817	919.156	31.859	36.506	0.834
7	(0)	(0)	(0)	24.022	22.635	795.726	44.128	36.539	0.475
8	(+1)	(+1)	(+1)	22.344	22.111	727.958	30.028	34.596	0.758
9	(-1)	(-1)	(+1)	24.645	24.266	917.110	35.330	36.991	1.216
10	(-1)	(+1)	(+1)	23.645	23.004	922.100	30.404	36.796	0.731
11	(+1)	(+1)	(-1)	22.313	20.771	605.654	30.918	32.837	0.440
12	(0)	(0)	(0)	23.796	23.238	700.702	40.374	38.991	0.783
13	(+1)	(-1)	(-1)	23.641	22.559	690.824	34.861	33.567	0.451
14	(+1)	(-1)	(+1)	24.751	23.658	938.145	42.309	39.190	1.020
15	(0)	(0)	(0)	24.056	22.720	697.690	40.794	37.533	0.690
16	(-1)	(+1)	(-1)	22.740	21.294	652.076	33.120	35.649	0.389
17	(+1)	(+1)	(+1)	22.288	21.426	703.482	30.131	36.685	0.661
18	(0)	(0)	(0)	23.401	22.297	717.314	43.544	36.708	0.831
19	(-1)	(-1)	(-1)	23.530	22.816	717.884	45.998	36.228	0.554
20	(+1)	(+1)	(-1)	22.100	20.538	609.110	30.477	33.159	0.408

Table 4 shows the six estimated responses for the properties studied. The p -value of the model was below 0.05, indicating a probability of less than 5% of the null hypothesis. ANOVA verifies this low p -value for all responses, shown in Supplementary Materials

(Appendix A), for which the models obtained have a p -value less than 0.05. R-squared values close to 1.0 indicate that the model accurately predicts a fitting curve and adjusted R^2 .

Table 4: Estimated parameters

Factor	Y1	Y2	Y3	Y4	Y5	Y6
Intercept	23.44	22.53	769.85	35.59	35.68	0.67
A	-0.28 ^a	-0.39 ^a	-33.30 ^a	-0.77	-0.42	-0.030
B	-0.73 ^a	-0.75 ^a	-45.88 ^a	-4.00 ^a	-0.77 ^a	-0.087 ^a
C	0.33 ^a	0.71 ^a	104.78 ^a	-0.82	1.42 ^a	0.21 ^a
AB	-0.17 ^a	-0.18 ^a	-29.12 ^a	-0.42	-0.18	0.011
AC	-0.064	-0.15	-14.90 ^a	3.06 ^a	0.48	-0.018
BC	-0.084	0.097	-10.58	-0.16	-0.19	-0.054
ABC	-0.12	-0.100	-25.14 ^a	-2.39 ^a	-0.39	0.000125
R-Squared	0.9445	0.9510	0.9739	0.9521	0.8329	0.8763
Adj R-Squared	0.9091	0.9198	0.9573	0.9216	0.7265	0.7976
Pred R-Squared	0.8299	0.8508	0.9504	0.8825	0.5387	0.6631
p-Value	<0.0001	<0.0001	<0.0001	<0.0001	0.0015	0.0003

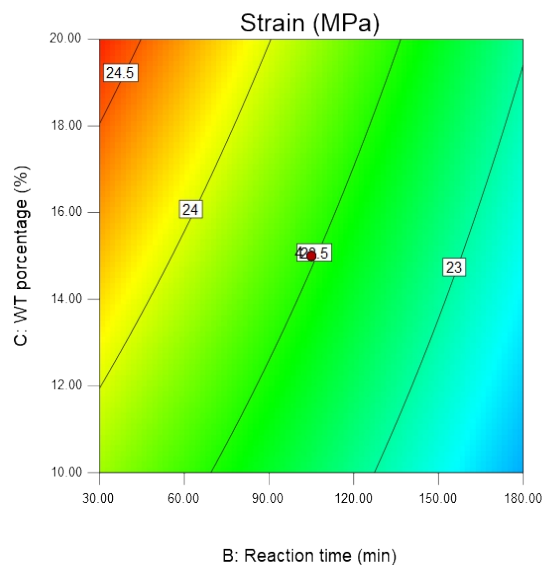
^astatistically significant value (model $p > 0.05$). Y1: Strain, Y2: Yield Strength, Y3: Young's Modulus, Y4: Impact Strength, Y5: Flexural Strength, Y6: Water Sorption.

i. Strain

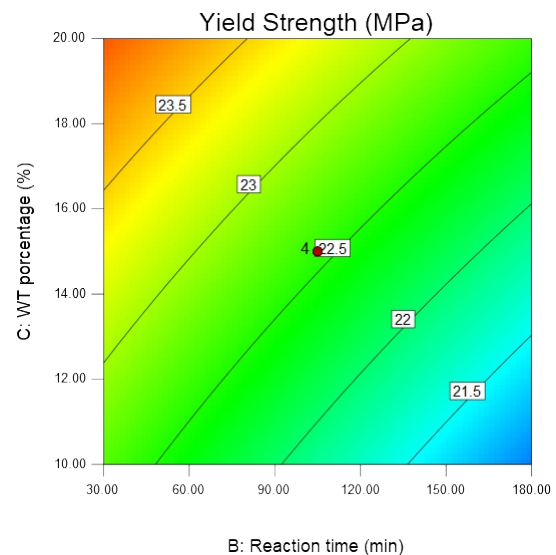
From the strain data (Y1 response), all main factors (A, B, C, and the AB interaction) were statistically significant, as predicted by the estimation. Figure 4a shows the contour plot for strain, where the optimal conditions for the composite concerning condition Y1 can be determined. The interaction between the amount

of wood fiber and reaction time (20% treated wood fiber for 30 min) produced the best results for strain since fibers treated for shorter times to preserve their structure, as seen in the SEM results. The fibers possess sufficient roughness for better interaction without gaps between the interface reinforcement and polymer [38,39].

(a)



(b)



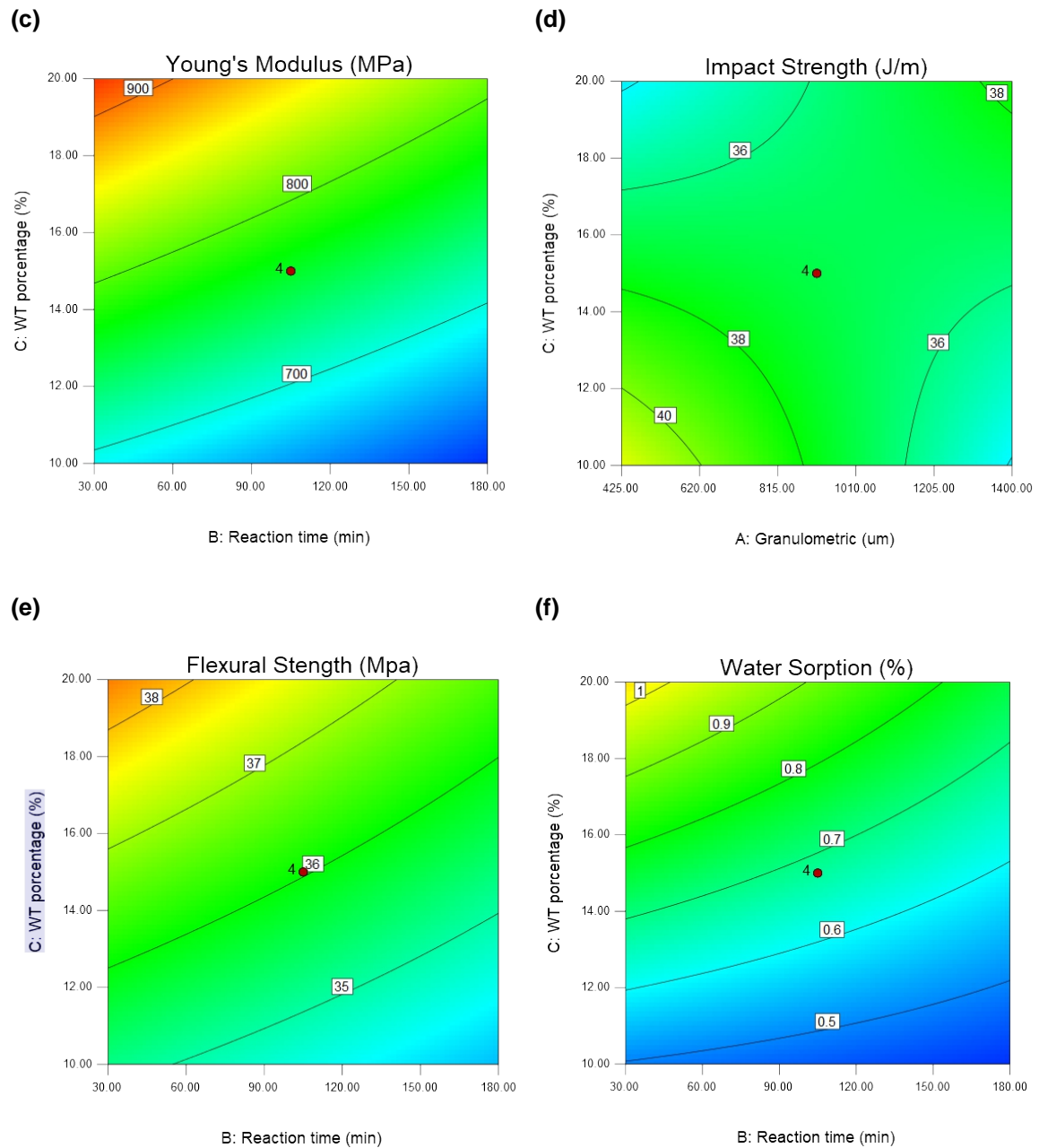


Figure 4: Surface contour response plot for (a) Strain, (b) Yield Strength, (c) Young's Modulus, (d) Impact Strength, (e) Flexural Strength and (f) Water Sorption

ii. Yield Strength

Considering the values obtained for the yield strength, the statistically significant interactions were the main factors A, B, C, and AB interaction, based on estimated data. The contour plot in Figure 4b demonstrates the best conditions for the composite for condition Y2. The interaction between wood fiber percentage and reaction time exhibited the best results, with 20% of wood waste treated for 30 min. Preserving the fiber lamellae in WT/30 produced good adhesion to RPP, increasing the yield strength of plastic deformation [39,40].

iii. Young's Modulus

The estimated parameters for Young's Modulus test indicate that all factors and interactions were statistically significant except for BC. Y3 values increase with WT percentage since adding fibers with greater stiffness causes increased stiffness in the composite, as shown in Figure 4c [41]. Young's Modulus values for the composite depend on the fibers' properties; a material's rigidity is determined when subjected to an external tensile stress [14,42]. Reaction time (B) also produced a significant effect, as low values of Y3 were observed for composites containing fibers treated for 180 min; that is, more significant elastic deformation occurred for these

fibers^[42]. Increased crystallinity is expected in fibers with higher reaction times; thus, hydrothermally treated fibers tend to form less amorphous structures, reducing the elastic characteristics of the material^[19]. Hydrothermal treatment can also cause the appearance of carbon microspheres incompatible with the polymer, thus reducing its Young's Modulus values^[43].

iv. Impact Strength

Impact strength measurements highlight that the main factor B and the AC and ABC interactions were statistically significant, according to estimated data. The surface contour plot of impact strength is shown in Figure 4d; this value can be used to determine the best conditions for the formulation of the composite. The interaction between WT percentage and granulometric size indicated that the best formulations for impact resistance include 10% treated wood fiber with a 425 μm particle size and 20% treated wood fiber with a particle size of 1400 μm . The first formulation is favorable because wood fibers with low granulometry behave as hardening centers, blocking the propagation of cracks under impact^[44]. In contrast, in the second formulation, the larger fibers bridge the cracks and increase crack propagation resistance by improving impact resistance^[42].

v. Flexural Strength

Estimated values from the flexural strength test indicate that only factors B and C were statistically significant. Figure 4e shows the surface contour plot, where the optimal conditions for the composite concerning Y5 can be observed. Interaction between wood fiber quantity and reaction time provided the best results for tensile strength, with 20% wood fiber treated for 30 min. In this work, we have shown that fibers treated for shorter times preserve their structures, as observed with SEM, presenting sufficient roughness for improved interaction without gaps between the interface reinforcement and polymer^[38,39].

vi. Water Sorption

Factors B and C were statistically significant from the water sorption estimated data. Figure 4f shows the surface contour plot where the optimal conditions for the composite to condition Y6 can be observed. The interaction between WT percentage and reaction time indicated higher water adsorption for the composite with 20% wood fiber treated for 30 min; composites with a more significant amount of fibers tend to absorb a greater amount of water^[14]. In addition, fibers treated for less time preserve their structure, as observed in SEM; thus, these fibers can absorb more water than those treated for longer times^[39].

IV. CONCLUSION

Hydrothermal treatment of wood waste fibers results in intercellular spaces and degradation along the

fibers that increase with reaction time to produce rough and highly porous WTs. Hydrophobicity of surface fibers and lignin degradation increase with reaction times exceeding 180 min. The mechanical properties of the RPP/WT composites indicate that reaction time (main factor B) was the most statistically significant. Fibers treated hydrothermally for 30 min obtained good adhesion with RPP since fiber lamellae were preserved and the polymer retained compatible porosity. WT percentage (main factor C) was statistically significant in all factors tested; the material's rigidity increased with the amount of fiber in the material.

One of the biggest advantages of the composites is the flexibility in preparation in relation to the composition rPP/WT, offering the possibility of producing materials with different final mechanical properties. The proposed recycling method has great chances of success, due to the low cost of the equipment used and the innovative and ecological factor, as well as the association of cooperatives to supply the recyclable material.

From these results, the polymer composite of recycled polypropylene and hydrothermally treated wood waste fibers possesses favorable mechanical properties, making it a promising construction material that benefits the economy and the environment. The composites can be used to produce a variety of materials, including domestic use such as buckets and bowls, as well as products for civil construction, for manufacturing pipes and floorings. So those materials to be developed, more research must be carried out to evaluate the degradation of materials over long periods of environmental exposure.

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Keywords: *friction coefficient, penstock, hydroelectric dam, theoretical approach, numerical approach.*

GJRE-A Classification: LCC: TC540-599



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Comparative Analysis of Friction Coefficient Evolution in Hydroelectric Dam Penstocks: Theoretical vs. Numerical Methods

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Djiako Thomas ^ω, Djeumako Bonaventure [¥], Tcheukam-Toko Denis [§] & Kenmeugne Bienvenu ^x

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Keywords: friction coefficient, penstock, hydroelectric dam, theoretical approach, numerical approach.

1. INTRODUCTION

Investing in a hydropower plant is very expensive, but it has an operating life of more than 30 years, with very low operating and maintenance costs compared to other power plants. In addition, in recent years, the experience gained in forecasting risks and production have facilitated the financing of hydropower projects, including private investments. The field of hydropower also boasts state-of-the-art technology. These days, turbines have attained efficiencies of over 97% and are extremely reliable. These data hence ensure that we have installations that function properly and also reduces the risk of downtime.

The function of penstocks is to transfer water from the reservoirs to the installations (turbines in a

hydroelectric plant) that convert hydraulic energy into electrical energy [1]. They are made up of sections with singularities where the local hydrodynamic pressures take on high values, as they follow the shape of the relief: slopes, obstacles, crossing of troughs, etc. The hydraulic plant therefore supports a pressure that is of the order of the head, but the effects of head loss reduce this value [1].

For a conduit flow, forces normal to walls are not involved; only tangential and therefore viscous forces contribute to the drag and power input [2]. The pressure drop thus depends on the type of flow, determined by the Reynolds number, and on the internal roughness of the pipe. It should be noted that the absolute roughness represents the average thickness of the surface roughness of the pipe material [3]. The gradient of the linear head loss, also called friction slope, depends on the friction coefficient, the flow volume and the geometrical characteristics of the structure [4], [5].

The friction coefficient is a function of the Reynolds number characterizing the flow, and the relative roughness of the pipe under consideration [6]. Since the roughness in a penstock is closely related to the coefficient of friction, an accurate estimate of roughness is crucial to plant performance.

According to ENEO [7], the Songloulou hydroelectric plant is the largest (384 MW) and the Lagdo plant one of the most recent (72 MW) of the Cameroonian plants. The first is the largest dam on the South Interconnected Network (RIS), and the second the only hydroelectric power station on the North Interconnected Network (RIN). The objective of this work is to approach the reality of the field with regard to the environment of these dams, based on their structures. We will therefore characterize the friction in the penstock of each. This study is one of the preliminary analyses of the flow in the penstock of a Cameroonian hydroelectric dam. No previous study on the friction parameters in the penstocks of the two dams studied is available.

Therefore, this study will examine the hydraulic characteristics influenced by the penstock walls on the flow within the penstock and at its outlet. The study focuses on the evolution of the friction coefficient in the

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whole structure, obtained analytically by the conventional method and by a graphical approach. Our analysis will enable us to identify the most interesting method for approximating the exact value of the coefficient of friction. It may also give rise to a new approach to monitoring these delicate structures.

II. MATERIALS

The materials on which our study is based are the Songloulou and Lagdo dams. Figures 1 and 2 below show the mesh of the structures in question (the Songloulou and the Lagdo dams study area respectively).

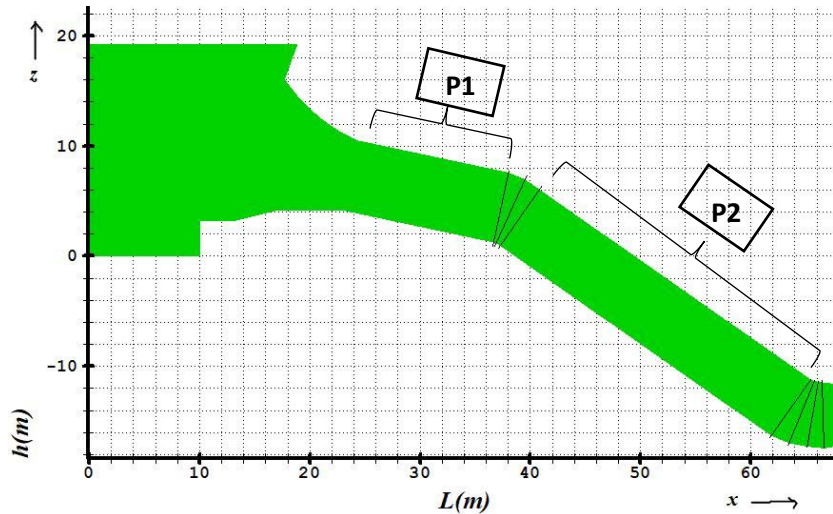


Figure 1: Songloulou Dam mesh

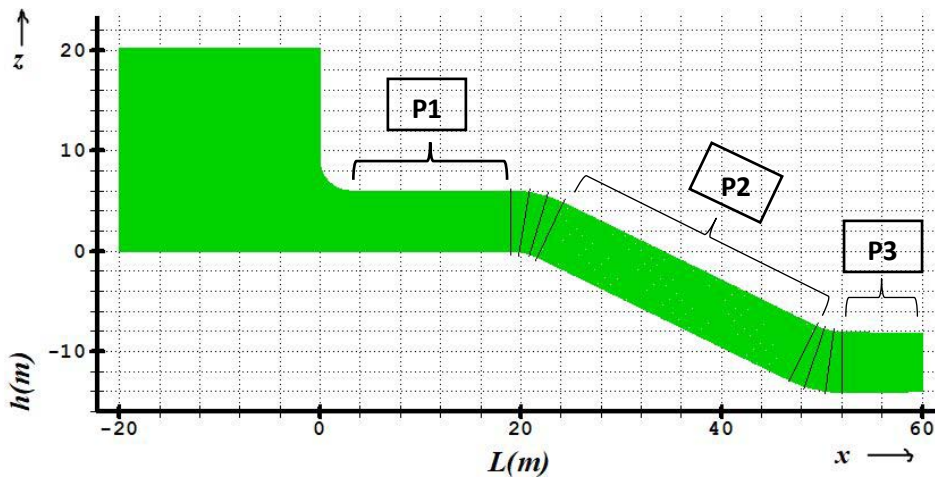


Figure 2: Lagdo Dam meshing

We modeled our power plants in Gambit 2.2. For Figure 1, the mesh type used is regular with 675845 quadrilateral meshes, 1108245 faces and 661186 nodes. For figure 2, the type of mesh used is regular with 197455 quadrilateral faces. On these figures, we have highlighted the portions in which we will present the results obtained (i.e. portions 1, 2 and 3 in the case of Lagdo). Thus for the case of Songloulou, we have P1 for the first portion of the pipe and P2 for the second portion. The same is true for the case of Lagdo where we have three portions, including P1, P2 and P3. Note that these portions represent the straight portions from which the results will be presented.

III. METHODS

a) Theoretical Approach

To determine the friction coefficient f of a turbulent flow, the empirical equation developed by Colebrook-White remains the reference equation. This equation is well known among hydraulic engineers, and continues to be the subject of research. The calculations and plots in this work are obtained from the following equation (equation 1) [8].

$$\frac{1}{\sqrt{f}} = -2 \log_{10} \left[\frac{\varepsilon/D}{3.7} + \frac{2.51}{Re \sqrt{f}} \right] \quad (1)$$

Where ε/D is the relative roughness (derived from the absolute roughness ε and the diameter D), Re

the Reynolds number and f the desired coefficient of friction. For the pressure drop, equation 2 below is used from the friction coefficient obtained previously [4], [5].

$$J = \frac{8f}{g\pi^2} \frac{Q^2}{D^5} \quad (2)$$

In the case of the conduite, another definition of the drag coefficient λ is used, called the friction coefficient [2]:

$$\lambda = \frac{d}{1/2\rho U_o^2} \frac{\Delta P}{l} \quad (3)$$

b) Numerical Approach

The Fluent software is used for this approach, as it has a large number of turbulence models to cope with many physical problems. The geometry of the structure as well as the type of boundary conditions of the physical domain of the parameters that characterize the fluid-structure interaction were modelled in Gambit version 2.2. Knowing that the fluid is incompressible, the motion is described using differential equations with derivatives in the following form [9]:

$$\frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} + \frac{\partial w}{\partial z} = 0 \quad (4)$$

For the u component parallel to the wall, the stationary Navier-Stokes equation simplifies considerably in the very near wall.

$$u \frac{\partial u}{\partial x} + v \frac{\partial u}{\partial y} = -\frac{1}{\rho} \frac{\partial p}{\partial x} + \nu \frac{\partial^2 u}{\partial x^2} + \nu \frac{\partial^2 u}{\partial y^2} \quad (5)$$

The convective terms tend to zero (adhesion condition) and the term $\frac{\partial^2 u}{\partial x^2}$ is negligible in front of the term $\frac{\partial^2 u}{\partial y^2}$ (weakly non-parallel flow condition). In short, all that remains is:

$$\frac{1}{\rho} \frac{\partial p}{\partial x} = \nu \frac{\partial^2 u}{\partial y^2} \quad (6)$$

It can be seen that the pressure gradient in the boundary layer imposes the curvature $\frac{\partial^2 u}{\partial y^2}$ of the velocity profile $u(y)$.

The energy equation is represented by the relation below [10]:

$$\frac{\partial T}{\partial t} + u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial y} + w \frac{\partial T}{\partial z} = a \left(\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} \right) \quad (7)$$

The decomposition of the viscous stress tensor is written:

$$\underline{\underline{\tau}} = \underline{\underline{\bar{\tau}}} + \underline{\underline{\tau'}} \quad (8)$$

with the viscous stress tensor given by:

$$\begin{cases} \bar{\tau}_{ij} = \mu \left(\frac{\partial \bar{u}_i}{\partial x_j} + \frac{\partial \bar{u}_j}{\partial x_i} \right) - \frac{2}{3} \mu \frac{\partial \bar{u}_k}{\partial x_k} \delta_{ij} \\ \tau'_{ij} = \mu \left(\frac{\partial u'_i}{\partial x_j} + \frac{\partial u'_j}{\partial x_i} \right) - \frac{2}{3} \mu \frac{\partial u'_k}{\partial x_k} \delta_{ij} \end{cases} \quad (9)$$

The equations of the turbulent kinetic energy and its dissipation rate give us:

$$\begin{cases} \rho \bar{u}_j \frac{\partial \bar{\epsilon}}{\partial x_j} = \frac{\mu_i}{\sigma_k} \frac{\partial^2 k}{\partial x_j^2} + \rho p_k - \rho \epsilon \\ \rho \bar{u}_j \frac{\partial \epsilon}{\partial x_j} = \frac{\mu_i}{\sigma_\epsilon} \frac{\partial^2 \epsilon}{\partial x_j^2} + \rho \frac{\epsilon}{k} (c_{\epsilon 1} p_k) - c_{\epsilon 2} \epsilon \end{cases} \quad (10)$$

The turbulence model is $k-\epsilon$ and the resolution method is RANS. The flow and site (location) parameters of each structure were considered in obtaining the different results. The Colebrook-White formula was integrated into the fluent solver via a calculation code.

IV. RESULTS AND DISCUSSION

a) Theoretical Approach

The theoretical approach is based on the Colebrook-white's formula [8] for a steel pipe in an evolving state, i.e. from new ($\epsilon = 0.03\text{mm}$) to worn ($\epsilon = 1\text{mm}$). We note that these values are those defined by hydraulic engineers and available in literature to characterize the evolution of roughness in steel pipes, as is the case in our dams.

i. Case of the Songloulou Dam

Starting from five velocities corresponding to the variation of velocity in the water intake for the periods of low water and flood, we calculated the friction coefficient f for four values of the absolute roughness ϵ . Table 1 below shows the results obtained.

Table 1: Evolution of the friction coefficient in our pipe as a function of ϵ

Velocity (V)	Ab. Roughness (ϵ)	Relative roughness (ϵ/D)	Friction Coef. (f)
4,045; 4,1.	0.05	0,00781	0,03495
	0.09	0,01406	0,04270
	0.3	0,04688	0,06946
	0.8	0,12500	0,11550
4,3; 4,5; 4,73.	0.05	0,00781	0,03495
	0.09	0,01406	0,04270
	0.3	0,04688	0,06946
	0.8	0,12500	0,11549

We find that the friction coefficient remains almost identical in the pipe with increasing Reynolds number for a given ϵ . Similarly, the only element influencing the friction coefficient f in this formula is ϵ . In the same logic, we have looked for the influence of these parameters on another structure, in order to compare the results obtained.

ii. Case of the Lagdo Dam

The approach is identical to that used for the Songloulou dam, with the speed ranges corresponding to the variation of speed in the water intake for the periods of low water and flood for this structure. Table 2 below presents the results obtained.

Table 2: Evolution of the friction coefficient in our pipe according to ϵ

Velocity (V)	Ab. Roughness (ϵ)	Relative roughness (ϵ/D)	Friction Coef. (f)
3,1; 3,3	0.05	0,00833	0,03571
	0.09	0,01500	0,04371
	0.3	0,05000	0,07156
	0.8	0,13333	0,12003
3,5; 3,7; 3,87	0.05	0,00833	0,03570
	0.09	0,01500	0,04371
	0.3	0,05000	0,07156
	0.8	0,13333	0,12003

The observation is the same as that made from Table 1. The friction coefficient remains practically the same in the pipe with the increase of the Reynolds number for a given ϵ . We can therefore conclude that for these results that whatever the other parameters in the Colebrook-White's formula, only the absolute roughness has an impact on the result obtained. However, in view of the two tables above, the coefficient of friction is also influenced by the flow rate (with higher values in the Lagdo dam penstock, which has a lower flow rate). We also note that the head loss corresponding to this approach according to formula 2 can only be static in a considered portion.

velocity of 4.045 m/s. Note that the same test was carried out on the upper wall.

b) Numerical Approach

This is the friction coefficient obtained graphically on each point constituting the portions of the penstock. The idea is to identify the evolution of the turbulent friction in each portion of the penstock according to the parameters of its environment. To do this, we seek by the same approach the maximum thickness of the asperities influencing the structure of the flow (what we will call real absolute roughness). Note that the fluid considered here is clear water with a density of 1000kg/m³.

i. Case of the Songloulou Dam

a. Research of the real value of ϵ

In order to find out the real thickness of the roughness influencing the friction coefficient in the pipe, we proceeded by the creation of graphic study zones. These study areas leave the walls towards the axis of the penstock at a small pitch (1mm to 5mm). Figure 3 below illustrates the evolution of turbulent friction in the penstock for three defined absolute roughness steps (starting $\epsilon=15\text{mm}$ towards the wall), with an initial

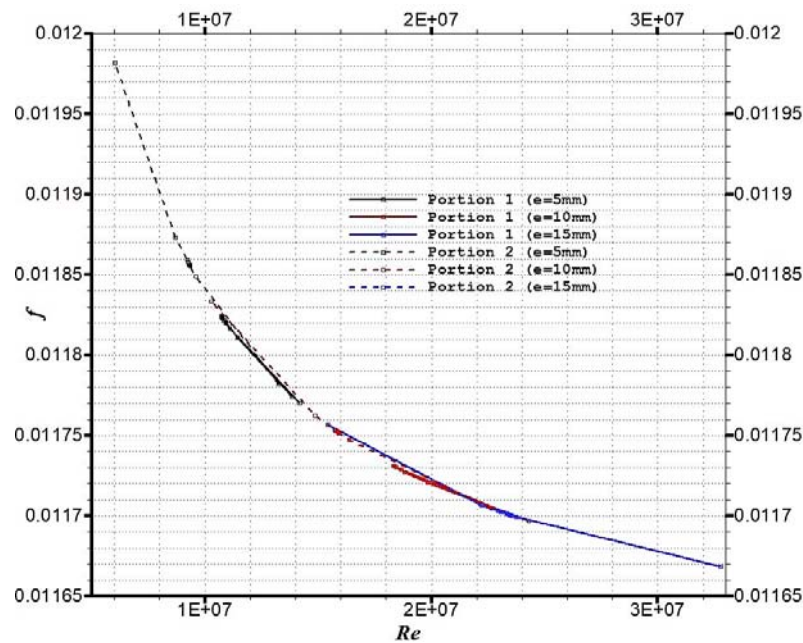


Figure 3: Evolution of turbulent friction in the pipe for $e = \varepsilon_{\max}$ different

The idea here is to see at what thickness e we have maximum friction. This thickness will then be considered as our maximum relative roughness (ε_{\max}). We can see from this figure that the maximum value of f is observed for $e=5\text{mm}$ ($\varepsilon_{\max}=5\text{mm}$). This is justified by the fact that as we approach the axis of our pipe ($e > 5\text{mm}$), turbulent friction begins to drop considerably. It is in this logic that in the continuation of our work on this dam, we will carry out the evaluation of turbulent friction

on a thickness of 5mm. It should also be noted that these plots are made only in the different portions.

b. *Determination of the evolution of the friction coefficient f in the penstock*

Figure 4 below shows the evolution of the friction coefficient f as a function of the actual parameters of the structure. Thus, for a good appreciation of the values, the result will be presented for each section separately.

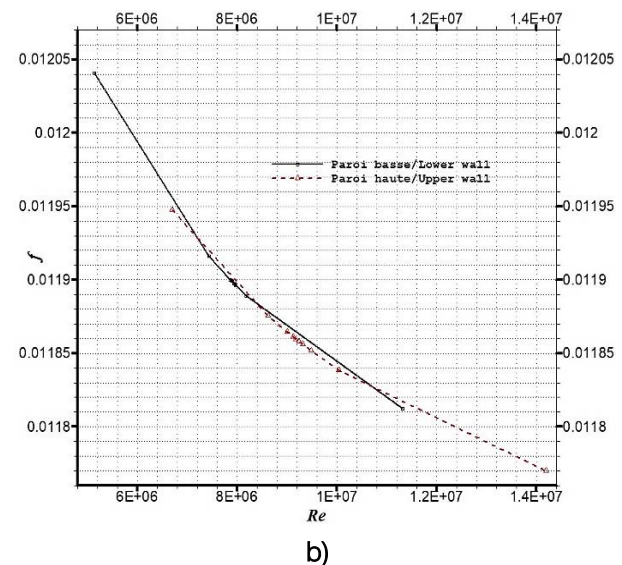
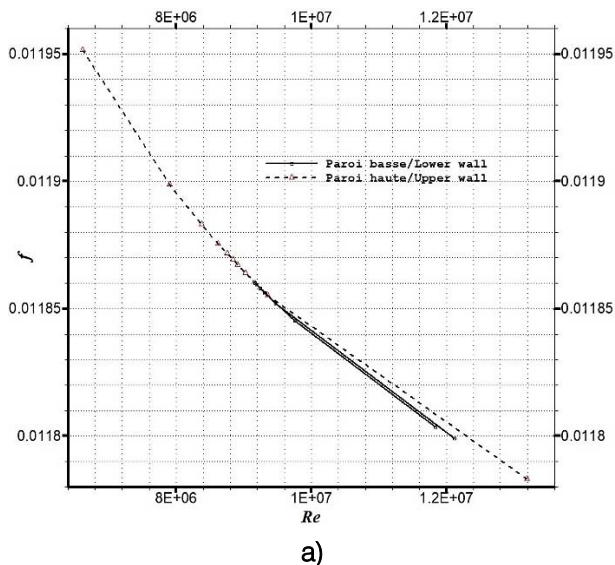


Figure 4: Evolution of the friction coefficient in section 1 (a) and section 2 (b) for $Re=26.9 \times 10^6$

We can see that the turbulent friction varies from one wall to the other. Also, it is not identical in the two portions. It is dominant on the upper wall in the first portion of the penstock of our dam (Figure 4.a). On the other hand, in the second portion of our penstock

(Figure 4.b), it is dominant on the lower wall with a higher value than the one observed in Figure 4.a. We will verify this hypothesis by applying this approach to another structure.

ii. Case of the Lagdo Dam

a. Finding the real value of ε

We proceed in the same way as in the case of the previous structure. Figure 5 below shows the turbulent friction in the pipe for $\varepsilon \leq 15\text{mm}$ on the low wall, with an initial speed of 3.1 m/s.

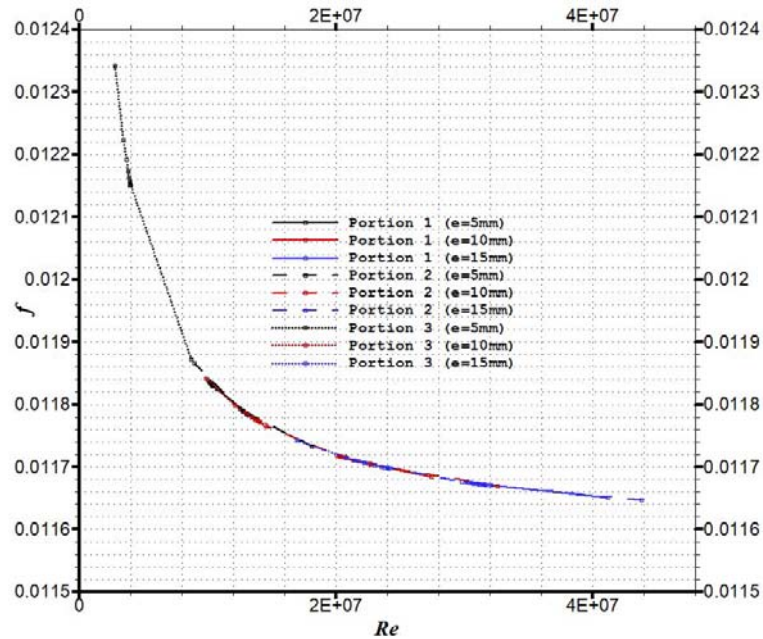
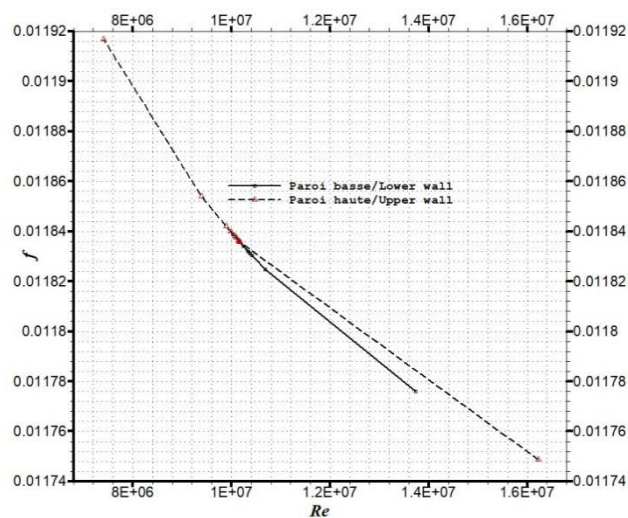


Figure 5: Evolution of the turbulent friction in the pipe for $\varepsilon = \varepsilon_{\max}$ different

As in the previous case, the maximum value of f is still found when $\varepsilon=5\text{mm}$ as shown in Figure 5 above. This is the case for the lower and upper walls.

b. Determination of the evolution of the friction coefficient f in the penstock

Figure 6 below shows the evolution of the friction coefficient f as a function of the actual parameters of the structure. The result is presented separately for each of the three sections for a good appreciation of the values.



a)

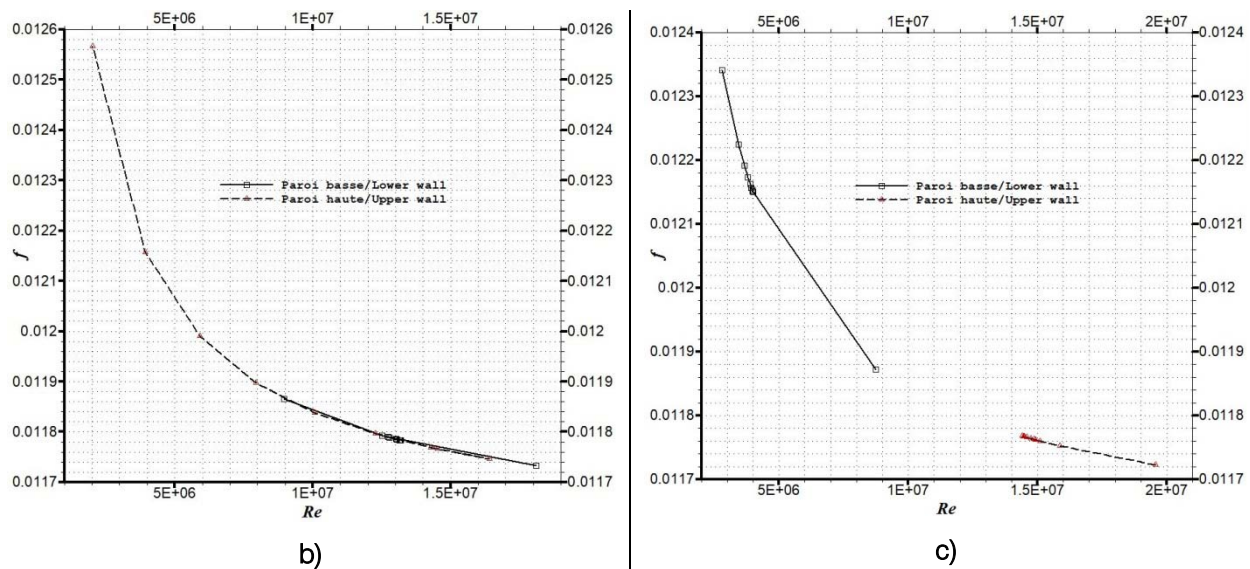


Figure 6: Evolution of turbulent friction in section 1 (a), section 2 (b) and section 3 (c) for different Re

We observe once again that the friction coefficient is not identical on the walls of the same section of pipe. It will be even less for each portion of our penstock. Apart from the third section of the pipe where turbulent friction is dominant on the lower wall (Figure 6.c), we rather have maximum turbulent friction on the upper wall for the other two sections (Figures 6.a and 6.b).

As a general remark, we observe that the friction coefficient is neither identical on the same wall nor identical in the same portion, as presented by the theoretical approach. More importantly, it is not identical on the same section of the pipe. As another remark, the impact of the Reynolds number is visible on the evolution of the friction and was not felt on the results of

the theoretical approach for our case. As a final remark, the head loss resulting from the numerical approach cannot be static in any portion of the pipe. This is not the case in the theoretical approach, given the shape of the curves obtained. Some researchers have already pointed out this shortcoming in the theoretical approach (for example Levin [11]).

c) Validation of the Numerical Model

To validate our model, the studies of Moss and Baker [12] and Abdalla et al. [13] were used. Figure 7 below shows the results obtained under similar study conditions. It shows the velocity profile over a reversal peak.

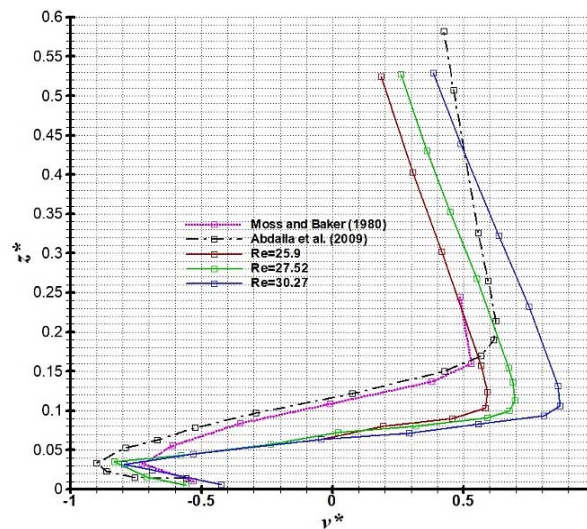


Figure 7: Flow structure on the sill for

In this figure, our study is presented by the Reynold's numbers Re . In addition to the determination on the reversal peak, we have applied it to our case

which is the Songloulou dam. The cause of the backdrop of our profiles is the transition from a free surface flow to a loaded flow, which doesn't exists in the

study of Abdalla et al [13]. In the same logic, Abdalla et al [13] in their study evaluate the impact of the threshold (peak) on the energy dissipation at this level. They observe the pointed shape which explains the recirculation phenomenon. This is also evident in our case as shown in Figure 8 below.

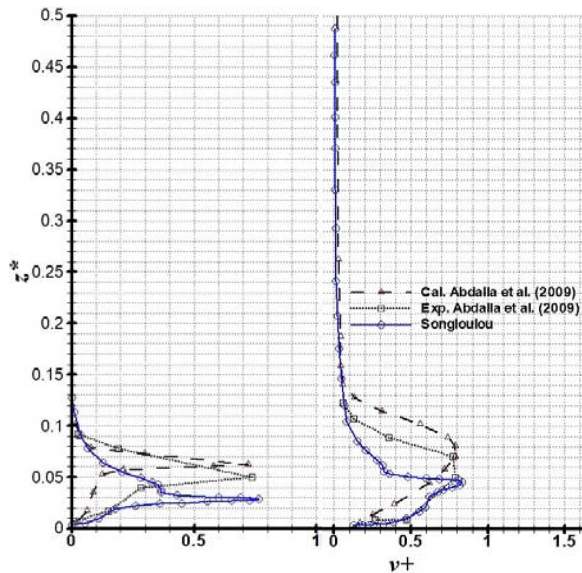


Figure 8: Turbulent intensity structure on the sill at $x=11m$ (left) and $x=13m$ (right)

V. CONCLUSION

The aim of this study was to investigate the impact of penstock walls on flow characteristics within the penstock and at its outlet. More precisely, it was to determine the evolution of the friction coefficient in the whole penstock. To carry out this work, two approaches were used. A theoretical approach, based on the direct determination by a commonly used formula (Colebrook-White's formula in this case), and another approach, the graphical, based on the numerical modelling of the structure under study.

In the theoretical approach, five velocities corresponding to the variation of velocity in the water intake for periods of low and high water were used. The coefficient of friction f was calculated for four values of absolute roughness ϵ . The results of this approach for our structures show that whatever the other parameters in the Colebrook-White formula (Reynolds number, diameter and absolute roughness), only absolute roughness has a visible impact on the result obtained. This can't be true, as clogging reduces the diameter and consequently modifies the flow rate.

In the numerical approach, we looked for the maximum thickness of the asperities influencing the structure of the flow; what we called real absolute roughness. We have thus observed that the maximum value of f is found for $e=5mm$ on the low and high walls. The results obtained from this approach show that the

coefficient of friction is neither identical on the same wall, nor identical in the same portion. More importantly, it is not identical on the same section of the conduit, contrary to the theoretical approach which is a fixed result in a portion of pipe. In the same way, the impact of the Reynolds number is visible on the evolution of the friction coefficient in the numerical approach. This is not felt in the results of the theoretical approach for our cases. These remarks demonstrate the limitations of the theoretical approach to evaluating and determining head loss in hydraulic structures. The numerical approach would appear to be more useful for planning the maintenance of such structures, and also for optimizing their performance.

We note, however, that apart from the geometric parameters of each structure studied, the usual data available in literature were used (absolute roughness, diameter and density of clear water). It is therefore recommended to take into account the characteristics of the sediments present in each site for the research to be upgraded.

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Thermal Vibration of Thick FGM Circular Cylindrical Shells by using Fully Homogeneous Equation and TSDT

By C. C. Hong

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Abstract- The third-order shear deformation theory (TSDT) effects on functionally graded material (FGM) thick circular cylindrical shells with entirely homogeneous equation under thermal vibration were investigated by using the generalized differential quadrature (GDQ) method. The nonlinear coefficient term of displacement field of TSDT was used in the equations of motion for thermal vibration of FGM thick circular cylindrical shells. Parametric effects of environment temperature and FGM power law index on the thermal stress and centre deflection of FGM thick circular cylindrical shells were investigated.

Keywords: shear deformation theory, sinusoidal, stress, displacement, environment temperature, shear correction coefficient.

GJRE-A Classification: LCC: TA654, TA660.C9



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1. INTRODUCTION

Some studies of shear deformation effects in functionally graded material (FGM) shells were presented. In 2018, Cong et al. [1] used Reddy's third-order shear deformation theory (TSDT) for the nonlinear displacements to study the time response of displacements of double curves shallow shells. The effecting numerical solutions for honeycomb materials in geometrical parameters, material properties and damping loads were presented. In 2017, Sobhaniaragh et al. [2] used the TSDT for the displacements to study the buckling loads of FGM carbon nano-tube (CNT)-reinforced shells in an environment (room temperature 300K) without thermal strains, parametric effects on material properties and critical buckling loads were presented by using the generalized differential quadrature (GDQ) method. In 2017, Dung and Vuong [3] used an analytical method with TSDT to study the buckling of FGM shells in elastic foundation under thermal environment and external pressure. In 2016, Dai et al. [4] presented a 2000-2015 review focused on coupled mechanics, e.g. thermo-mechanical responses with the first-order shear deformation theory (FSDT) models, HSDT models in widely used TSDT to study the bending, buckling, free and forced vibrations of FGM cylindrical shells by using various theoretical, analytical and numerical methods. In 2016, Fantuzzi et al. [5] used the numerical GDQ methods to study the free vibration

of FGM spherical and cylindrical shells, frequency solutions in FGM exponent number and thickness ratio were included. There were some numerical studies in the thick shells. In 2016, Kar and Panda [6] used the code of finite element method (FEM) and the TSDT displacements to get the numerical static bending results of deflections and stresses for the heated FGM spherical shells under thermal load and thermal environment. In 2015, Kurtaran [7] used the methods of GDQ and FSDT to get the numerical transient results of moderately thick laminated composite spherical and cylindrical shells. In 2012, Viola et al. [8] presented static analyses of FGM cylindrical shells under mechanical loading by using the GDQ method and a 2D unconstrained third-order shear deformation theory (UTSDT). The numerical solutions for stresses without thermal effect are obtained. In 2010, Sepiani et al. [9] used the FSDT formulation to get the numerical free vibration and buckling results for the FGM cylindrical shells without considering the thermal effect.

Many GDQ computational experiences in the composited FGM shells and plates were presented by including and considering the effects of thermal temperature of environment and heating loads. In 2017, Hong [10] presented the numerical thermal vibration results of FGM thick plates by considering the FSDT model and the varied shear correction factor effects. In 2017, Hong [11] presented the numerical thermal vibration and flutter results of a supersonic air flowed over FGM thick circular cylindrical shells by considering the FSDT model and the varied shear correction factor effects. In 2017, Hong [12] presented the numerical displacement and stresses results of FGM thin laminated magnetostrictive shells by considering with velocity feedback and suitable control gain values under thermal vibration. In 2016, Hong [13] presented the thermal vibration of Terfenol-D FGM circular cylindrical shells by considering the FSDT model and the constant modified shear correction factor effects. The computational GDQ solutions for the parametric effects of thickness of mounted Terfenol-D layer, control gain values, temperature of environment and power law index were investigated. In 2016, Hong [14] presented the transient response of Terfenol-D FGM circular cylindrical shells without considering the shear deformation effects. The computational GDQ solutions for the normal

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direction displacement and thermal stress were obtained. In 2016, Hong [15] presented the investigation of rapid heating-induced vibration of Terfenol-D FGM circular cylindrical shells. The computational GDQ solutions for both the amplitudes of displacement and stress are increasing with rapid heating value. In 2015, Hong [16] presented the rapid heating-induced vibration of Terfenol-D FGM cylindrical shells with FSDT transverse shear effects. The computational GDQ solutions for thermal stresses and centre deflections in the parametric effects of magnetostrictive layer thickness, control gain values, temperature of environment, power law index of FGM shells and applied heat flux were obtained. In 2014, Hong [17] presented the thermal vibration and transient response of Terfenol-D FGM plates by considering the FSDT model and the varied modified shear correction factor effects. The computational GDQ solutions for the effect of different mechanical boundary conditions were investigated. In 2014, Hong [18] presented the rapid heating-induced vibration of Terfenol-D FGM circular cylindrical shells without considering the effects of shear deformation. The computational GDQ solutions for the displacement of Terfenol-D FGM shells versus the thickness of Terfenol-D is stable for all power law index values. It was interesting to investigate the thermal stresses and centre displacement of GDQ computation in this nonlinear TSDT vibration approach and the varied effects of shear correction coefficient of FGM circular cylindrical shells with four edges in simply supported boundary conditions. Two parametric effects of environment temperature and FGM power law index on the thermal stress and centre displacement of FGM circular cylindrical shells including the effect of varied shear correction coefficient were also investigated.

II. FORMULATION PROCEDURE

For a two-material FGM circular cylindrical shell is shown in Fig. 1 with thickness h_1 of inner layer FGM material 1 and thickness h_2 of outer layer FGM material 2, L is the axial length of FGM shells, h^* is the total thickness of FGM shells. The material properties of power-law function of FGM shells are considered with Young's modulus E_{fgm} of FGM in standard variation form of power law index R_n . The others are assumed in the simple average form [19]. The properties of individual constituent material of FGMs are functions of environment temperature T . The time-dependent of nonlinear displacements u , v and w of thick FGM circular cylindrical shells are assumed in the nonlinear coefficient c_1 term of TSDT equations can be referred to the displacement equations [20], with the parameters u_0 and v_0 are tangential displacements in the in-surface coordinates x and θ axes direction, respectively, w is transverse displacement in the out of

surface coordinates z axis direction of the middle-plane of shells, ϕ_x and ϕ_θ are the shear rotations, R is the middle-surface radius of FGM shell, t is time.

Coefficient for $c_1 = \frac{4}{3(h^*)^2}$ is used in the TSDT expression. Thus $c_1 = 0$ is used and became the FSDT mode.

For the normal direction stresses (σ_x and σ_θ) and the shear stresses ($\sigma_{x\theta}$, $\sigma_{\theta z}$ and σ_{xz}) in the thick FGM circular cylindrical shells under temperature difference ΔT for the k th layer can be referred to the constitutive equations in terms of stresses, stiffness and strains [21-22], with the parameters α_x and α_θ are the coefficients of thermal expansion, $\alpha_{x\theta}$ is the coefficient of thermal shear, \bar{Q}_{ij} is the stiffness of FGM shells. ϵ_x , ϵ_θ and $\epsilon_{x\theta}$ are in-plane strains, not negligible $\epsilon_{\theta z}$ and ϵ_{xz} are shear strains.

A temperature difference ΔT between the FGM shell and curing area is given in functions of cylindrical coordinates x , θ and t . The heat conduction equation in simple form for the FGM shell in the cylindrical coordinates is used [13]. The dynamic equations of motion with TSDT for an FGM shell can be referred to the partial derivatives of external forces, thermal loads, mechanical loads and inertia terms [23], with the

parameters $I_i = \sum_{k=1}^{N^*} \int_k^{k+1} \rho^{(k)} z^i dz$, ($i = 0, 1, 2, \dots, 6$), in which N^* is total number of layers, $\rho^{(k)}$ is the density of k th ply. $J_i = I_i - c_1 I_{i+2}$, ($i = 1, 4$), $K_2 = I_2 - 2c_1 I_4 + c_1^2 I_6$.

The Von Karman type of strain-displacement relations with $\frac{\partial v_0}{\partial z} = \frac{-v_0}{R}$, $\frac{\partial u_0}{\partial z} = \frac{-u_0}{R}$ and $\frac{\partial w}{\partial z} = \frac{\partial \phi_x}{\partial z} = \frac{\partial \phi_\theta}{\partial z} = 0$ are used to simplify the strain equations. Thus the dynamic equilibrium differential equations in the cylindrical coordinates with TSDT of FGM shells in terms of partial derivatives of displacements and shear rotations subjected to partial derivatives of thermal loads, mechanical loads (p_1, p_2, q) and inertia terms can be derived in matrix forms. By assuming that mid-plane strain terms $\frac{1}{2}(\frac{\partial w}{\partial x})^2$, $\frac{\partial w}{\partial x} \frac{\partial w}{R \partial \theta}$ and $\frac{1}{2}(\frac{\partial w}{R \partial \theta})^2$ are in constant values, the relative parameters are listed as follows,

$$(A_{i^s j^s}, B_{i^s j^s}, D_{i^s j^s}, E_{i^s j^s}, F_{i^s j^s}, H_{i^s j^s}) = \int_{-\frac{h^*}{2}}^{\frac{h^*}{2}} \bar{Q}_{i^s j^s}(1, z, z^2, z^3, z^4, z^6) dz, \quad (i^s, j^s = 1, 2, 6),$$

$$(A_{i^* j^*}, B_{i^* j^*}, D_{i^* j^*}, E_{i^* j^*}, F_{i^* j^*}, H_{i^* j^*}) = \int_{-\frac{h^*}{2}}^{\frac{h^*}{2}} k_\alpha \bar{Q}_{i^* j^*}(1, z, z^2, z^3, z^4, z^5) dz, \quad (i^*, j^* = 4, 5), \quad (1)$$

in which p_1 and p_2 are external in-plane distributed forces in x and θ direction respectively. q is external pressure load, k_α is the shear correction coefficient, computed and varied values of k_α are usually functions of total thickness of shells, FGM power law index and environment temperature [24]. The $\bar{Q}_{i^s j^s}$ and $\bar{Q}_{i^* j^*}$ for FGM thick circular cylindrical shells with z/R terms cannot be neglected are used in the simple forms in 2010 by Sepiani et al. [9] [24].

The time sinusoidal displacements and shear rotations are varied with $\sin(\omega_{mn} t)$ can be referred [13]. The time sinusoidal temperature difference ΔT of thermal vibration is assumed in the following simple equation varied with $\sin(\gamma t)$,

$$\Delta T = \frac{z}{h} \bar{T}_1 \sin(\pi x/L) \sin(\pi \theta/R) \sin(\gamma t), \quad (2)$$

where ω_{mn} is the natural frequency in mode shape subscript numbers m and n of the shells, γ is the frequency of applied heat flux, \bar{T}_1 is the amplitude of applied temperature.

The GDQ numerical method is presented and referred [17][25-27]. The boundary conditions in dynamic GDQ discrete equations approach are to be considered for four sides simply supported, not symmetric, orthotropic of laminated FGM thick circular cylindrical shells and assumed that $A_{16} = A_{26} = 0$, $B_{16} = B_{26} = 0$, $D_{16} = D_{26} = 0$, $E_{16} = E_{26} = 0$, $F_{16} = F_{26} = 0$, $H_{16} = H_{26} = 0$, $A_{45} = D_{45} = F_{45} = 0$ under sinusoidal temperature loading. For a typical grid point (i, j) , the dynamic GDQ discrete equations can be rewritten into the matrix form as follows,

$$[A]\{W^*\} = \{B\}, \quad (3)$$

where $[A]$ is a dimension of N^{**} by N^{**} coefficient matrix ($N^{**} = (N-2) \times (M-2) \times 5$), $\{W^*\}$ is a N^{**} th-order unknown column vector and $\{B\}$ is a N^{**} th-order row external loads vector.

III. NUMERICAL RESULTS

The FGM material 1 is SUS304, the FGM material 2 is Si_3N_4 used for the numerical computations under four sides simply supported. The frequency γ of applied heat flux for the thermal loads is

given in the heat conduction equation can be reduced and simplified [13]. It is needed to get the calculation values of ω_{mn} under $p_1 = p_2 = 0$ and $q = 0$ for the free vibration. It is reasonable to assumed that u_0, v_0, w, ϕ_x and ϕ_θ are expressed in the referred time sinusoidal form of free vibration and expressed in the referred entirely homogeneous matrix equation [29] with varied parameters $m\pi/L$ and $n\pi/R$. The determinant of the coefficient matrix in the entirely homogeneous equation vanishes for obtaining non-trivial solution of amplitudes, thus the ω_{mn} and γ can be found.

The non-dimensional frequency $\Omega = (\omega_{11} L^2 / h^*) \sqrt{\rho_1 / E_1}$ and $f^* = 4\pi \omega_{11} R \sqrt{I_2 / A_{11}}$ for SUS304/ Si_3N_4 thick circular cylindrical shells with entirely homogeneous equation and TSDT under free vibration are compared with published literature as shown in Table 1, in which ω_{11} is the fundamental first natural frequency $m = n = 1$, ρ_1 is the density of FGM material 1. The present value of $\Omega = 1.906986$ on $c_1 = 0.925925/\text{mm}^2$, $L/h^* = 10$, $h^* = 1.2\text{mm}$, $T = 700\text{K}$, $R_n = 1$ is in close to the value of $\Omega = 1.65127$ with the material variation type A, three layers thickness ratio 1-8-1, the L directional radius of curvature is ∞ , $L/h^* = 10$, $R_n = 1$ for the FGM sandwich shell presented by Chen et al. [28]. The present value of $f^* = 5.041756$ at $L/h^* = 10$, $h^* = 2\text{mm}$, $T = 300\text{K}$, $R_n = 0.5$ is in close to the value of $f^* = 5.10$ on $n = 9$ with silicon nitride-nickel under classical shell theory (CST), no external pressure ($K_e = 0$) by Sepiani et al. [9].

The following coordinates x_i and θ_j for the grid points numbers N and M of FGM thick circular cylindrical shells are used to study the GDQ results,

$$x_i = 0.5[1 - \cos(\frac{i-1}{N-1}\pi)]L, \quad i = 1, 2, \dots, N,$$

$$\theta_j = 0.5[1 - \cos(\frac{j-1}{M-1}\pi)]R, \quad j = 1, 2, \dots, M. \quad (4)$$

The convergence of centre deflection $w(L/2, 2\pi/2)$ (mm) in the thermal vibration under $c_1 = 0.925925$ and $c_1 = 0$ for FGM thick circular cylindrical shells $L/h^* = 10$ with applied heat flux $\gamma = 0.2618004/\text{s}$ and $L/h^* = 5$ with $\gamma = 0.2618019/\text{s}$, respectively at $t = 6\text{s}$, $L/R = 1$, $h^* = 1.2\text{mm}$, $h_1 = h_2 = 0.6\text{mm}$, $T = 100\text{K}$,

$\bar{T}_1=100\text{K}$ are presented in Table 2. Considering the varied effects of k_α and ω_{11} for three values of R_n , in the nonlinear case of $c_1=0.925925/\text{mm}^2$: (a) for value of $R_n=0.5$, $k_\alpha=0.111874$ and $\omega_{11}=0.0001156/\text{s}$ are obtained. (b) for value of $R_n=1$, $k_\alpha=0.149001$ and $\omega_{11}=0.0001151/\text{s}$ are obtained. (c) for value of $R_n=2$, $k_\alpha=0.231364$ and $\omega_{11}=0.000114/\text{s}$ are obtained. In the linear case of $c_1=0/\text{mm}^2$: (d) for value of $R_n=0.5$, $k_\alpha=0.111874$ and $\omega_{11}=0.001000/\text{s}$ are obtained. (e) for value of $R_n=1$, $k_\alpha=0.149001$ and $\omega_{11}=0.001000/\text{s}$ are obtained. (f) for value of $R_n=2$, $k_\alpha=0.231364$ and $\omega_{11}=0.001000/\text{s}$ are obtained. The error accuracy is 0.000011 for the $w(L/2, 2\pi/2)$ in $R_n=0.5$ and $L/h^*=10$. The $N \times M=13 \times 13$ grid points can be used in the following GDQ computations of time responses for deflection and stress.

The $w(L/2, 2\pi/2)$ (mm) for the thermal vibration of FGM thick circular cylindrical shells are calculated with varied γ and ω_{11} . The γ values are decreasing from $\gamma=15.707960/\text{s}$ at $t=0.1\text{s}$ to $\gamma=0.523601/\text{s}$ at $t=3.0\text{s}$ used for $L/h^*=5$, from $\gamma=15.707963/\text{s}$ at $t=0.1\text{s}$ to $\gamma=0.523599/\text{s}$ at $t=3.0\text{s}$ used for $L/h^*=10$. Fig. 2 shows the response values of $w(L/2, 2\pi/2)$ (mm) versus time t under $c_1=0.925925/\text{mm}^2$ and $c_1=0$ for thick $L/h^*=5$ and 10, respectively, $L/R=1$, $R_n=1$, $k_\alpha=0.120708$, $T=600\text{K}$, $\bar{T}_1=100\text{K}$ during $t=0.1\text{s}$ - 3.0s . The maximum absolute value of $w(L/2, 2\pi/2)$ is 33.955818mm occurs at $t=0.1\text{s}$ for thick $L/h^*=5$ with $c_1=0.925925/\text{mm}^2$ and $\gamma=15.707963/\text{s}$. The maximum absolute value of $w(L/2, 2\pi/2)$ is 267.064789mm occurs at $t=0.1\text{s}$ for thick $L/h^*=10$ with $c_1=0/\text{mm}^2$ and $\gamma=15.707963/\text{s}$.

Usually the normal direction stress σ_x and shear stress $\sigma_{x\theta}$ are three-dimensional components and in functions of x , θ and z for the thermal vibration of nonlinear TSDT FGM circular cylindrical shells as shown in Fig. 3. Typically their values vary through the thickness direction of circular cylindrical shells. Fig. 3a shows the normal direction stress σ_x (GPa) versus z/h^* and Fig. 3b shows the shear stress $\sigma_{x\theta}$ (GPa) versus z/h^* on centre position $x=L/2$ and $\theta=2\pi/2$ of FGM circular cylindrical shells, respectively at $t=3.0\text{s}$ for thick $a/h^*=10$ with $c_1=0.925925/\text{mm}^2$ and $R_n=1$. The value 1.702E-03GPa of σ_x on $z=-0.5h^*$ is found in the greater value than the value 1.955E-04GPa of $\sigma_{x\theta}$ on $z=0.5h^*$, thus the σ_x can be treated as the dominated stress. Figs. 3c-3d show the time responses of the σ_x (GPa) on centre position of inner surface $z=-0.5h^*$ for $R_n=1$, thick $L/h^*=5$ and 10 with $c_1=0.925925/\text{mm}^2$, respectively. The maximum value of σ_x

is 2.005E-03GPa occurs at $t=0.1\text{s}$ in the periods $t=0.1\text{s}$ - 3s for thick $L/h^*=5$.

In Fig. 4 shows the response values of $w(L/2, 2\pi/2)$ (mm) versus T in 100K, 600K and 1000K with R_n values from 0.1 to 10 for thick $L/h^*=5$ and 10, respectively under $c_1=0.925925/\text{mm}^2$, $\bar{T}_1=100\text{K}$, γ and k_α at $t=0.1\text{s}$. Fig. 4a shows the curves of $w(L/2, 2\pi/2)$ vs. T and R_n for the $L/h^*=5$ case, the maximum absolute value of $w(L/2, 2\pi/2)$ is 49.057815mm occurs in $T=1000\text{K}$ for $R_n=2$. The $w(L/2, 2\pi/2)$ absolute values are all decreasing versus T from $T=600\text{K}$ to $T=1000\text{K}$, for $R_n=5$ only, it can withstand for higher $T=1000\text{K}$. Fig. 4b shows the curves of $w(L/2, 2\pi/2)$ vs. T and R_n for the $L/h^*=10$ case, they are almost located in the same curves for all value of R_n . The maximum value of $w(L/2, 2\pi/2)$ is 7.039238mm occurs in $T=1000\text{K}$ for all value of R_n . The $w(L/2, 2\pi/2)$ values are all increasing versus T for all value of R_n , the amplitude $w(L/2, 2\pi/2)$ of the $L/h^*=10$ cannot withstand for higher $T=1000\text{K}$.

In Fig. 5 shows the σ_x (GPa) on centre position of inner surface $z=-0.5h^*$ versus T and all different values R_n for the thermal vibration of thick $L/h^*=5$ and 10 cases. Fig. 5a shows the curves of σ_x vs. T and R_n for the $L/h^*=5$ case, the values of σ_x versus T are all increasing from $T=100\text{K}$ to $T=600\text{K}$ and then all decreasing from $T=600\text{K}$ to $T=1000\text{K}$ for all value of R_n . The maximum value of σ_x is 0.002018GPa occurs on $T=600\text{K}$ for $R_n=2$. The stress σ_x of the $L/h^*=5$ can withstand for higher $T=1000\text{K}$. Fig. 5b shows the curves of σ_x vs. T and R_n for the $L/h^*=10$ case, they are all located in the same curves for all value of R_n , the values of σ_x versus T are all increasing from $T=100\text{K}$ to $T=600\text{K}$ and then all decreasing from $T=600\text{K}$ to $T=1000\text{K}$ for all value of R_n . The maximum value of σ_x is 0.001745GPa occurs on $T=600\text{K}$. The stress σ_x of the $L/h^*=10$ can withstand for higher $T=1000\text{K}$.

IV. CONCLUSIONS

The GDQ solutions could be calculated and investigated for the deflections and stresses in the thermal vibration of FGM thick circular cylindrical shells by considering the varied effects of shear correction coefficient and nonlinear coefficient c_1 term of TSDT. The novel contribution of the GDQ stress and deflection solutions work is to investigate the effects of over estimation and under estimation of nonlinear coefficient term of TSDT in the thermal vibration of FGM circular shells. The natural frequency and parameter results of frequency including the entirely element effect in the homogeneous equation are also investigated and used to calculate the corresponding results in dynamic

convergence, vibration response of deflections and stresses.

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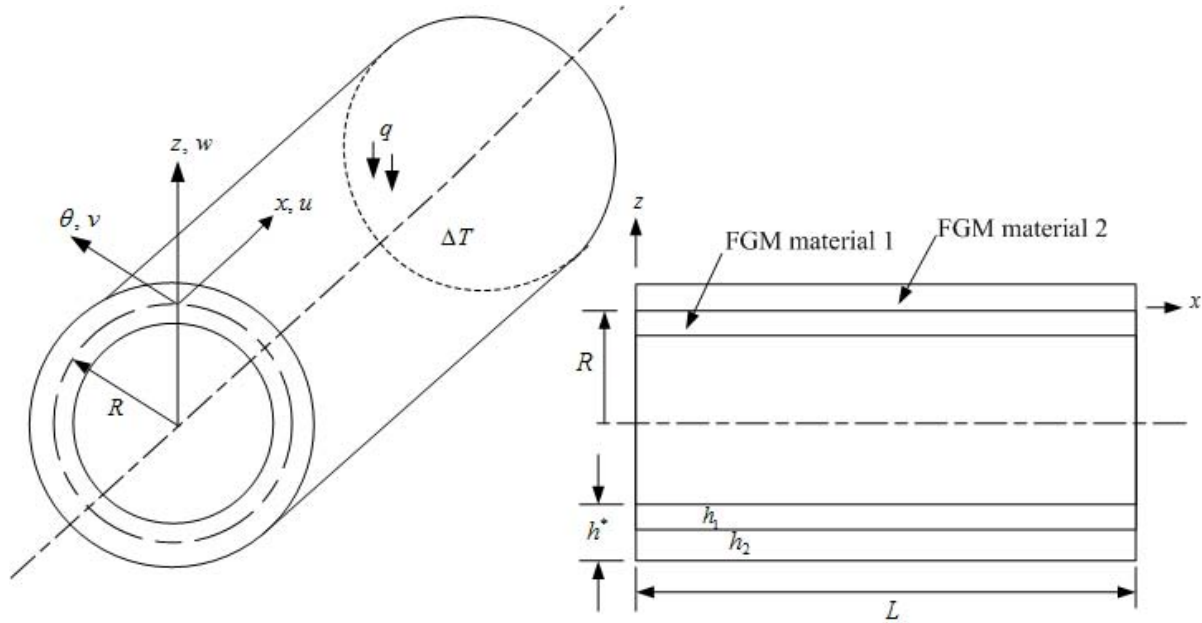


Fig. 1: Two-material FGM circular cylindrical shell

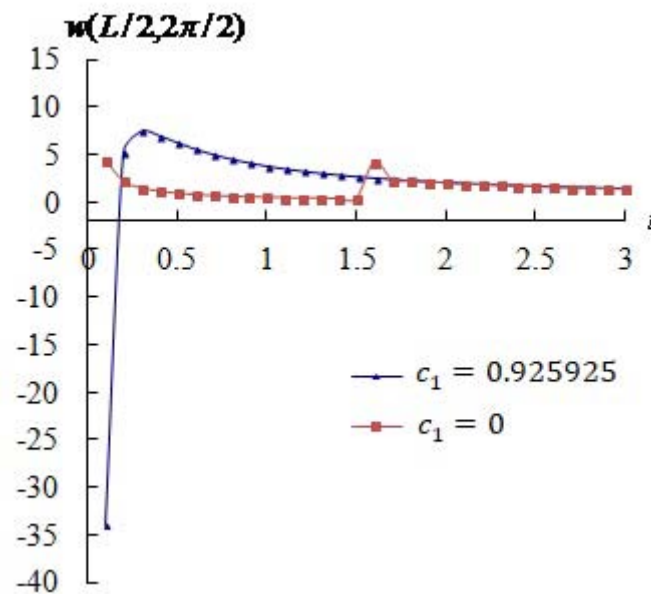


Fig. 2a: $w(L/2, 2\pi/2)$ versus t for $L/h^* = 5$

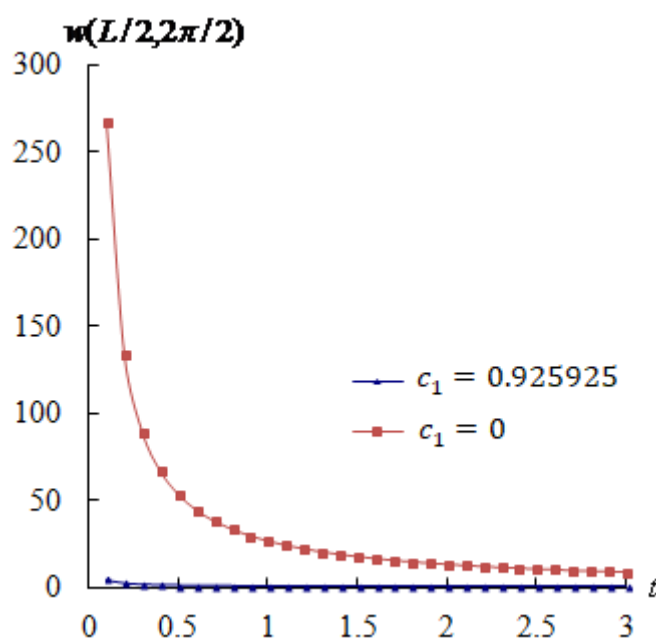


Fig. 2b: $w(L/2, 2\pi/2)$ versus t for $L/h^* = 10$

Fig. 2: $w(L/2, 2\pi/2)$ (mm) versus t (s) for $L/h^* = 5$ and 10

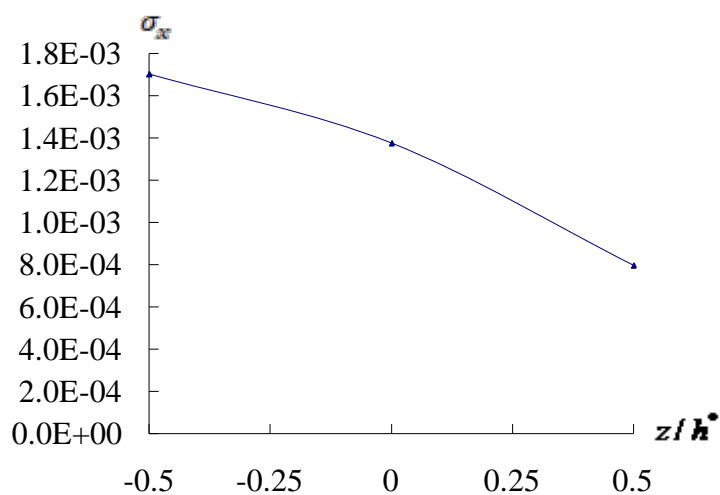


Fig. 3a: σ_x versus z/h^* for $L/h^* = 10$

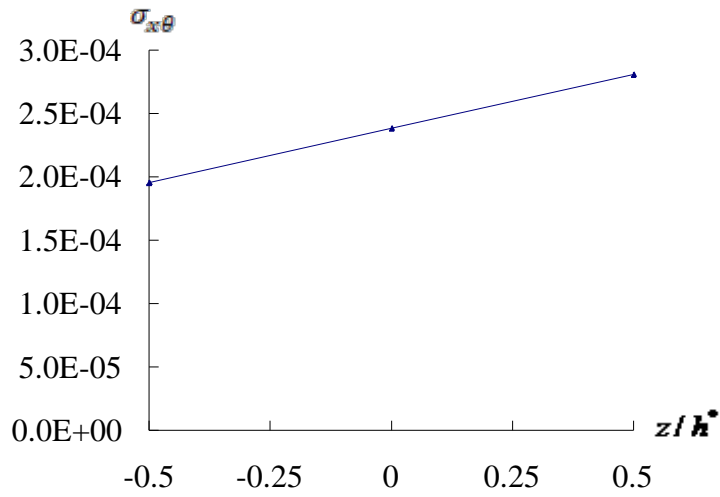


Fig. 3b: $\sigma_{x\theta}$ versus z/h^* for $L/h^* = 10$

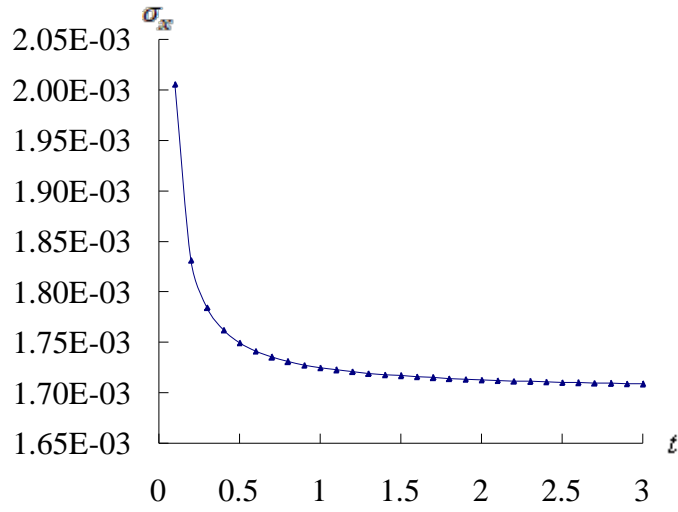


Fig. 3c: σ_x versus t for $L/h^* = 5$

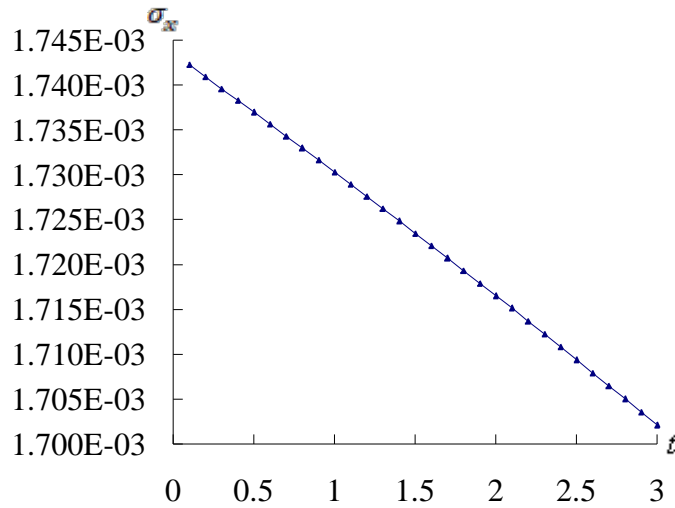


Fig. 3d: σ_x versus t for $L/h^* = 10$

Fig. 3: Stresses (GPa) versus z/h^* and t (s) for $L/h^* = 5$ and 10

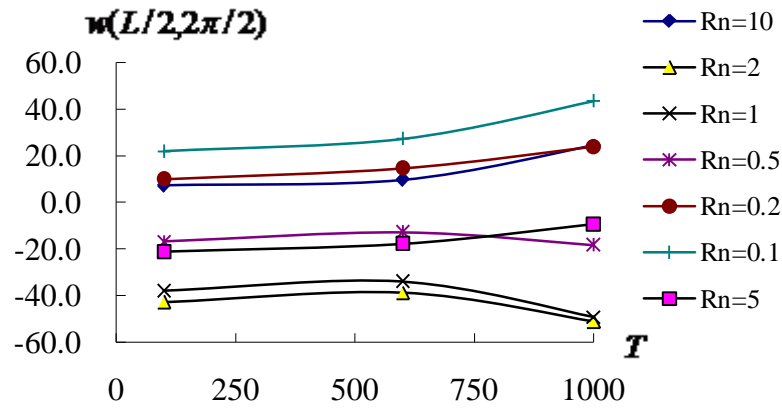


Fig. 4a: $w(L/2, 2\pi/2)$ versus T for $L/h^* = 5$ with R_n from 0.1 to 10

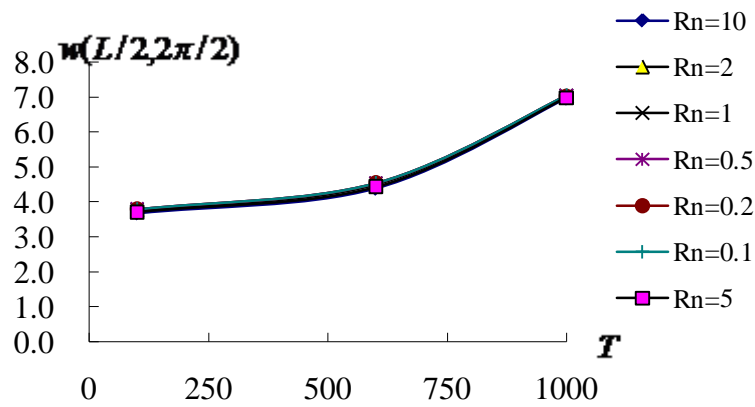


Fig. 4b: $w(L/2, 2\pi/2)$ versus T for $L/h^* = 10$ with R_n from 0.1 to 10

Fig. 4: $w(L/2, 2\pi/2)$ (mm) versus T (K) for $L/h^* = 5$ and 10

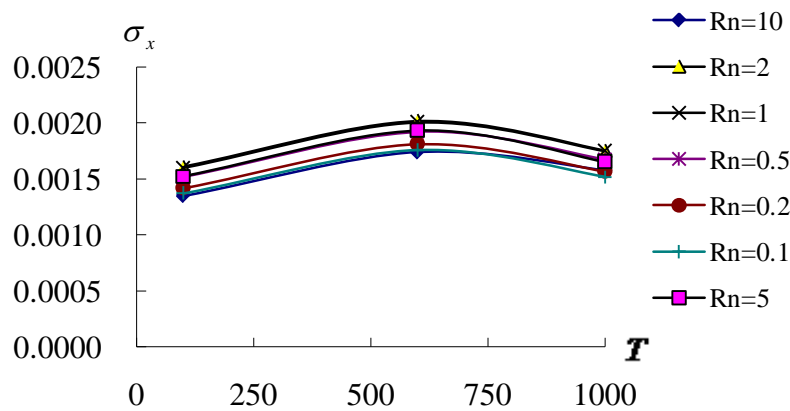


Fig. 5a: σ_x versus T for $L/h^* = 5$

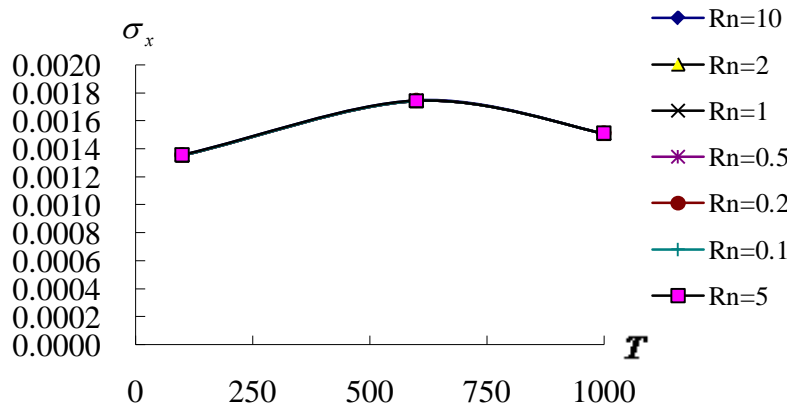


Fig. 5b: σ_x versus T for $L/h^* = 10$

Fig. 5: σ_x (GPa) versus T (K) for $L/h^* = 5$ and 10

Table 1: Compared values of Ω and f^* for SUS304/Si₃N₄

c_1 (1/mm ²)	h^* (mm)	Ω		f^*	
		Present method, $T=700K$ $R_n = 1$	Chen et al. 2017, type A, 1-8-1 $R_n = 1$ [28]	Present method, $T=300K$, $R_n = 0.5$	Sepiani et al. 2010, silicon nitride–nickel [9]
0.925925	1.2	1.906986	1.65127	0.517127	-
0.333333	2	11.14739	-	5.041756	5.10
0.000033	200	17704.31	-	800796.6	-
0.000014	300	3771.748	-	253980.0	-

Table 2: Dynamic convergence of the nonlinear TSDT FGM thick circular cylindrical shells

c_1 (1/mm ²)	L/h^*	GDQ grids			
		$N \times M$	$R_n = 0.5$	$R_n = 1$	$R_n = 2$
0.925925	10	7×7	5.093878	5.113268	5.162740
		9×9	5.086901	5.107157	5.157094
		11×11	5.086960	5.107170	5.157115
		13×13	5.086899	5.107180	5.157098
	5	7×7	0.591193	0.595686	0.615068
		9×9	0.589491	0.594395	0.614112
		11×11	0.589487	0.594404	0.614114
		13×13	0.589495	0.594400	0.614107
	0	7×7	23.807663	20.502832	21.591730
		9×9	3.210114	3.212350	3.193146
		11×11	3.216680	3.219491	3.210730
		13×13	3.213940	3.215927	3.197326
	5	7×7	0.524448	0.621863	0.832298
		9×9	0.505710	0.599383	0.798741
		11×11	0.505719	0.599400	0.798779
		13×13	0.505719	0.599401	0.798781



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The Main Methodological Postulate of Pyrometry and the Need for its Revision

By Alexandr Frunze

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Keywords: *pyrometry, methodological principles, radiation laws, spectral emissivity, temperature dependence of emissivity, reference means for spectral emissivity.*

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Abstract- The analysis of the methodological principles on which pyrometry is based is carried out. Special attention is paid to the main methodological postulate and its consequences. The roots of its formation are considered. It is shown that the large methodological errors characteristic of pyrometry are a direct consequence of the system of priorities arising from this postulate. A new basic methodological postulate of pyrometry is formulated, it is shown that the development of the ideas contained in it will reduce the number of methodological errors by an order of magnitude or more, and the necessary and sufficient conditions for this reduction are formulated.

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I. INTRODUCTION

The rapid development of microelectronics and microprocessor technology in the last quarter of the 20th century made it possible to bring instrument engineering to a qualitatively higher level. In many industries, the instrumental errors of measuring instruments have decreased to fractions of a percent. Pyrometers are no exception here.

But at the same time, as is known, any of the pyrometry methods has inherent methodical errors¹, the magnitude of which can reach 10 ... 15%, i.e. an order of magnitude or more exceeding the instrumental ones. There are still no ways to *guarantee* their reduction to the level of 1-2%. And the most significant thing is that over the past half century, it has not been possible to reduce the magnitude of these methodological errors in relation to any material whose non-contact temperature measurement may be in demand. And the reason for this, of course, is not at all due to dishonesty or low qualifications of researchers.

II. ON THE NEED TO ANALYZE THE METHODOLOGY OF PYROMETRY

The author of this work argues that the problems of pyrometry are methodological in nature. Their solution requires analysis and possible revision of the methodological principles of pyrometry. The work [1] is devoted to this analysis.

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For applied science, methodology is understood as a system (complex, interconnected set) of postulates and principles of research activity, which a scientist relies on in the course of obtaining and developing knowledge within a given specific scientific discipline or several scientific disciplines ([2]).

Obviously, there are methodological postulates, principles and approaches common to all branches of technical sciences (the obligation of mathematical calculations or modeling, the correspondence of calculated data to experimental data, etc.), and there are also particular, specific ones that apply only to individual industries or to one specific industry. It is quite obvious that it is not methodological principles and postulates common to all branches of technical sciences that slow down the development of pyrometry, because even in related fields (for example, in contact methods of temperature control) there are no problems of large methodical errors. Therefore, the source of irreducible methodical errors should be sought in specific methodological postulates and principles of pyrometry.

However, the methodological postulates and principles specific to pyrometry have not yet been clearly formulated. The reason is that most specialists in this field have not yet realized that the problems of incessant huge methodical errors are methodical in nature. The desire to formulate methodological principles and postulates that determine the course of development of a particular industry arises only after realizing the futility of trying to solve the problem within the framework of existing knowledge. Half a century of stumping in place on the issue of reducing methodical errors from a 10...20 percent level to units or fractions of a percent just suggests that there is a need to revise the methodological postulates and principles of pyrometry.

III. METHODOLOGICAL PRINCIPLES AND APPROACHES IN PYROMETRY

The principles and approaches most characteristic of modern pyrometry are listed below ([1]):

1. Regular resumption of attempts to find the temperature of the measured object only by its radiation, based only on Planck's or Wien's laws, without taking into account its radiative properties.

¹ The errors of the method

2. Consideration of the emissivity² as a minor, secondary, and even interfering factor, both in theoretical constructions and in the practical implementation of pyrometry methods.
3. The use of only those reference means during verification³ and calibration that perfectly implement the laws of radiation ("absolutely black bodies", BB).
4. The lack of reference means and measuring instruments of spectral emissivity.
5. The almost universal disregard of the dependence of the spectral emissivity characteristic of most objects on the temperature of the object and on the state of its surface (roughness, the presence of liquids, oil films, etc.).
6. Reduction (both in theoretical calculations and in practice) of the complex influence of emissivity to a one-dimensional effect described by a simple numerical coefficient, with complete disregard for the fact that emissivity is not a coefficient, but a function of at least two variables.
7. The use of various adjustment organs in almost all modern pyrometers, which make it possible to adjust the measurement results in any direction within a fairly wide range.
8. The lack of developed algorithms for determining the actual temperature of an object by its pseudo-temperature (brightness, partial radiation, radiation or spectral ratio), taking into account the temperature dependence of the emissivity characteristic of most objects.

As for paragraphs 3, 4 and 7, they are obvious. The statement of paragraph 2 also becomes obvious when analyzing almost all books published over the past 50 years, the authors of which try to cover pyrometry as a whole, rather than highlight certain selected issues. In these books, the laws of Planck, Wien, Stefan-Boltzmann, Rayleigh-Jeans, Kirchhoff, Lambert are usually described in detail first, and only after that the concept of emissivity is introduced, characterizing the difference between the radiation of real objects and the radiation of the BB.

According to claim 1, measurements in polarized light can be noted [3], the use of multiband spectral-ratio pyrometers with narrow spectral bands [4], the use of spectrometers [5], etc.

As a confirmation of what was said in paragraphs 5, 6 and 8, the following can be cited.

In the known relations present in almost all books on pyrometry, linking the actual temperature of an object T_d with its brightness or radiation temperature, the emissivity appears in the form of constants ε_λ , ε_s :

$$\frac{1}{T_d} = \frac{1}{T_b} + \frac{\lambda}{c_2} \ln \varepsilon_\lambda \quad (1)$$

where T_d is the actual temperature, K; T_b is the brightness temperature measured by the pyrometer, K; $c_2 = 1.4380 \cdot 10^{-2} \text{ m} \cdot \text{K}$; λ is the operating wavelength of the monochromatic brightness pyrometer, m; ε_λ is the radiation coefficient of the object at the wavelength λ .

$$T_d = T_r / \sqrt[4]{\varepsilon_s} \quad (2)$$

where T_d is the actual temperature, K; T_r is the radiation temperature measured by the pyrometer, K; ε_s is the integral radiation coefficient.

However, if we take into account that ε_λ and ε_s are not constants, but functions of wavelength λ and temperature T_d , and instead of *constants* substitute *functions* $\varepsilon(\lambda, T_d)$ in (1) and $\varepsilon_s(T_d)$ in (2), then simple calculation relations (1) and (2) turn into equations unsolvable in analytical form. There are no algorithms for solving these equations in general.

IV. THE MAIN METHODOLOGICAL POSTULATE OF PYROMETRY

The analysis of the above methodological principles and approaches characteristic of modern pyrometry allows us to identify something common to all of them without exception. This is the *implicitly postulated priority of radiation laws in this industry over all other laws and patterns* used to determine the temperature of heated bodies by their radiation. It is she who is today the main methodological postulate specific to pyrometry, which hinders its development.

This methodological postulate has a historical origin, since the laws of radiation were formulated back in the XIX century, and there is still no theory that would link the radiative characteristics of a substance with its physico-chemical constants, and at the same time would not diverge from experimental data *in the entire spectral range*.

Of particular importance is the fact that this priority is postulated implicitly, by stealth. None of the researchers claims that finding the *exact* temperature of an object by its radiation without knowing its radiative properties is possible. But in practice, all modern research in pyrometry is aimed precisely at finding the temperature of heated objects without knowing their radiative characteristics. After all, if this succeeds, it will be possible to forget about the dreary measurements of the emissivity, depending on both the state of the object's surface and its temperature. From this point of view, the game is worth the candle, since there are still no devices for measuring emissivity, and experimental installations that allow this to be done are large, expensive, low-mobility, measurements on them require

² Further, everywhere by the emissivity of an object we will understand its spectral $\varepsilon(\lambda)$ or temperature-spectral $\varepsilon(\lambda, T)$ emissivity

³ In Russia, verification is the confirmation by one of the accredited state metrological centers of the declared metrological characteristics of the device being verified

high qualifications and a lot of time. Therefore, the prospect of learning how to measure temperature by radiation without knowing the radiative properties looks very tempting.

The most likely solution to this problem seems to be using spectrometers, so today most research is conducted in this area [5]. However, the possibility of such a solution *for any predetermined material* is not yet obvious.

If we return to the pyrometers, then the following should be noted. The above-mentioned emissivity characterizes the difference between the radiation of a real object and the radiation of an BB. If the differences are small, then the measurement error with a pyrometer calibrated according to the BB is also small. But for many objects that have to be measured with pyrometers, the differences in their radiation spectrum are quite large from the spectrum of the radiation of the BB.

An BB-calibrated pyrometer, by definition, cannot correctly measure the temperature of an object that does not emit as an BB. The error that occurs during such measurements is the main methodical error, it is determined not by the quality of calibration, but by the problem of the measurement method (i.e., the need to measure an object that emits differently from the sample from which the pyrometer was calibrated). How can such an error be reduced or eliminated altogether? In today's practice, pyrometers are equipped with regulators, with the help of which a certain coefficient can be entered into them, usually taking a value from 0.1 to 0.99...1. This coefficient is usually called the "blackness coefficient", "degree of blackness" or "radiation coefficient". Using this coefficient, the operator can change the measurement result. It is assumed that he knows the "correct" value of this coefficient, and by setting it, he will correct the pyrometer readings and eliminate the mentioned methodical error.

To understand the negative consequences of this approach, you need to ask yourself – where do these coefficients come from? In the best case, measurements once made under these conditions are usually quite rough, with a small number of samples, without fully taking into account all factors affecting the result, without estimating the error. But more often – from literary sources compiled according to the same measurement results, performed by unknown people, unknown when, and with the same disadvantages.

The main thing here is that with this approach, *the correction value is not calculated, but determined experimentally by selecting the radiation coefficient for the value at which the pyrometer will show the correct result (or one that is considered correct for one reason or*

*another)*⁴. Let's add to this that manufacturers do not provide information about what the algorithm for correcting the measurement results of the radiation coefficient entered into this pyrometer is. The latter completely excludes the possibility of correctly accounting for the effect on the pyrometer measurement result of the difference between the radiation spectrum of the measuring object and the frequency response spectrum, the dependence of this difference on the temperature of the object itself, and on the spectral range of the pyrometer, and on the width of the range, and on the state of the surface of the object, and a number of other parameters. As a result, fitting to the expected result remains the only way to correct. In production practice, this leads to the fact that the technologist does not know which of the radiation coefficients to choose from the abundance available in various sources. As a result, the selection is made "by eye" so that the measurement result corresponds to the expected one. This is where users have measurement errors with pyrometers up to 10-20%.

That is, the user is trying to eliminate the methodical error, but the method used today to exclude it does *not guarantee* its reduction. With a successful combination of circumstances, it can decrease to the level of 1-2%, and if unsuccessful, it can remain at the same level of 10-20%. And at the same time, the instrumental errors of modern pyrometers often do not exceed 0.2... 0.5%. That is, the improvement of pyrometers in terms of further reducing the instrumental error at this stage is meaningless, because it does not lead to an increase in measurement accuracy. *Improving the accuracy of measurements in pyrometry has run into a barrier of methodical errors.* How to overcome it?

⁴ This is a problem that many still do not realize. Correction by experimentally selected coefficients causes very serious complaints from the point of view of metrology. This can be explained using such a simple example. Let's assume that we measure small voltage values in a printed circuit assembly with a DC microvoltmeter. As is known, when the copper probe of the device comes into contact with the Kovar pin of the chip, a fairly significant contact potential difference occurs, about 30 mV at room temperature. It is quite obvious that if, instead of subtracting this potential difference from the measurement result (adjusted, moreover, taking into account the temperature of the output of the chip), we smoothly "tweak" the gain of the microvoltmeter to the value that, according to someone once made estimates, gives the correct value of the measured value, then not only about the unity of measurements in radio engineering, but also their accuracy can be forgotten. *It is unacceptable to exclude errors by the method of "fitting an experimentally selected coefficient to the correct result", without relying on the measurement of the influencing quantity and on knowledge of its dependencies on certain environmental parameters.*

V. THE NEED TO ISOLATE THE PLANCK COMPONENT FROM THE TOTAL RADIATION OF THE OBJECT

To overcome this barrier, it is necessary to realize what happens when measuring when we neglect the influence of the radiative properties of the measured object. Let's turn to Fig. 1. Here are the Spectral Radiance (SR) of transformer steel when heated to 1127°C in a nitrogen-hydrogen atmosphere (1) and

high-alumina firebrick heated in air to the same temperature (2). The results are obtained based on the data given in [4]. Here, for comparison, the SR of the source of ideal Planck radiation – BB (3) at the same 1127°C is given. Obviously, the isolation of the Planck component⁵ 3 from dependencies 1 and 2 is an operation completely unobvious, none of the radiation temperature measuring devices is designed to solve this problem.

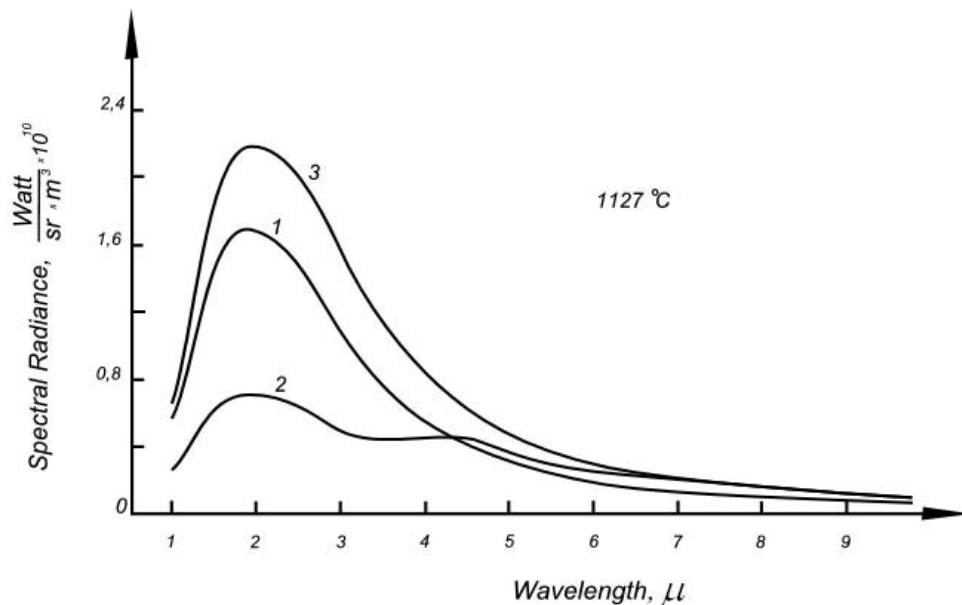


Figure 1: Spectral Radiance (SR) of transformer steel when heated to 1127°C in a nitrogen-hydrogen atmosphere (1), high-alumina firebrick heated in air to the same temperature (2) and BB at 1127°C (3)

The need to isolate the Planck component from the complete SR of an object arose at the dawn of the development of practical pyrometry. This turned out to be necessary because pyrometers are universally calibrated by BB, and after such calibration they can correctly measure the temperature of only those objects that emit as BB – "black" and "gray". When measuring other objects, it is that part of their radiation that distinguishes it from the radiation of the BB, and introduces an additional error, which we call methodical. Therefore, in order to exclude it, one way or another, its banking component must be isolated from the entire radiation of the object. Or somehow exclude the influence of non-Planck component.

VI. SPECTRAL EMISSIVITY AND ITS ROLE IN ELIMINATING METHODOLOGICAL ERRORS

The spectral emissivity can be defined as the result of the functional division of the SR of a real object into the SR of an BB (hereinafter, functional division is understood to be the division of the ordinate of the function-divisible by the ordinate of the divisor function

for the same abscissa for the set of all possible abscissae).

Figure 2 shows the dependences on the wavelength of the spectral emissivity of transformer steel (1) and high alumina firebrick (2), corresponding to a temperature of 1127°C [4]. As noted, we call these *functions* spectral emissivity in order to distinguish them from the *coefficients*⁶ introduced into energy pyrometers and still called by inertia by many users of pyrometers and authors of articles on pyrometry "emissivity". Let's add that the spectral emissivity is also a function of the temperature of the object.

⁵ Here and further, under the Planck component (Planck curve), we will understand the CR BB.

⁶ In this paper, the author calls this coefficient the "radiation coefficient", not emissivity.

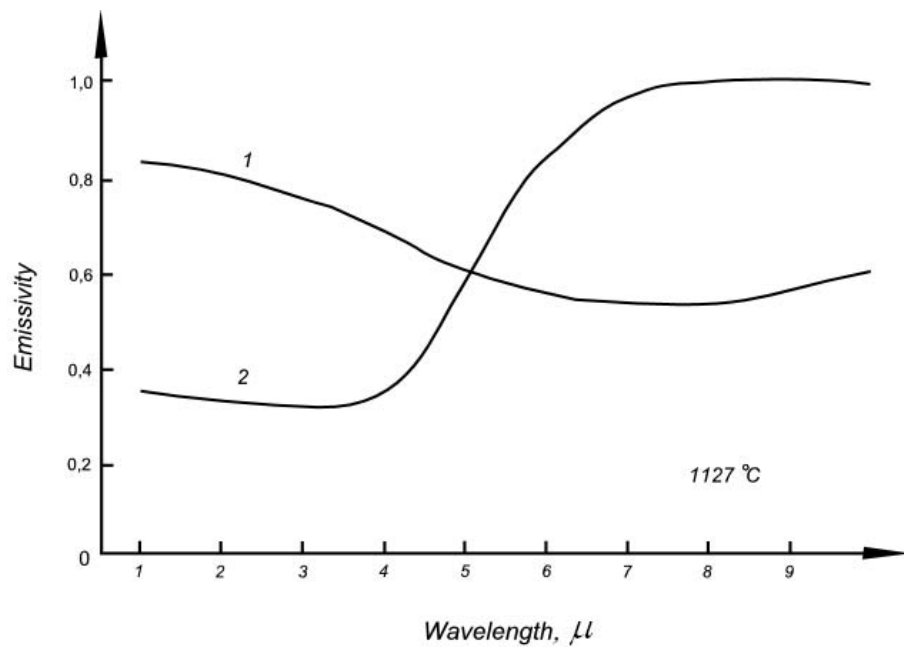


Figure 2: Spectral emissivity of transformer steel when heated to 1127°C in a nitrogen-hydrogen atmosphere (1), and high-alumina firebrick heated in air to the same temperature (2)

Thus, the spectral emissivity is a function that contains the difference between the SR of a real object and the SR of an BB having an equal temperature with the object. Its non-accounting (or incorrect accounting) does not make it possible to correctly convert the SR of the measured object into a Planck curve of equal temperature with the object, according to which this temperature can be measured without any systematic errors using an BB-calibrated pyrometer.

One of the ways of such a separation of the Planck component from the total radiation spectrum is the functional division of the SR of a real object into its real spectral emissivity. However, this is possible only if we have at our disposal almost complete spectral dependences of the SR and emissivity, i.e. lying in the wavelength range where the values of the Planck curves for these objects exceed 0.5-1% of their maxima. This is typical for spectral pyrometry, which is at the initial stage of its development. But pyrometers are not spectrometers, and such a functional division is impossible for them. Nevertheless, knowledge of the spectral emissivity is also necessary for classical energy pyrometers⁷ and pyrometers of spectral ratio (at least in the range of spectral sensitivity of these measuring instruments). However, it should be used in the allocation of the Planck component in a slightly different way. Here is one of the options for such a selection.

First, the full view of the object's SR is calculated for all measured temperatures (for example,

with a given step within the entire measurement range). To do this, for each of the temperatures, its spectral emissivity is functionally multiplied by its Planck function. Next, using the obtained SR, a set of pseudo-temperatures (brightness, or radiation, or spectral ratio - depending on which type of devices the correction is performed) is calculated. This calculation can be performed using calibration⁸ functions. Then a table is formed in which the actual temperature is assigned to each of the obtained pseudo-temperatures - the one whose Planck curve was used to calculate this pseudo-temperature. And at the last stage, the actual temperature of the measured object is determined based on the result of the pyrometer measurement using the above-mentioned recalculation table ([6]). When using the real spectral emissivity in this algorithm, methodical errors are excluded, and the error is determined only by the instrumental errors of the pyrometer used.

In fact, in this case, we modify the scale determining function (inverse of the calibration function) of the pyrometer so that it takes into account the difference between the SR of the measured object and the SR of the BB. Which is essentially equivalent to separating the Planck component from the total flux of its radiation.

Once again, I would like to draw readers attention to an important statement – if we want to measure temperature with a pyrometer without

⁷ Energy pyrometers are understood to be all pyrometers having only one radiation receiver, which determine the temperature by the magnitude of the signal from the receiver, i.e. by the magnitude of the energy flow that came to it

⁸ The calibration function is, in this case, the dependence of the voltage at the output of the receiver signal amplifier (for an energy pyrometer) or the spectral ratio (for a pyrometer of the spectral ratio) on the temperature of the BB

methodical errors, we must somehow isolate the aforementioned Planck component from all the radiation that came to the pyrometer. Its measurement with an BB-calibrated pyrometer will give the desired result. Or, one way or another, exclude the influence of a non-Planck component, which will lead to the same measurement result.

And how do we allocate the Planck component today? Few people have thought about this – the extraction operation, which is the result of a complex mathematical calculation, has been replaced by correction using coefficients⁹ introduced into pyrometers, most often determined experimentally. The task of isolating the Planck component in today's pyrometry has been simplified to the limit – the radiation flux that came to the energy pyrometer is actually simply divided by the radiation coefficient introduced into it. A complex functional transformation is ultimately reduced to division by a constant taken from tables, which very often have a very distant relation to the measured object. Hence the methodical errors, which are absolutely independent of the pyrometer's own instrumental error.

Naturally, classical pyrometers are not spectrometers, they do not measure the SR with some kind of normalizing coefficient, but its integral value in the form of a signal at the output of their radiation receiver. Therefore, the functional division mentioned above is not a task for them. If we are talking about energy pyrometers, then correction by the radiation coefficient is all they can do. But then the procedure for isolating the Planck component should somehow "migrate" at least to the calculation of the radiation coefficient. Such a calculation of the radiation coefficient, taking into account the spectral emissivity $\varepsilon(\lambda, T)$, is described below (see (4)). However, it has certain limitations, which will be described later. Therefore, the task of comprehensively eliminating the influence of temperature-spectral emissivity on the pyrometer measurement result is still relevant. Taking into account all the above, it will be formulated and specified in the last subsection of this article.

It is safe to say that devices for remote temperature measurement will continue to be calibrated according to BB in the future. Consequently, the task of isolating the Planck component from the full SR, according to which these devices will measure temperature, will also remain. And its solution is

impossible without knowledge of the spectral emissivity. And the more precisely it is determined (as well as the more accurately the calibration of the measuring instrument used is carried out), the more accurately the temperature of the measured object will be determined. In a different way, using some averaged coefficients introduced into pyrometers, it will not be possible to get rid of methodical errors.

VII. ABOUT THE NEW BASIC METHODOLOGICAL POSTULATE OF PYROMETRY

All of this means the need to rethink the basic methodological postulate mentioned above, which is specific to pyrometry, which consists in the fact that to determine the temperature of an object by its radiation, it is enough to know and use only the laws of radiation, ignoring accumulated or still missing knowledge about the radiative properties of specific objects. It should be replaced by a postulate proclaiming that *the exclusion of methodical errors in non-contact measurement of its temperature is impossible without knowledge of the real (not generalized or averaged!) the spectral emissivity of a particular measured object, and its correct accounting*. Attempts to deceive nature and continue to ignore the need to accumulate knowledge about the radiative properties of objects will leave unchanged methodical errors that have hindered the development of pyrometry for more than half a century.

However, this is not all. Since until now there has not been a theory that adequately connects the spectral emissivity with the physico-chemical constants of the object's material, it will be necessary to obtain the necessary information about the spectral emissivity experimentally. At the same time, it should be noted that at the moment there are no specialized measuring instruments for spectral emissivity on the market. Nevertheless, in the works of the author ([7, 8] the technical possibility of creating such measuring instruments is demonstrated, two such devices are described, one of which is protected by a patent of the Russian Federation.

The author also argues the need to have a verification scheme for such devices, as well as a currently missing standard of emissivity, which will stand at the top of this verification scheme ([9, 10]).

VIII. ABOUT WHAT ELSE IS NEEDED

However, knowledge of the spectral emissivity is only a necessary condition for reducing or completely eliminating large methodical errors inherent in pyrometry methods. It is not sufficient for this reduction, since algorithms and methods for minimizing/eliminating these errors with full consideration of temperature-spectral emissivity are either insufficiently developed or

⁹ For very narrow-band pyrometers and for full-radiation pyrometers, correction in accordance with (1) and (2) is quite acceptable, but with certain reservations: firstly, it is still necessary to know the temperature-spectral emissivity of the object, and secondly, the radiation coefficients depend on temperature, sometimes strongly, and for the choice of their exact values require knowledge of the very temperature of the object for which they are needed to measure. The latter greatly limits the correction according to (1) and (2) for accurate measurements.

absent. Therefore, it is necessary to solve the following scientific problems.

1. As is known, when measuring of “non-gray” objects with spectral-ratio pyrometers, they have a methodical error, the mechanism of which is discussed in detail in [11]. The ratio (3) is known, which allows (knowing the spectral emissivity) to compensate for this methodical error:

$$\frac{1}{T_d} - \frac{1}{T_{sp.rel}} = \ln \frac{\varepsilon_{\lambda_1}}{\varepsilon_{\lambda_2}} \frac{1}{c_2} \frac{1}{\frac{1}{\lambda_1} - \frac{1}{\lambda_2}} \quad (3)$$

where T_d is the actual temperature, K; $T_{sp.rel}$ is the temperature of the spectral ratio measured by the pyrometer, K; $c_2 = 1.4380 \cdot 10^{-2} \text{ m} \cdot \text{K}$; λ_1 and λ_2 are the operating wavelengths of the narrowband pyrometer of

the spectral ratio, m; ε_{λ_1} and ε_{λ_2} are the emission coefficients at wavelengths λ_1 and λ_2 .

However, this ratio is valid only for pyrometers with narrow (no more than 10...20nm) spectral bands.

At the same time, the vast majority of spectral-ratio pyrometers produced today are broadband, the width of the spectral bands of their sensitivity is tens or even hundreds of nanometers. As a result, ratio (3) is essentially inapplicable for the absolute majority of pyrometers used in practice, and has more theoretical than practical value. Therefore, a universal method is needed to correct the methodical error that occurs when measuring the temperature of “non-gray” objects with any pyrometers of spectral ratio. This method is developed and described in [6, 12]. They present an algorithm for machine calculation of the temperature of the spectral ratio of “non-gray” objects using a calibration function, and an experimental study of the method is carried out. However, the proposed method (as well as in the ratio (3)) does not take into account the temperature dependence of the spectral emissivity. Therefore, the method needs to be improved, taking into account this dependence.

Thus, the problem can be formulated as follows: the above-mentioned universal method for correcting pyrometers of the spectral ratio must be improved in such a way as to take into account the temperature dependence of the spectral emissivity. After that, it, together with information about the spectral emissivity, will become necessary and sufficient conditions for minimizing/eliminating methodical errors in the method of pyrometry of the spectral ratio.

The solution of this problem is described by the author in [13].

2. In contrast to the spectral ratio pyrometry method, the emissivity correction is fundamentally necessary in the energy pyrometry method. To do this, before

measuring, a correction factor is introduced into the energy pyrometer, which in this work is called the radiation coefficient. The radiation coefficients introduced into pyrometers are almost universally determined experimentally, by adjusting this coefficient to the value at which the result of temperature measurement using a pyrometer is close to the result of measurement by contact methods. Once selected in this way, the radiation coefficient is then usually transferred to all pyrometers that have to measure such an object. The measurement errors caused by such a transfer are described in [14]. And, moreover, with this approach, it is impossible to correctly take into account not only the spectral range of the pyrometer used, but also the temperature dependence of the emissivity. And this in turn leads to the appearance of additional methodical errors described in [15].

In [16], a ratio is given that allows the recognition of the spectral emissivity and spectral sensitivity characteristics of a photodiode pyrometer to correctly determine the radiation coefficient:

$$\varepsilon_{\lambda,T} = \frac{\int_{\lambda_1}^{\lambda_2} \varepsilon(\lambda, T) S(\lambda) E(\lambda, T) d\lambda}{\int_{\lambda_1}^{\lambda_2} S(\lambda) E(\lambda, T) d\lambda} \quad (4)$$

where $\varepsilon_{\lambda,T}$ is the radiation coefficient at wavelength λ for temperature T , $\varepsilon(\lambda, T)$ is the spectral emissivity of the object; $S(\lambda)$ is the spectral characteristic of the pyrometer sensitivity; $E(\lambda, T)$ is the Planck function; λ_1 and λ_2 are the lower and upper limits of spectral sensitivity.

Since (4) represents the ratio of two definite integrals that are practically insoluble analytically, its use in practice by metrologists and technologists of enterprises is hardly possible – for this, a specialist must have a legally purchased package such as MathCad or Matlab and be able to use it. Therefore, it is necessary to develop simple and freely distributed programs with which users with minimal computer skills could determine the radiation coefficient according to (4). One of the variants of the set of such programs is given in [17].

Further, since $\varepsilon(\lambda, T)$ and $E(\lambda, T)$ depend on the temperature of the object, the coefficient $\varepsilon_{\lambda,T}$ also depends on temperature. That is, the radiation coefficient found using (4) depends on the temperature. In this case, a vicious circle arises – in order to measure the temperature correctly with an energy pyrometer, you need to enter the correct value of the radiation coefficient into it. But to find the correct value of the radiation coefficient, you need to know the temperature

of the object, which we are still only going to measure. The established practice of adjusting the radiation coefficient to the correct result, if this correct result is unknown in advance, does not solve the problem.

It follows from the above that for the accurate correction of energy pyrometers for emissivity, not only the ratio (4) is required, but also a preliminary knowledge of the temperature to be measured, because without this it is impossible to correctly select those $\varepsilon(\lambda, T)$ and $E(\lambda, T)$ that are necessary to calculate $\varepsilon_{\lambda, T}$ according to (4).

Therefore, the task can be formulated as follows: for energy pyrometers, it is necessary to develop a method of correction for emissivity, different from the one currently used, in which there is no need for prior knowledge of the temperature to be measured in order to correctly use the temperature-dependent radiation coefficient $\varepsilon_{\lambda, T}$ i.e. it is necessary to break this vicious circle when you need to know its correct value to measure temperature, and in order to calculate it correctly in accordance with (4), we need to know this temperature, which is still unknown to us. The current method of correction is not capable of breaking it without some additional information.

It is the above-mentioned method of correction, which differs from the currently used one, in combination with knowledge of temperature-dependent spectral emissivity, that will be the necessary and sufficient means to minimize/exclude methodical errors in the method of energy pyrometry.

The solution of the mentioned problem is planned by the author to be published in one of the next issues of one of the periodicals covering measuring topics. A general approach to solving this problem is formulated in [18].

The implementation of solutions to the formulated tasks will dramatically reduce the methodical errors in pyrometry to a level comparable to the level achieved by instrumental errors.

IX. CONCLUSION

1. The main methodological postulate specific to pyrometry is formulated – the implicitly postulated priority of radiation laws in this branch over all other laws and patterns used to determine the temperature of heated bodies. It is shown that it is the unconscious adherence to this postulate that does not allow for half a century to solve the problem of reducing/eliminating methodical errors in non-contact temperature control methods.
2. A new, alternative to the above, basic methodological postulate specific to pyrometry is formulated. He proclaims that without knowledge and use of the real (not generalized or averaged!) the temperature-spectral emissivity of a particular measured object it is impossible to exclude

methodical errors in the non-contact measurement of its temperature.

3. Since to date there has not been a theory that adequately connects the temperature-spectral emissivity with the physico-chemical constants of the object's material, it is argued that it will be necessary to obtain the necessary information about the spectral emissivity experimentally. At the same time, it should be noted that at the moment there are no specialized measuring instruments for spectral emissivity on the market. Nevertheless, in a number of the author's works, the technical possibility of their creation is demonstrated, two such devices are described, one of which is protected by a patent of the Russian Federation.
4. However, knowledge of the spectral emissivity is only a necessary condition for reducing or completely eliminating large methodical errors inherent in pyrometry methods. For sufficiency, it is necessary to develop algorithms and methods for accounting for the effect on the spectral emissivity of the temperature of the measured object, which are now either insufficiently developed or completely absent.
5. References are given to the algorithms developed by the author of this article for taking into account the influence of temperature on the spectral emissivity of an object used in the methods of spectral ratio pyrometry and energy pyrometry.
6. The algorithms noted in paragraph 5 (recognition of temperature-spectral emissivity), when implemented, will reduce the methodical errors of pyrometry methods to a level comparable to the level achieved by instrumental errors.

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Studies of the Mechanism of Adhesion of Polymer Coatings on the Oxidized Surface of Aluminum and Magnesium Alloys

By Makarychev Yu. B., Kuzenkov Yu. A., Chugunov D. O. & Grafov O. Yu.

Abstract- The adhesion of the polymer paint Lakroten E-244 was investigated on oxidized aluminum A5154 and magnesium MA-20 alloys. It is shown that the adhesion of the polymer coating on the chemically oxidized surface of the aluminum alloy is significantly greater than on the anodic film of the magnesium alloy. Using photoelectron XPS spectroscopy and SEM analysis, the chemical composition and morphology of the alloys have been investigated. The dependence of these factors on the adhesion force of the polymer coating to the oxidized surface of aluminum and magnesium alloys has been established. Methods of modification of conversion coatings with ethylen glycol, which significantly improve the adhesion of polymer layers, have been proposed. The chemical composition and morphology of these coatings have been investigated. An explanation of the mechanism of improvement of adhesion properties of modified coatings on aluminum and magnesium alloys is given. The mechanism of improving the adhesive properties of modified coatings on aluminum and magnesium alloys is explained.

Keywords: *adhesion strength, x-ray photoelectron spectroscopy, conversion coatings, paints.*

GJRE-A Classification: LCC: TP1180, QD549



STUDIES OF THE MECHANISM OF ADHESION OF POLYMER COATINGS ON THE OXIDIZED SURFACE OF ALUMINUM AND MAGNESIUM ALLOYS

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Studies of the Mechanism of Adhesion of Polymer Coatings on the Oxidized Surface of Aluminum and Magnesium Alloys

Makarychev Yu. B. ^α, Kuzenkov Yu. A. ^σ, Chugunov D. O. ^ρ & Grafov O. Yu. ^ω

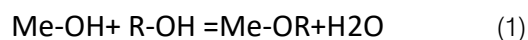
Abstract- The adhesion of the polymer paint Lakroten E-244 was investigated on oxidized aluminum A5154 and magnesium MA-20 alloys. It is shown that the adhesion of the polymer coating on the chemically oxidized surface of the aluminum alloy is significantly greater than on the anodic film of the magnesium alloy. Using photoelectron XPS spectroscopy and SEM analysis, the chemical composition and morphology of the alloys have been investigated. The dependence of these factors on the adhesion force of the polymer coating to the oxidized surface of aluminum and magnesium alloys has been established. Methods of modification of conversion coatings with ethylen glycol, which significantly improve the adhesion of polymer layers, have been proposed. The chemical composition and morphology of these coatings have been investigated. An explanation of the mechanism of improvement of adhesion properties of modified coatings on aluminum and magnesium alloys is given. The mechanism of improving the adhesive properties of modified coatings on aluminum and magnesium alloys is explained.

Keywords: adhesion strength, x-ray photoelectron spectroscopy, conversion coatings, paints.

I. INTRODUCTION

Aluminum and magnesium alloys are widely used as structural materials in the aerospace, electronics and automotive industries due to their low density and high strength. The disadvantage of these alloys is their low corrosion resistance in aqueous electrolytes, which limits their use without special preparation. To protect aluminum and magnesium alloys from corrosion, paints and varnishes are commonly used, which are applied to conversion coatings. To improve the adhesion of aluminum and magnesium alloys to paints, alkaline treatment is often recommended to increase the concentration of -Me-OH atomic groups on their surface. To enhance the adhesive activity of aluminum and magnesium alloys to paints, alkaline treatment is often recommended in order to increase the concentration of Me-OH groups of atoms on their surface. At the same time, an increase in the adhesive force is associated with the chemical interaction of surface metal hydroxides with paintwork components. Most coatings in their composition contain

monomers with the general formula R-OH and R-COOH, where R is aliphatic or cyclic hydrocarbons. In this case, the chemical interaction of the paint with metal occurs as a result of the condensation reaction, which can be represented in the form of equations:



Thus, for the paint to form chemical bonds with the conversion coating, chemically active hydroxyl groups of atoms must be present in its composition. Most studies on the modification of conversion coatings are aimed at changing their structure rather than their chemical composition, based on the assumption that the main factor that increases adhesion is the mechanical bonding of the paint to the pore walls. However, the rapid development of technologies based on the use of organosilane primers indicates the significant role of chemical interaction in the formation of strong bonds of coating materials with the surface of metals. Conversion coatings on aluminum and magnesium alloys consist mainly of metal oxides and hydroxides. Only metal hydroxides participate in the formation of chemical bonds by the mechanism of hydrolytic condensation, so the adhesion strength should depend on the MeO/MeOH ratio in the conversion coating. In this work, the chemical compositions of anodized layers on aluminum A5154 and magnesium MA-20 alloys will be investigated by XPS methods in order to determine the influence of the chemical composition of conversion coatings on the adhesion force to paints. The coatings impregnation was carried out with acrylic paint based on Lakroten E-244 dispersion. The copolymer of this dispersion is acrylic acid which can interact with metal hydroxides by reaction (2).

II. MATERIAL AND METHODS

The work used samples of aluminum and magnesium alloys, the composition of which is shown in Table 1.

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Table 1: Compositions of aluminum alloy A 5154 and magnesium alloy MA-20

Alloy grade	Mass fraction of elements, %							
	Fe	Cu	Mn	Mg	Cr	Zn	Ce	Al
A5154	0.25	0.4	0.1	0.1	3.1	15	0.2	93.8
MA20	0.05	0.03	0.04	97.94	1.5	0.12	0.02	0.2

The following substances were used to compose inhibitory compositions:

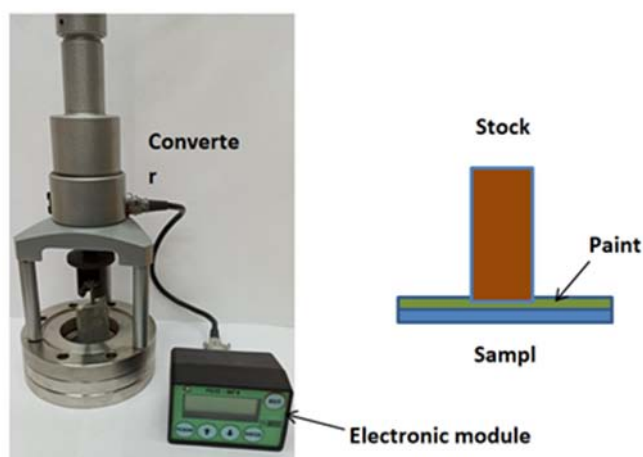
- Ethylene glycol (EG), $\text{HO}-\text{CH}_2-\text{CH}_2-\text{OH}$ (Chem.Russia);
- Vinyltrimethoxysilane (VS) (WitcoCo);

Compositions of oxidation electrolytes of aluminum alloy A 5154 and magnesium alloy MA-20 is shown in Table 2.

Table 2: Compositions of oxidation electrolytes of aluminum alloy A 5154 and magnesium alloy MA-20

Sample	Conversion Coating	Modified Conversion Coating
A5154	35g/l NaH_2PO_4 , 10g/l AlNO_3 , 5 г/л NaOH	35g/l NaH_2PO_4 , 10g/l AlNO_3 , 5 г/л NaOH, 50 г/л EG
MA20	55g/l NaOH, 35g/l Na_2HPO_4 , 10g/l $\text{Al}(\text{NO}_3)_3$	55g/l NaOH, 35g/l Na_2HPO_4 , 10g/l $\text{Al}(\text{NO}_3)_3$, 50 г/л EG

a) Adhesion Measurement

*Fig. 1:* Electronic adhesion meter PSO-MG4g

The coating adhesion test during separation was carried out using an electronic adhesion meter PSP-MP4 (Russia), shown in Fig. 1. Samples with conversion coating were placed in a fixed holder. Super glue was applied to the surface of an aluminum rod with a diameter of 10 mm and then brought into contact with the test samples. After curing the glue, the joint was stretched at a constant speed of 5 mm/min. For each test, five repeating samples with the specified mean value were used.

b) X-ray Photoelectron Spectroscopy (XPS) Study

X-ray photoelectron spectra (XPS) were recorded using the Omicron+ X-ray photoelectron spectrometer (FRG). The pressure in the analyzer chamber did not exceed 10^{-8} Torr. The radiation of an X-ray Al-anode (radiation energy 1486.6 eV, power 200 W) was used as a source. The transmission energy of

the analyzer is set to 20 eV. The binding energy of the electrons was calibrated according to the XPS peak of C1s electrons, the binding energy of which was assumed to be 285.0 eV and which is due to the settled layers of diffusion oil vapor. Photoionization cross sections of the corresponding electron shells taken from [19] were used for quantitative evaluation. The integral peak intensities were obtained after subtracting the background using the Shirley method [20] and by fitting the observed peaks with Gauss curves with the contribution of the Lorentz component. To obtain information about the thicknesses of the layers formed on the surface, the MultiQuant program was used [21], the photoionization sections of the corresponding electron shells given by Schofield [22] were selected. To calculate the thicknesses of the layers, the values of electron free path lengths (or average attenuation

coefficients) calculated by the method proposed by Kampson and Sih were used [23].

c) SEM of the Study

SEM images were obtained by a JSM-6400 scanning microscope with an electron-beam intensity of 20 keV, unless otherwise written. Analysis of the elements was performed with SEM equipped with energy-dispersive X-ray analysis (EDX) with a WinEDS EUMEX analyzer (Germany). The thickness of the polymer siloxane coatings was determined using the ZAF correction algorithm for Ka-ratios samples with a known coating thickness.

d) EIS Study

An electrochemical complex consisting of a potentiostat with a frequency analyzer (FRA) is used to measure the impedance. The measurements were carried out in a three-electrode electrochemical cell with a corrosion potential using the Solartron electrochemical complex with a change in the frequency of alternating current from 60,000 to 0.1 Hz ($\Delta E = 10$ Mv). The impedance data was analyzed using ZView® electrochemical impedance software (version Scribner Associates, Inc., USA).

III. EXPERIMENTAL

The adhesion of coating materials to metal surfaces depends mainly on the chemical composition

Table 3: The elemental composition of the conversion coating on the MA-20 alloy

Sample	Mg,% at	Al,% at	C,% at	O,% at	Thickness, mkm
MA-20	30.3	8.4	19.2	42.1	23.0±5.0

From the results of the chemical analysis given in Table 2, it can be seen that the composition of the conversion coating includes significant amounts of aluminum compounds that fall into the coating from the substrate or electrolyte during oxidation. Figure 2 shows the XPS spectra of Mg2p and Al2p in the conversion coating. After decomposition of the spectra into components, it is seen that the conversion coating consists mainly of magnesium and aluminum oxides. A

study of the adhesion of acrylic paint on anodised magnesium alloy surfaces showed extremely low adhesion strength. To increase the adhesive strength of aluminium and magnesium alloys, it is often recommended to hydroxylate the surface by treatment in alkaline solutions. It is possible to increase the concentration of adhesively active R-OH radicals in the conversion coating by adding polyfunctional alcohols to the anodizing electrolytes.

a) XPS and SEM Investigation of the Chemical Composition of the Conversion Coating on the MA-20 Alloy

The elemental composition of the conversion coating obtained by anodizing the MA-20 alloy in an alkaline solution is shown in Table 3.

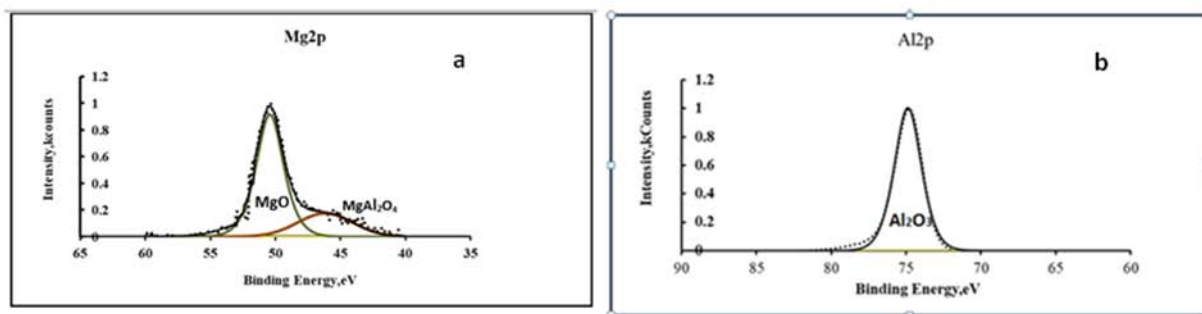


Fig. 2: X-ray spectra of Mg2p (a) and C1s (b) in the conversion coating

In the articles [26,27], studies were carried out on the anodizing of magnesium alloys in aqueous solutions of ethylene glycol and glycerin. The authors of these works found that the conversion coatings obtained in these solutions are formed at sufficiently high speeds, have a porous structure, have high hardness and wear resistance. However, studies of the

chemical composition and adhesive strength were not carried out in these works. Morphology and chemical composition of the magnesium layers obtained with ethylene glycol additives in anodizing solutions were studied. Figure 4 shows SEM images of the surface of the anodized alloy before and after modification with ethylene glycol.

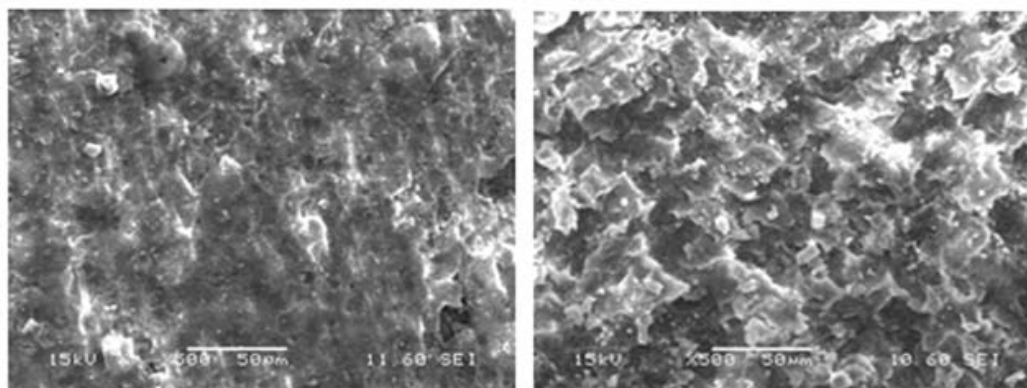


Fig. 3: SEM images of the surface of the anodized alloy before and after modification with ethylene glycol

The addition of ethylene glycol to the electrolyte leads to an increase in the roughness and thickness of the coating. The increase in roughness is apparently due to the formation of magnesium glycolate

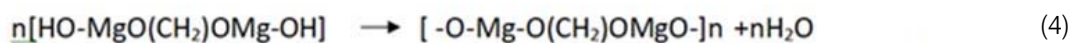
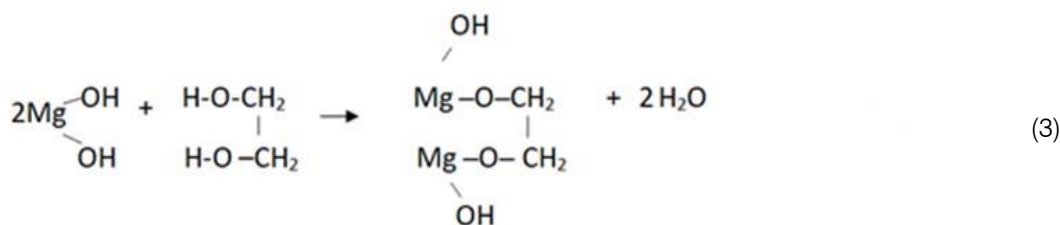
macromolecules [29] linked together by a bridging bond. Table 4 shows the chemical composition of the modified composite coating.

Table 4: Elemental composition of the modified composite coating

Sample	Mg, % at.	Al, % at.	C, % at.	O, % at.	Thickness, mkm
MA-20	22,3	2.4	43.2	32,1	32.4.0±5.0

The results of chemical analysis show that the composite coating contains a significant amount of carbon. According to [29], magnesium hydroxide molecules interact with ethylene glycol to form

magnesium ethylene glycolate by reaction (3). Magnesium glycolates can interact with each other to form macromolecules by reaction (4).



Macromolecules, together with magnesium oxides, are embedded in the structure of globules that make up conversion coating [29,30]. Polyglycolates contain in their composition adhesive active groups – OH, through which interaction with paints and individual globules with each other is carried out. On the XPS spectra of Mg2, magnesium ethylene glycolates can be

observed by the appearance of a peak with $E_b = 53.5$ eV (Fig. 5), and on the spectra of C1s with $E_b = 288.2$ eV (Fig. 6).

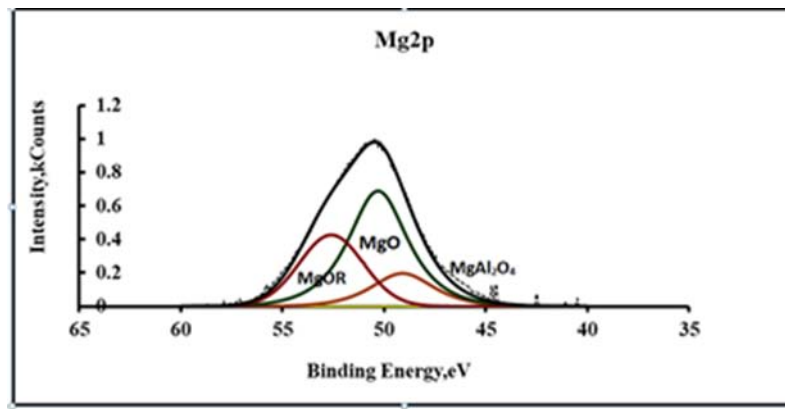


Fig. 4: XPS spectra of Mg2p in a modified inversion coating

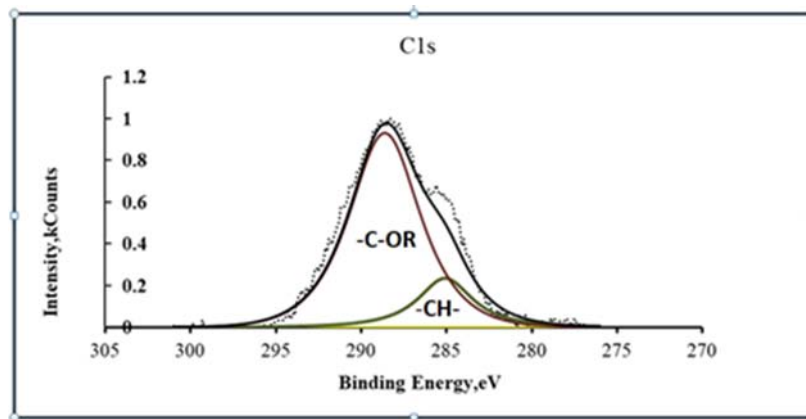


Fig. 5: XPS spectra of C1S carbon in a modified conversion coating

The amounts of ethylene glycolates and magnesium hydroxides in the original and modified conversion coating are shown in Table 5.

Table 5: Chemical composition and tensile strength of the initial and modified MA-20 alloy

Type of coating	MgO, % at.	Mg glycolate% at.	MgAlO ₄ % at.	Adhesion strength KPa
1. Initial coating	84,6	-	15.4	300±50
2. Modified	57.2	36.2	18.6	1180±50

The inclusion of magnesium glycolates in the conversion coating should change the concentration of chemically active groups -OH. Figure 6 shows the XPS spectra of O1s oxygen in the modified conversion coating.

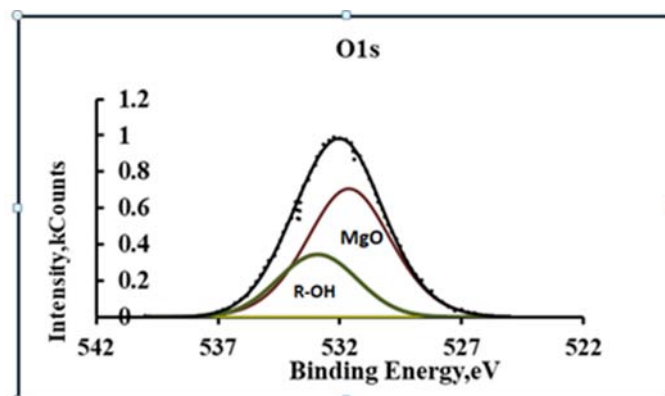


Fig. 6: XPS-spectra of O1s oxygen in the modified conversion coating

Quantitative calculations have shown that the ratio of $[O]/[OH]$ concentrations in the conversion coating changes towards an increase in hydroxide ions mainly due to magnesium ethylene glycolate. An increase in the adhesive strength of the modified conversion coating may be due to an increase in roughness or due to the formation of chemical bonds

with the paint. With the help of vapor-gas deposition of vinyltrimethoxysilane (VS) [31], its roughness was reduced on the surface of the modified conversion coating. The coating thickness was 150-200 nm. To assess the roughness of the conversion coating, EIS studies were conducted.

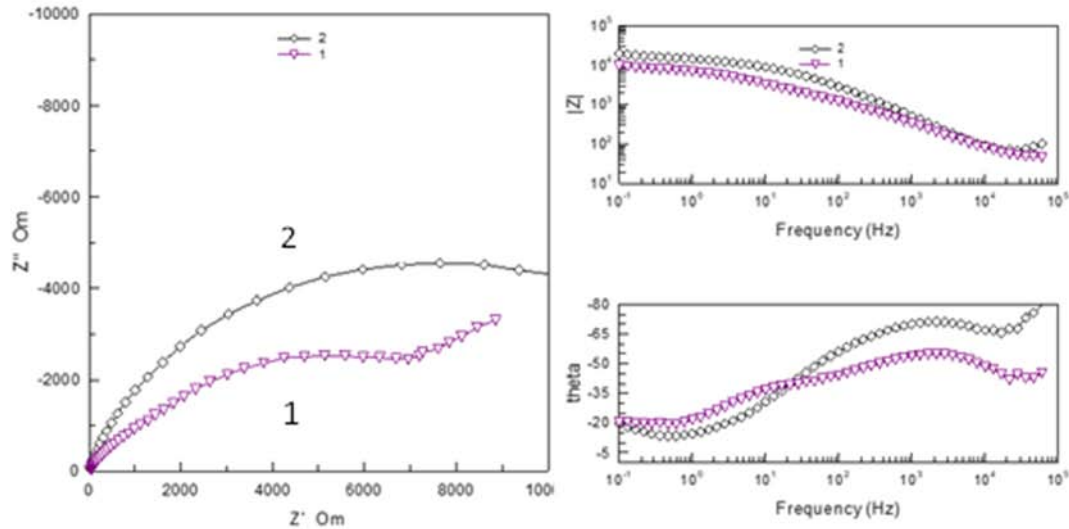


Fig. 7: Nyquist and Bode plots for modified conversion coating (plot 1) and after VS impregnation (plot 2)

In plot 1 (Fig. 7), there is a rectilinear section with an angle of 45° in the high frequency region. This form of the Nyquist graph is characteristic of mass transfer in porous systems. After impregnation, the Nyquist graph has the form of a hemisphere, characteristic of a smoothed surface. The Bode graphs for the initial conversion coverage show the presence of two well-defined time constants. In the high frequency range the time constant can be explained by the barrier properties of coatings. In the low-frequency region, 10-50 Hz is attributed to corrosion processes at the boundary of the coating and metal. After the vapor-gas modification of the conversion coverage, the through pores are filled with polymer and a wide area of change in the phase angle of the barrier coating remains. Silane

coating practically does not change $|Z|$, which indicates partial filling of the pores and a slight change in the structure of the coating. The decrease in the roughness and pore volume of the conversion coverage should have reduced the adhesive force at the same time, it increased after steam-gas treatment by $\sim 20-30\%$, which indicates the predominant influence of chemical interaction over mechanical. Impregnation of the coatings with vinylsilane increases the number of functional groups involved in the formation of chemical bonds in connection with which it can be used to enhance the adhesion of the coating with paint. Table 6 shows the adhesion strength of Lakroten E-244 to oxidized alloys MA-20 and A5154 with paint.

Table 6: Adhesive strength of Lakroten E-244 to oxidized alloys MA-20 and A5154

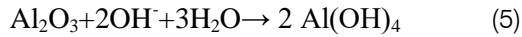
Conversion coverage on MA-20	Adhesion strength, KPa
1. Initial	300 ± 50
2. Modified with ethylene glycol	1180 ± 50
3. Modified with ethylene glycol+VS	1350 ± 100
Conversion coverage on A 5154	
4. Initial	1600 ± 100
5. Conversion coverage on A 5154	1850 ± 100

From the data given in Table 6, it can be seen that the adhesion of Lakroten E-244 on the modified chemically oxidized alloy A5154 is maximum. To clarify

the influence of the morphology and chemical composition of the aluminum alloy on the adhesion of the paint, XPS and SEM studies were conducted.

b) *XPS and SEM Study of the Chemical Composition of the Conversion Coating on Aluminum Alloy A5154*

The conversion coating on chemically oxidized alloy A5154 consists mainly of $\text{Al}(\text{OH})_4$, which is obtained by the equation:



This explains the high adhesion of coatings to these alloys. Chemical oxidation is used along with electrochemical, but is inferior to the latter in terms of performance characteristics, primarily mechanical strength and wear resistance. Modification of a chemically oxidized conversion coating with ethylene

glycol can increase its mechanical properties due to additional chemical bonds between individual amorphous $\text{Al}(\text{OH})_4$ globules of which conversion coating consists [32]. Along with high adhesive activity, such coatings could significantly expand their application areas.

The conversion coating on alloy A 5154 was modified with ethylene glycol by adding it to oxidation solutions. Figure 8 shows a SEM images of the surface of the conversion coating on the original and modified alloy.

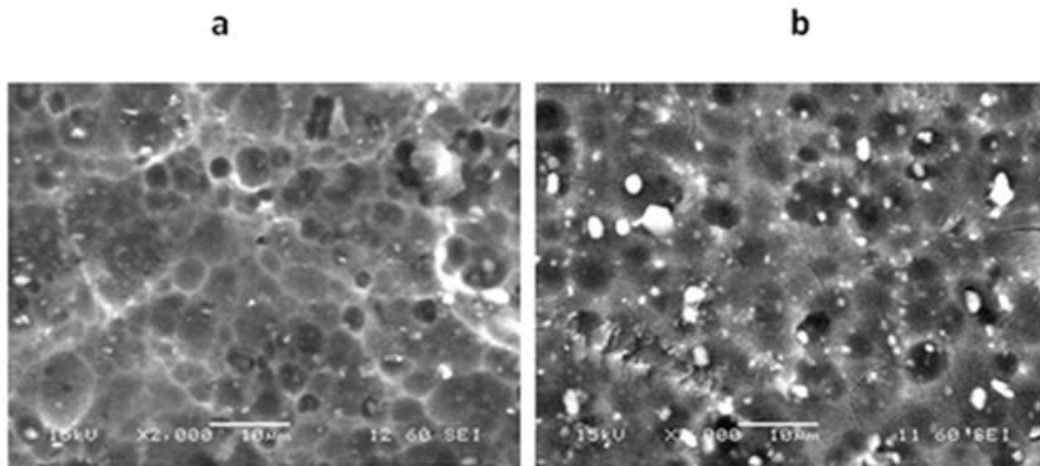


Fig. 8: SEM images of the surface of the conversion coating on the original (a) and modified (b) alloy

It can be seen that the structure of conversion coating A5154 is fine-grained with vertical pores of 5-10 microns in size, in contrast to magnesium MA-20. On the modified alloy the structure is also granular but the

number and diameter of the pores are larger. Inclusions of white colour according to X-ray microanalysis data are microphases of magnesium glycolate. The chemical composition of the alloys is given in Table 7.

Table 7: Chemical composition of the initial and modified conversion coating on alloy A5154

Conversion coating on alloy A5154	Mg, % at.	Al, % at.	C, % at.	O, % at.	Thickness, mKm
Initial	2.3	14.4	25.1	58.2	3.2
Modified	1.8	11.6	42.8	43.8	2.8

Comparing the chemical analysis data of the original and modified coating, it is clear that the latter contains a significant amount of carbon belonging to magnesium and aluminum glycolates. Aluminum glycolates can be detected by the appearance of a peak on the Al 2p spectra with an energy of 71.4 eV and a peak with an energy of 288.2 eV on the C1s spectra (Fig. 9).

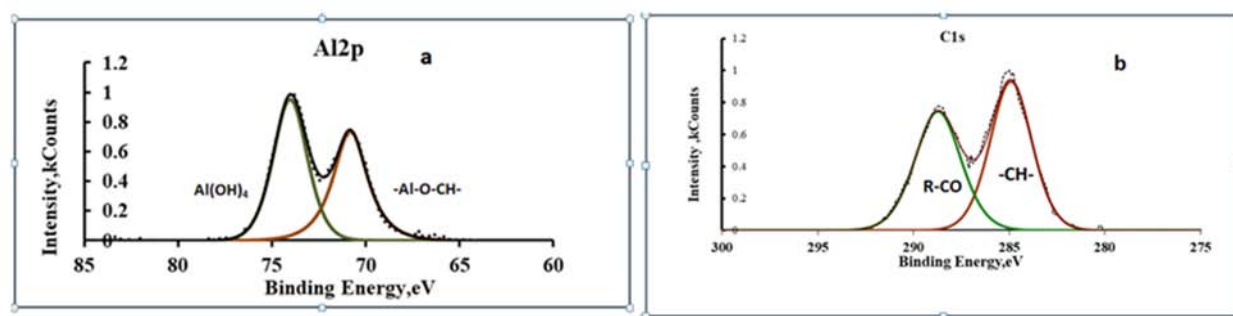
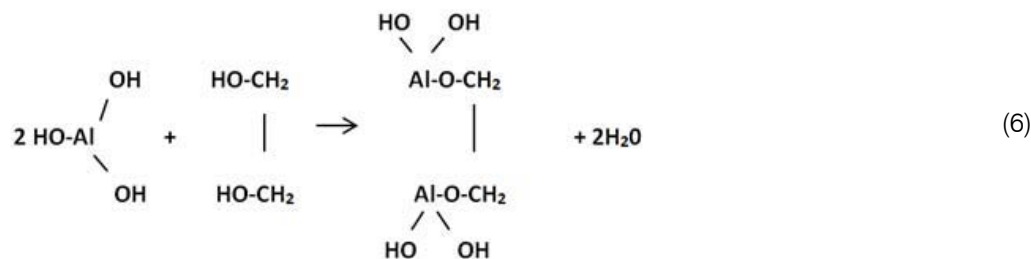


Fig. 9: XPS spectra of Al 2p(a) and C1s (b) for a modified conversion coating on alloy A 5154

A symmetrical doublet on the Al2p spectrum indicates the existence of two types of chemical compounds of aluminum distributed in the conversion coating as separate phases. One of these phases is Al(OH)_3 , the other is aluminum glycolate. Quantitative calculations of the XPS spectra belonging to aluminum

glycolates have shown that two aluminum atoms account for approximately the same number of carbon atoms. This ratio corresponds to the reaction of the formation of aluminum glycolate according to equation (6).



Aluminum glycolate molecules can interact with each other to form macromolecules and individual phases in the conversion coating. A significant amount of aluminum glycolate in the coating indicates a high chemical activity of ethylene glycol to aluminum hydroxides. The inclusion of aluminum glycolate in the composition of the conversion coating changes its structure and possibly mechanical properties in the direction of increasing strength due to the formation of new chemical bond. The adhesive strength of the modified conversion coating is slightly higher than the initial one due to mechanical adhesion, since the concentration of chemically active $-\text{OH}$ groups varies slightly.

IV. RESULTS AND DISCUSSION

Published studies of paint adhesion to aluminium and magnesium alloys are random and adhesion strength is mainly attributed to mechanical adhesion in the pore volume. At the same time, our studies have shown that the adhesion strength of paint to the surface of these alloys can vary by an order of magnitude. In this work, the surface of the alloys was subjected to chemical and anodic oxidation. Conversion

coatings on aluminium alloy A5154 were obtained by chemical oxidation in alkaline solutions. Oxide coatings on magnesium alloys Ma-20 were obtained by anodising in alkaline solutions. From the point of view of the theory of mechanical adhesion of paints with the alloy surface, conversion coatings on Ma-20 alloy are more preferable because they have a more developed structure on the surface and inside the oxide layer. However, studies have shown that the adhesion force on aluminium alloys is 5-6 times greater than on magnesium alloys. It has been suggested that this is due to the chemical composition of the conversion layers. On aluminium alloys they consist of aluminium hydroxides Al(OH)_3 and on magnesium alloys of oxides. According to equations (1,2) the formation of strong chemical bonds with paints is possible only on aluminium alloy. Modification of the conversion coating with ethylene glycol leads to an increase in its concentration of $-\text{OH}$ groups capable of chemical interaction with paintwork materials, as a result of which the adhesive force increases. It is possible to increase the concentration of hydroxides in the coating by introducing compounds that form $-\text{R-OH}$ groups of atoms as a result of hydrolysis, such as organosilanes,

into the anodising electrolyte. Modification of aluminium alloys with ethylene glycol does not lead to a significant increase in adhesion strength because the conversion coating consists of aluminium hydroxides. However, there is an increase in the number and size of pores, which increases the adhesive strength.

V. CONCLUSIONS

Studies of the adhesion of the acrylic paint Lakrotan E-244 to aluminum A5154 and magnesium Ma-20 alloys have been carried out. Significant differences in the adhesive activity of these alloys to polymer coatings were found. With the help of XPS and SEM, studies of the chemical composition and morphology of alloys were carried out. It is shown that the adhesion of coatings to the surface of alloys depends mainly on the ratio of MeO/MeO in the conversion coating. Modification of the conversion coating on a magnesium alloy with ethylene glycol significantly increases the adhesive strength of the alloy to paints. The chemical composition of magnesium ethylene glycolate macromolecules and their amount in the conversion coating were determined with the help of XPS studies. The addition of ethylene glycol increases the thickness of the conversion coating and its strength due to the formation of additional chemical bonds of magnesium ethylene glycol macromolecules with inside and on the border of globules. Studies of the chemical composition, morphology and adhesive activity of conversion coatings on the surface of aluminum alloys modified with ethylene glycol have been carried out. It is shown that the increase in the adhesive activity of modified aluminum alloys is mainly due to an increase in the number and size of pores. The chemical composition and quantity have been established macromolecules of aluminum ethylene glycolate in the conversion coating of aluminum alloy. Modification of the ethylene glycol conversion coating on magnesium and aluminum alloys changes their structure and chemical composition, which contributes to a significant increase in their adhesive activity.

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- c) Up to 10 keywords that precisely identify the paper's subject, purpose, and focus.
- d) An introduction, giving fundamental background objectives.
- e) Resources and techniques with sufficient complete experimental details (wherever possible by reference) to permit repetition, sources of information must be given, and numerical methods must be specified by reference.
- f) Results which should be presented concisely by well-designed tables and figures.
- g) Suitable statistical data should also be given.
- h) All data must have been gathered with attention to numerical detail in the planning stage.

Design has been recognized to be essential to experiments for a considerable time, and the editor has decided that any paper that appears not to have adequate numerical treatments of the data will be returned unrefereed.

- i) Discussion should cover implications and consequences and not just recapitulate the results; conclusions should also be summarized.
- j) There should be brief acknowledgments.
- k) There ought to be references in the conventional format. Global Journals recommends APA format.

Authors should carefully consider the preparation of papers to ensure that they communicate effectively. Papers are much more likely to be accepted if they are carefully designed and laid out, contain few or no errors, are summarizing, and follow instructions. They will also be published with much fewer delays than those that require much technical and editorial correction.

The Editorial Board reserves the right to make literary corrections and suggestions to improve brevity.



FORMAT STRUCTURE

It is necessary that authors take care in submitting a manuscript that is written in simple language and adheres to published guidelines.

All manuscripts submitted to Global Journals should include:

Title

The title page must carry an informative title that reflects the content, a running title (less than 45 characters together with spaces), names of the authors and co-authors, and the place(s) where the work was carried out.

Author details

The full postal address of any related author(s) must be specified.

Abstract

The abstract is the foundation of the research paper. It should be clear and concise and must contain the objective of the paper and inferences drawn. It is advised to not include big mathematical equations or complicated jargon.

Many researchers searching for information online will use search engines such as Google, Yahoo or others. By optimizing your paper for search engines, you will amplify the chance of someone finding it. In turn, this will make it more likely to be viewed and cited in further works. Global Journals has compiled these guidelines to facilitate you to maximize the web-friendliness of the most public part of your paper.

Keywords

A major lynchpin of research work for the writing of research papers is the keyword search, which one will employ to find both library and internet resources. Up to eleven keywords or very brief phrases have to be given to help data retrieval, mining, and indexing.

One must be persistent and creative in using keywords. An effective keyword search requires a strategy: planning of a list of possible keywords and phrases to try.

Choice of the main keywords is the first tool of writing a research paper. Research paper writing is an art. Keyword search should be as strategic as possible.

One should start brainstorming lists of potential keywords before even beginning searching. Think about the most important concepts related to research work. Ask, "What words would a source have to include to be truly valuable in a research paper?" Then consider synonyms for the important words.

It may take the discovery of only one important paper to steer in the right keyword direction because, in most databases, the keywords under which a research paper is abstracted are listed with the paper.

Numerical Methods

Numerical methods used should be transparent and, where appropriate, supported by references.

Abbreviations

Authors must list all the abbreviations used in the paper at the end of the paper or in a separate table before using them.

Formulas and equations

Authors are advised to submit any mathematical equation using either MathJax, KaTeX, or LaTeX, or in a very high-quality image.

Tables, Figures, and Figure Legends

Tables: Tables should be cautiously designed, uncrowned, and include only essential data. Each must have an Arabic number, e.g., Table 4, a self-explanatory caption, and be on a separate sheet. Authors must submit tables in an editable format and not as images. References to these tables (if any) must be mentioned accurately.



Figures

Figures are supposed to be submitted as separate files. Always include a citation in the text for each figure using Arabic numbers, e.g., Fig. 4. Artwork must be submitted online in vector electronic form or by emailing it.

PREPARATION OF ELETRONIC FIGURES FOR PUBLICATION

Although low-quality images are sufficient for review purposes, print publication requires high-quality images to prevent the final product being blurred or fuzzy. Submit (possibly by e-mail) EPS (line art) or TIFF (halftone/ photographs) files only. MS PowerPoint and Word Graphics are unsuitable for printed pictures. Avoid using pixel-oriented software. Scans (TIFF only) should have a resolution of at least 350 dpi (halftone) or 700 to 1100 dpi (line drawings). Please give the data for figures in black and white or submit a Color Work Agreement form. EPS files must be saved with fonts embedded (and with a TIFF preview, if possible).

For scanned images, the scanning resolution at final image size ought to be as follows to ensure good reproduction: line art: >650 dpi; halftones (including gel photographs): >350 dpi; figures containing both halftone and line images: >650 dpi.

Color charges: Authors are advised to pay the full cost for the reproduction of their color artwork. Hence, please note that if there is color artwork in your manuscript when it is accepted for publication, we would require you to complete and return a Color Work Agreement form before your paper can be published. Also, you can email your editor to remove the color fee after acceptance of the paper.

TIPS FOR WRITING A GOOD QUALITY ENGINEERING RESEARCH PAPER

Techniques for writing a good quality engineering research paper:

1. Choosing the topic: In most cases, the topic is selected by the interests of the author, but it can also be suggested by the guides. You can have several topics, and then judge which you are most comfortable with. This may be done by asking several questions of yourself, like "Will I be able to carry out a search in this area? Will I find all necessary resources to accomplish the search? Will I be able to find all information in this field area?" If the answer to this type of question is "yes," then you ought to choose that topic. In most cases, you may have to conduct surveys and visit several places. Also, you might have to do a lot of work to find all the rises and falls of the various data on that subject. Sometimes, detailed information plays a vital role, instead of short information. Evaluators are human: The first thing to remember is that evaluators are also human beings. They are not only meant for rejecting a paper. They are here to evaluate your paper. So present your best aspect.

2. Think like evaluators: If you are in confusion or getting demotivated because your paper may not be accepted by the evaluators, then think, and try to evaluate your paper like an evaluator. Try to understand what an evaluator wants in your research paper, and you will automatically have your answer. Make blueprints of paper: The outline is the plan or framework that will help you to arrange your thoughts. It will make your paper logical. But remember that all points of your outline must be related to the topic you have chosen.

3. Ask your guides: If you are having any difficulty with your research, then do not hesitate to share your difficulty with your guide (if you have one). They will surely help you out and resolve your doubts. If you can't clarify what exactly you require for your work, then ask your supervisor to help you with an alternative. He or she might also provide you with a list of essential readings.

4. Use of computer is recommended: As you are doing research in the field of research engineering then this point is quite obvious. Use right software: Always use good quality software packages. If you are not capable of judging good software, then you can lose the quality of your paper unknowingly. There are various programs available to help you which you can get through the internet.

5. Use the internet for help: An excellent start for your paper is using Google. It is a wondrous search engine, where you can have your doubts resolved. You may also read some answers for the frequent question of how to write your research paper or find a model research paper. You can download books from the internet. If you have all the required books, place importance on reading, selecting, and analyzing the specified information. Then sketch out your research paper. Use big pictures: You may use encyclopedias like Wikipedia to get pictures with the best resolution. At Global Journals, you should strictly follow [here](#).



6. Bookmarks are useful: When you read any book or magazine, you generally use bookmarks, right? It is a good habit which helps to not lose your continuity. You should always use bookmarks while searching on the internet also, which will make your search easier.

7. Revise what you wrote: When you write anything, always read it, summarize it, and then finalize it.

8. Make every effort: Make every effort to mention what you are going to write in your paper. That means always have a good start. Try to mention everything in the introduction—what is the need for a particular research paper. Polish your work with good writing skills and always give an evaluator what he wants. Make backups: When you are going to do any important thing like making a research paper, you should always have backup copies of it either on your computer or on paper. This protects you from losing any portion of your important data.

9. Produce good diagrams of your own: Always try to include good charts or diagrams in your paper to improve quality. Using several unnecessary diagrams will degrade the quality of your paper by creating a hodgepodge. So always try to include diagrams which were made by you to improve the readability of your paper. Use of direct quotes: When you do research relevant to literature, history, or current affairs, then use of quotes becomes essential, but if the study is relevant to science, use of quotes is not preferable.

10. Use proper verb tense: Use proper verb tenses in your paper. Use past tense to present those events that have happened. Use present tense to indicate events that are going on. Use future tense to indicate events that will happen in the future. Use of wrong tenses will confuse the evaluator. Avoid sentences that are incomplete.

11. Pick a good study spot: Always try to pick a spot for your research which is quiet. Not every spot is good for studying.

12. Know what you know: Always try to know what you know by making objectives, otherwise you will be confused and unable to achieve your target.

13. Use good grammar: Always use good grammar and words that will have a positive impact on the evaluator; use of good vocabulary does not mean using tough words which the evaluator has to find in a dictionary. Do not fragment sentences. Eliminate one-word sentences. Do not ever use a big word when a smaller one would suffice.

Verbs have to be in agreement with their subjects. In a research paper, do not start sentences with conjunctions or finish them with prepositions. When writing formally, it is advisable to never split an infinitive because someone will (wrongly) complain. Avoid clichés like a disease. Always shun irritating alliteration. Use language which is simple and straightforward. Put together a neat summary.

14. Arrangement of information: Each section of the main body should start with an opening sentence, and there should be a changeover at the end of the section. Give only valid and powerful arguments for your topic. You may also maintain your arguments with records.

15. Never start at the last minute: Always allow enough time for research work. Leaving everything to the last minute will degrade your paper and spoil your work.

16. Multitasking in research is not good: Doing several things at the same time is a bad habit in the case of research activity. Research is an area where everything has a particular time slot. Divide your research work into parts, and do a particular part in a particular time slot.

17. Never copy others' work: Never copy others' work and give it your name because if the evaluator has seen it anywhere, you will be in trouble. Take proper rest and food: No matter how many hours you spend on your research activity, if you are not taking care of your health, then all your efforts will have been in vain. For quality research, take proper rest and food.

18. Go to seminars: Attend seminars if the topic is relevant to your research area. Utilize all your resources.

19. Refresh your mind after intervals: Try to give your mind a rest by listening to soft music or sleeping in intervals. This will also improve your memory. Acquire colleagues: Always try to acquire colleagues. No matter how sharp you are, if you acquire colleagues, they can give you ideas which will be helpful to your research.

20. Think technically: Always think technically. If anything happens, search for its reasons, benefits, and demerits. Think and then print: When you go to print your paper, check that tables are not split, headings are not detached from their descriptions, and page sequence is maintained.



21. Adding unnecessary information: Do not add unnecessary information like "I have used MS Excel to draw graphs." Irrelevant and inappropriate material is superfluous. Foreign terminology and phrases are not apropos. One should never take a broad view. Analogy is like feathers on a snake. Use words properly, regardless of how others use them. Remove quotations. Puns are for kids, not grunt readers. Never oversimplify: When adding material to your research paper, never go for oversimplification; this will definitely irritate the evaluator. Be specific. Never use rhythmic redundancies. Contractions shouldn't be used in a research paper. Comparisons are as terrible as clichés. Give up ampersands, abbreviations, and so on. Remove commas that are not necessary. Parenthetical words should be between brackets or commas. Understatement is always the best way to put forward earth-shaking thoughts. Give a detailed literary review.

22. Report concluded results: Use concluded results. From raw data, filter the results, and then conclude your studies based on measurements and observations taken. An appropriate number of decimal places should be used. Parenthetical remarks are prohibited here. Proofread carefully at the final stage. At the end, give an outline to your arguments. Spot perspectives of further study of the subject. Justify your conclusion at the bottom sufficiently, which will probably include examples.

23. Upon conclusion: Once you have concluded your research, the next most important step is to present your findings. Presentation is extremely important as it is the definite medium through which your research is going to be in print for the rest of the crowd. Care should be taken to categorize your thoughts well and present them in a logical and neat manner. A good quality research paper format is essential because it serves to highlight your research paper and bring to light all necessary aspects of your research.

INFORMAL GUIDELINES OF RESEARCH PAPER WRITING

Key points to remember:

- Submit all work in its final form.
- Write your paper in the form which is presented in the guidelines using the template.
- Please note the criteria peer reviewers will use for grading the final paper.

Final points:

One purpose of organizing a research paper is to let people interpret your efforts selectively. The journal requires the following sections, submitted in the order listed, with each section starting on a new page:

The introduction: This will be compiled from reference matter and reflect the design processes or outline of basis that directed you to make a study. As you carry out the process of study, the method and process section will be constructed like that. The results segment will show related statistics in nearly sequential order and direct reviewers to similar intellectual paths throughout the data that you gathered to carry out your study.

The discussion section:

This will provide understanding of the data and projections as to the implications of the results. The use of good quality references throughout the paper will give the effort trustworthiness by representing an alertness to prior workings.

Writing a research paper is not an easy job, no matter how trouble-free the actual research or concept. Practice, excellent preparation, and controlled record-keeping are the only means to make straightforward progression.

General style:

Specific editorial column necessities for compliance of a manuscript will always take over from directions in these general guidelines.

To make a paper clear: Adhere to recommended page limits.

Mistakes to avoid:

- Insertion of a title at the foot of a page with subsequent text on the next page.
- Separating a table, chart, or figure—confine each to a single page.
- Submitting a manuscript with pages out of sequence.
- In every section of your document, use standard writing style, including articles ("a" and "the").
- Keep paying attention to the topic of the paper.



- Use paragraphs to split each significant point (excluding the abstract).
- Align the primary line of each section.
- Present your points in sound order.
- Use present tense to report well-accepted matters.
- Use past tense to describe specific results.
- Do not use familiar wording; don't address the reviewer directly. Don't use slang or superlatives.
- Avoid use of extra pictures—include only those figures essential to presenting results.

Title page:

Choose a revealing title. It should be short and include the name(s) and address(es) of all authors. It should not have acronyms or abbreviations or exceed two printed lines.

Abstract: This summary should be two hundred words or less. It should clearly and briefly explain the key findings reported in the manuscript and must have precise statistics. It should not have acronyms or abbreviations. It should be logical in itself. Do not cite references at this point.

An abstract is a brief, distinct paragraph summary of finished work or work in development. In a minute or less, a reviewer can be taught the foundation behind the study, common approaches to the problem, relevant results, and significant conclusions or new questions.

Write your summary when your paper is completed because how can you write the summary of anything which is not yet written? Wealth of terminology is very essential in abstract. Use comprehensive sentences, and do not sacrifice readability for brevity; you can maintain it succinctly by phrasing sentences so that they provide more than a lone rationale. The author can at this moment go straight to shortening the outcome. Sum up the study with the subsequent elements in any summary. Try to limit the initial two items to no more than one line each.

Reason for writing the article—theory, overall issue, purpose.

- Fundamental goal.
- To-the-point depiction of the research.
- Consequences, including definite statistics—if the consequences are quantitative in nature, account for this; results of any numerical analysis should be reported. Significant conclusions or questions that emerge from the research.

Approach:

- Single section and succinct.
- An outline of the job done is always written in past tense.
- Concentrate on shortening results—limit background information to a verdict or two.
- Exact spelling, clarity of sentences and phrases, and appropriate reporting of quantities (proper units, important statistics) are just as significant in an abstract as they are anywhere else.

Introduction:

The introduction should "introduce" the manuscript. The reviewer should be presented with sufficient background information to be capable of comprehending and calculating the purpose of your study without having to refer to other works. The basis for the study should be offered. Give the most important references, but avoid making a comprehensive appraisal of the topic. Describe the problem visibly. If the problem is not acknowledged in a logical, reasonable way, the reviewer will give no attention to your results. Speak in common terms about techniques used to explain the problem, if needed, but do not present any particulars about the protocols here.

The following approach can create a valuable beginning:

- Explain the value (significance) of the study.
- Defend the model—why did you employ this particular system or method? What is its compensation? Remark upon its appropriateness from an abstract point of view as well as pointing out sensible reasons for using it.
- Present a justification. State your particular theory(-ies) or aim(s), and describe the logic that led you to choose them.
- Briefly explain the study's tentative purpose and how it meets the declared objectives.



Approach:

Use past tense except for when referring to recognized facts. After all, the manuscript will be submitted after the entire job is done. Sort out your thoughts; manufacture one key point for every section. If you make the four points listed above, you will need at least four paragraphs. Present surrounding information only when it is necessary to support a situation. The reviewer does not desire to read everything you know about a topic. Shape the theory specifically—do not take a broad view.

As always, give awareness to spelling, simplicity, and correctness of sentences and phrases.

Procedures (methods and materials):

This part is supposed to be the easiest to carve if you have good skills. A soundly written procedures segment allows a capable scientist to replicate your results. Present precise information about your supplies. The suppliers and clarity of reagents can be helpful bits of information. Present methods in sequential order, but linked methodologies can be grouped as a segment. Be concise when relating the protocols. Attempt to give the least amount of information that would permit another capable scientist to replicate your outcome, but be cautious that vital information is integrated. The use of subheadings is suggested and ought to be synchronized with the results section.

When a technique is used that has been well-described in another section, mention the specific item describing the way, but draw the basic principle while stating the situation. The purpose is to show all particular resources and broad procedures so that another person may use some or all of the methods in one more study or referee the scientific value of your work. It is not to be a step-by-step report of the whole thing you did, nor is a methods section a set of orders.

Materials:

Materials may be reported in part of a section or else they may be recognized along with your measures.

Methods:

- Report the method and not the particulars of each process that engaged the same methodology.
- Describe the method entirely.
- To be succinct, present methods under headings dedicated to specific dealings or groups of measures.
- Simplify—detail how procedures were completed, not how they were performed on a particular day.
- If well-known procedures were used, account for the procedure by name, possibly with a reference, and that's all.

Approach:

It is embarrassing to use vigorous voice when documenting methods without using first person, which would focus the reviewer's interest on the researcher rather than the job. As a result, when writing up the methods, most authors use third person passive voice.

Use standard style in this and every other part of the paper—avoid familiar lists, and use full sentences.

What to keep away from:

- Resources and methods are not a set of information.
- Skip all descriptive information and surroundings—save it for the argument.
- Leave out information that is immaterial to a third party.

Results:

The principle of a results segment is to present and demonstrate your conclusion. Create this part as entirely objective details of the outcome, and save all understanding for the discussion.

The page length of this segment is set by the sum and types of data to be reported. Use statistics and tables, if suitable, to present consequences most efficiently.

You must clearly differentiate material which would usually be incorporated in a study editorial from any unprocessed data or additional appendix matter that would not be available. In fact, such matters should not be submitted at all except if requested by the instructor.



Content:

- Sum up your conclusions in text and demonstrate them, if suitable, with figures and tables.
- In the manuscript, explain each of your consequences, and point the reader to remarks that are most appropriate.
- Present a background, such as by describing the question that was addressed by creation of an exacting study.
- Explain results of control experiments and give remarks that are not accessible in a prescribed figure or table, if appropriate.
- Examine your data, then prepare the analyzed (transformed) data in the form of a figure (graph), table, or manuscript.

What to stay away from:

- Do not discuss or infer your outcome, report surrounding information, or try to explain anything.
- Do not include raw data or intermediate calculations in a research manuscript.
- Do not present similar data more than once.
- A manuscript should complement any figures or tables, not duplicate information.
- Never confuse figures with tables—there is a difference.

Approach:

As always, use past tense when you submit your results, and put the whole thing in a reasonable order.

Put figures and tables, appropriately numbered, in order at the end of the report.

If you desire, you may place your figures and tables properly within the text of your results section.

Figures and tables:

If you put figures and tables at the end of some details, make certain that they are visibly distinguished from any attached appendix materials, such as raw facts. Whatever the position, each table must be titled, numbered one after the other, and include a heading. All figures and tables must be divided from the text.

Discussion:

The discussion is expected to be the trickiest segment to write. A lot of papers submitted to the journal are discarded based on problems with the discussion. There is no rule for how long an argument should be.

Position your understanding of the outcome visibly to lead the reviewer through your conclusions, and then finish the paper with a summing up of the implications of the study. The purpose here is to offer an understanding of your results and support all of your conclusions, using facts from your research and generally accepted information, if suitable. The implication of results should be fully described.

Infer your data in the conversation in suitable depth. This means that when you clarify an observable fact, you must explain mechanisms that may account for the observation. If your results vary from your prospect, make clear why that may have happened. If your results agree, then explain the theory that the proof supported. It is never suitable to just state that the data approved the prospect, and let it drop at that. Make a decision as to whether each premise is supported or discarded or if you cannot make a conclusion with assurance. Do not just dismiss a study or part of a study as "uncertain."

Research papers are not acknowledged if the work is imperfect. Draw what conclusions you can based upon the results that you have, and take care of the study as a finished work.

- You may propose future guidelines, such as how an experiment might be personalized to accomplish a new idea.
- Give details of all of your remarks as much as possible, focusing on mechanisms.
- Make a decision as to whether the tentative design sufficiently addressed the theory and whether or not it was correctly restricted. Try to present substitute explanations if they are sensible alternatives.
- One piece of research will not counter an overall question, so maintain the large picture in mind. Where do you go next? The best studies unlock new avenues of study. What questions remain?
- Recommendations for detailed papers will offer supplementary suggestions.



Approach:

When you refer to information, differentiate data generated by your own studies from other available information. Present work done by specific persons (including you) in past tense.

Describe generally acknowledged facts and main beliefs in present tense.

THE ADMINISTRATION RULES

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CRITERION FOR GRADING A RESEARCH PAPER (COMPILATION)
BY GLOBAL JOURNALS

Please note that following table is only a Grading of "Paper Compilation" and not on "Performed/Stated Research" whose grading solely depends on Individual Assigned Peer Reviewer and Editorial Board Member. These can be available only on request and after decision of Paper. This report will be the property of Global Journals.

Topics	Grades		
	A-B	C-D	E-F
<i>Abstract</i>	Clear and concise with appropriate content, Correct format. 200 words or below	Unclear summary and no specific data, Incorrect form Above 200 words	No specific data with ambiguous information Above 250 words
<i>Introduction</i>	Containing all background details with clear goal and appropriate details, flow specification, no grammar and spelling mistake, well organized sentence and paragraph, reference cited	Unclear and confusing data, appropriate format, grammar and spelling errors with unorganized matter	Out of place depth and content, hazy format
<i>Methods and Procedures</i>	Clear and to the point with well arranged paragraph, precision and accuracy of facts and figures, well organized subheads	Difficult to comprehend with embarrassed text, too much explanation but completed	Incorrect and unorganized structure with hazy meaning
<i>Result</i>	Well organized, Clear and specific, Correct units with precision, correct data, well structuring of paragraph, no grammar and spelling mistake	Complete and embarrassed text, difficult to comprehend	Irregular format with wrong facts and figures
<i>Discussion</i>	Well organized, meaningful specification, sound conclusion, logical and concise explanation, highly structured paragraph reference cited	Wordy, unclear conclusion, spurious	Conclusion is not cited, unorganized, difficult to comprehend
<i>References</i>	Complete and correct format, well organized	Beside the point, Incomplete	Wrong format and structuring





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