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Charting Atomic Characteristics: A Pathway to Element Classification By Ousmane BARRY, Mamadou Yaya BALDE, Souleymane BALDE, Lamine KABA & Aboubacar Safie SYLLA

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Abstract- The classification of elements serves as a rich source of knowledge in chemistry. This article explores the atomic model based on wave mechanics, represented by a graph characterizing the atom, to elucidate both periodic and ordered classifications of elements. Three types of classifications emerge from this graph: periodic, ordered, and hybrid classifications, with the latter resembling periodic tables but incorporating elements of the ordered form. The periodic table stands out as one of the most profound and unifying concepts in modern science. New illustration methods, such as condensed tables with orders and periods, are introduced. The results underscore the conclusiveness of the findings, revealing that classifications extend beyond periodicity, encompassing ordered types of tables as well. This research sheds light on the diverse approaches to classifying elements and opens avenues for further exploration in the field of chemistry.

Keywords: periodic table, chemical element, period, affine equation, quantum number, order, wave mechanics, classification, hybrid classification, electronic structure.

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Charting Atomic Characteristics: A Pathway to Element Classification

Ousmane BARRY ^a, Mamadou Yaya BALDE ^c, Souleymane BALDE ^e, Lamine KABA ^w & Aboubacar Safie SYLLA[¥]

Abstract- The classification of elements serves as a rich source of knowledge in chemistry. This article explores the atomic model based on wave mechanics, represented by a graph characterizing the atom, to elucidate both periodic and ordered classifications of elements. Three types of classifications emerge from this graph: periodic, ordered, and hybrid classifications, with the latter resembling periodic tables but incorporating elements of the ordered form. The periodic table stands out as one of the most profound and unifying concepts in modern science. New illustration methods, such as condensed tables with orders and periods, are introduced. The results underscore the conclusiveness of the findings, revealing that classifications extend beyond periodicity, encompassing ordered types of tables as well. This research sheds light on the diverse approaches to classifying elements and opens avenues for further exploration in the field of chemistry.

Keywords: periodic table, chemical element, period, affine equation, quantum number, order, wave mechanics, classification, hybrid classification, electronic structure.

I. INTRODUCTION

n any chemistry class or laboratory, there is almost always a periodic table hanging on the wall. This table, which includes all the known elements, provides a lot of information on each. As one progresses in the study of chemistry, the usefulness of the periodic table becomes more and more obvious[1].

Atoms were first suggested by the Greek philosophers Democritus and Leucippus around 400 BC, the concept was primarily based on a hunch. In fact, for several centuries thereafter, no convincing experimental evidence was available to support the existence of atoms. The first real scientific data were collected by Lavoisier and others from quantitative measurements of chemical reactions. Thus around 1790 the scientist was already able to identify 33 chemical elements. The results of these stoichiometric experiments allowed John Dalton to propose the first systematic atomic theory. This one, although crude, has stood the test of time extremely well. Once atoms were admitted, logically a number of questions arose such as: What is the nature of an atom? How is it composed? What are the constituent parts, their properties etc.?

The most striking phenomenon in chemistry was very early the discovery of an analogy and the periodic repetition of the properties of the elements with each other. Several groups of elements with great similarities in their chemical behavior were identified. This is how the need to classify the elements arose. During the first 30 years of the 20th century, the appearance of a new theory, quantum mechanics, made it possible to explain the behavior of atoms under the effect of light. This so-called modern theory made it possible to develop the electronic structure of atoms. This confirmed the periodic behavior of chemical elements in terms of electronic configuration and then the properties of atoms.

In 1817 Döbereiner succeeded in relating the atomic mass of certain elements to their properties. He noticed the existence of similarities between elements grouped in threes which he called "triads". He highlighted the fact that the mass of one of the three elements of the triad was the intermediate (the average) of the other two. In 1850, we could count more than twenty triads to arrive at a first coherent classification.

In 1862 Chancourtois, a French geologist, highlighted a certain periodicity between the properties of the elements of the table. Later the English chemist Newlands announced that "the eighth element which follows a given element resembles the first as the eighth note of the octave resembles the first". But this law could not apply to elements beyond calcium: "The notions of transition elements were unknown". This classification therefore remained insufficient, but the periodic table was beginning to take shape.

In 1869 and 1870 respectively, Dmitri Mendeleev and Lothar Meyer claimed that the properties of elements could be represented as periodic functions of their atomic weight and presented their ideas in the form of a periodic table. As new elements have been discovered, the original form of the periodic table has been significantly modified and it is now recognized that periodicity is a consequence of variation in ground state electronic configurations. A modern periodic table emphasizes blocks of 2, 6, 10, and 14 elements that result from filling the s, p, d, and f atomic orbitals, respectively. An exception is He, which for chemical reasons is placed in the rare group [2].

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The most famous of the periodic classifications was that of MENDELEIEV in 1869. It is to him that the merit goes to having presented the first, a wellstructured and coherent classification of all the elements known at his time. He then realized the periodic trend in the properties of these placed in increasing order of their atomic masses [3]. This table had 6 columns with short periods with empty boxes according to his manuscript. This intuition of the scientist was confirmed by the later discovery of these new tenants corresponding to his prediction. More than a century after the death of the main founder of the periodic system, different types of periodic tables have emerged. It is time to revisit the origins, the precursors and even the status of this classification which has had a profound influence on the development of chemistry and modern physics. Scientists today are still debating the best possible presentation of the periodic table [4].

The American chemist Glenn T. Seaborg, Lawrence Berkeley National Laboratory, Berkely, California, United States of America whose only address is composed of five chemical elements (Sg, Lr, Bk, Cf, and Am respectively seaborgium, lawrencium, berkelium, californium and americium) had the merit of completing MENDELEF's periodic table of elements with the addition of Actinides in 1945 [5]. From Lavoisier in 1790 to Glenn T. Seaborg, nearly two centuries of scientific exchange and research have led to several types of classifications, all periodic. It only differs in the mode of presentation.

In 1990 FERNANDO Dufour proposed a threedimensional classification which he called "element tree" or "tree of elements" also called "periodic tree".

In 1995 PIERRE Demers of the University of Montreal proposed a classification system called "Québécium", a pyramid with 4 faces and 4 levels; the upper level made up only of 2 elements of the "s" block (1st period); the next is composed of the elements of the "s and p" blocks including 4 of "s" and 12 of "p" (the 2nd and 3rd periods), the penultimate level includes the elements of the "s, p and d" blocks » including respectively 4, 12 and 20 elements (periods 4 and 5) and the last level the base containing the elements of the four blocks "s, p, d and f" respectively 4, 12, 20 and 28 components of periods 6 and 7. The atomic number values Z provide an undeniable if not sufficient ordering principle. A classification called modern, somewhat analogous to the Quebec system but more explicit, is also presented as a pyramid of 7 levels, each corresponding to a period. From the top "the 1st period" to the base "the 7th period" of the pyramid, lines evolve from top to bottom materializing the identity of the valence electrons, that is to say the families (columns) of the periodic table of elements [6].

Eric SCERRI, for the sake of greater regularity, proposes a table in which the hydrogen and helium of the 1st period are brought back to the 2nd period respectively at the level of halogens and rare gases with the columns np^5 and np^6 of block p cut and moved to the right, this leads to a disrupted periodic table of elements, with a regularity of the type 8 8 18 18 32 32 2 as opposed to 2 8 8 18 18 32 32 [7].

All these classifications present a certain discontinuity from one period to the next according to certain researchers, this seems to mark a disagreement with the logical sequence of the usual ordination of natural numbers in which the atomic numbers Z of the atoms would be found (Z ranging from 1118).

This reason indicated above (mentioned) aroused concern and caught the attention of other researchers and served as a reference for them to promote new types of classification. Thus were born the spiral classifications. Professors PHILIP Steaward and JAN Scholten each proposed a spiral classification model, periodic spiral systems or "periodic tables" which according to them respect the natural ordination of atomic numbers. Théodor BENFEY's snail system from the 1960s and others are also of the spiral model. It should be noted that the difference between these presentation models is the simple notion of parallel and spiral linear chains. Periodicity is the only criterion for all these classifications. The period is an instantaneous phenomenon, it is the time that a phenomenon takes to complete the phases of its duration, it is the time interval at the end of which a phenomenon reproduces under the same conditions, such as the properties of the elements chemical; In chemistry periodicity is examined in terms of the number of elements with which identical properties recur from one group to another. It is this phenomenon that the English chemist Newlands wanted to explain with his famous quote: "the eighth element which follows a given element resembles the first as the eighth note of the octave resembles the first".

In 1989 Ouahès, R. made a proposal for a new periodic table of elements. This other table is still called periodic of 8 periods: He explains his table in these terms: "This means, in terms of atomic structure, that each period corresponds to the filling of shells and subshells according to n + l = k (*k*: constant) where *n* and *l*are the principal and azimuthal quantum numbers. The constant *k* is the rank of the period. The results are given in table 1. The number of periods is 8 instead of 7."

This article will answer questions such as:

- Can we explain the periodic classification using a more practical vision than that, entirely empirical, of Mendeleev?
- How does the distribution of electrons in an atom help explain periodic properties?
- Is this other table periodic?
- Is the number of periods greater than 7?

In general, the electronic structure gives all the information about an element, mainly its coordinates in

the different tables. Its physical and chemical properties also depend on it. Valence electrons provide the main concepts of chemical language. They open the way to forms of classification of elements, hence the different methods of illustrating the classification [8].Obtaining specific graphs from the characteristic graph of the atom was explained in the methodology section. The types of classification of chemical elements were illustrated and discussed and a conclusion was drawn.

II. Methodology

The specific graphs describing the orders and periods concretely explain the principles of stability,

• Electronic structure according to order (y-axis)

aufbau, and in general the Kletchkovky rule [9]. The distribution of the electrons of the different atoms along the axes of these graphs (abscissa and ordinate) by the periods and the orders respectively make it possible to explain each classification model in the literature and even to propose new methods of illustrating the classifications of the elements. A new criterion for classification of chemical elements becomes the order of the energy level " O_E ", absolutely different from the period "*n*", has even emerged, it leads to a new classification called the ordered classification of the elements [10].



Graphe 1: Electronic structure of elements according to order

• Electronic structure along the period axis



Graphe 2: Electronic structure of elements according to period

The ordered tables are obtained according to graph 1. The classification is made according to the ordinate axis " O_E " which becomes the new classification criterion leading to tables of a new type called ordered tables, the set of ordered tables is summarized by a socalled condensed order table. This goes without saying all periodic classifications, the only classic for

classification criterion of which is the period "n" according to graph 2 and are all also summarized by a condensed period table. These two condensed tables are analogous but different, the energy order "O_F" is different from the period "n". See below the respective condensed forms which are effectively new methods of illustrating the classifications.

Order	1	2	3	4	5	6	7	8
Underlays	1s	2s	2p3s	3p4s	3d4p5s	4d5p6s	4f5d6p7s	5f6d7p
Z	1-2	3- 4	5- 12	13- 20	21- 38	39- 56	57- 88	89-118
Type of tables	2	2	8	8	18	18	32	30

Table 2: Condensed period table

Table 1: Condensed table	to	order
--------------------------	----	-------

				•			
Periods	1	2	3	4	5	6	7
Underlays	1s	2s2p	3s3p	4s3d4p	5s4d5p	6s4f5d6p	7s5f6d7p
Ζ	1-2	3- 10	11- 18	19- 36	37- 54	55-86	87-118.
Type of tables	2	8	8	18	18	32	32

III. RESULTS

To make the classification, whether periodic or ordered, it is enough to arrange vertically the stacks of the different sub-layers in each case. Give each block the number of columns corresponding to its electronic capacity: 2 for block "s", 6 for "p", 10 for "d" and 14 for "f". The classifications are either short, medium and long depending on the stacks chosen as a basis. The main stacks are "sp / ps" for tables with 8 columns "sdp /dps" for those with 18 columns and finally "sfdp/fdps" for long classifications with 32 columns.

The vertical arrangements of the sublayers by stacking the periods and/or order offer the appearance of tables: the periodic tables are fairly well-known emblematic figures which adorn quite a few scientific circles today. Ordered tables constitute a completely new classification model and appear in the form of stairs with regular steps. The longest ordered arrays are made up of four steps corresponding to the four blocks of elements. These stairs start from block "f" via those of "d, p" and end with that of "s", from bottom to top respectively.

	a)	Perioc	lic			b) Or	dered			O
n		_						1s		1
1	1s							2s		2
2	2s			2р			2р	3s		3
3	3s			3р			3р	4s		4
4	4s		3d	4p		3d	4р	5s		5
5	5s		4d	5р		4d	5р	6s		6
6	6s	4f	5d	6р	4f	5d	6р	7s		7
7	7s	5f	6d	7р	5f	6d	7р		-	8

Comparing these two classifications, it appears that the order is obtained by the transfer of the "s" block from the left of Mendeleev's periodic table to its right with a shift of one level upwards; which allows you to go from 7 periods to 8 orders. Orders always end with "s" while periods begin with "s" and are composed respectively as follows: (n-3) f (n-2) d (n-1) p ns and ns (n-2) f (n-1) d np.

The ordered classification puts the transitional elements, blocks f and d, of group "B" before the nontransitional elements, blocks s and p, of group "A". This leads to a more regular, coherent presentation, compared to the periodic system which inserts transition elements between normal and non-normal ones. In the periodic system the elements "d" start with IIIB progress to VIIIB then follow groups IB and IIB marking the end of the "d" block. The ordered system puts the normal elements "s and p" of the large group "A" in the same arrangement as those of the transition. The "p" elements start with IIIA progress to VIIIA and follow the IA and IIA of the "s" block.

In an ordered array the transitional elements are at the beginning and the non-transitional ones at the end. This can justify the reason which divides the four blocks of elements into two large groups A (normal elements) and B (transition elements). By the amplitude of the stacks or the types of tables of orders and/or periods, the methods of illustrating the classifications are summarized by series with values corresponding to the number of elements by order " O_E " and/or by period « n »:

Order series « O_E » : 2 2 8 8 18 18 32 30
Series with period « n » : 2 8 8 18 18 32 32

IV. Discussion

Knowing that for the block "s" the order is identical to the period and for the three other blocks "p, d, f" the period is less than the order of one unit, it is possible to go from one classification ordered to a periodic system. This is possible by tilting the "s" block of the ordered system one level down to obtain a periodic system like the ordered one, thus we go from 8 orders to 7 periods corresponding to the periodic and ordered system. This hybrid classification conveys the characteristics of both types of classifications.



The transformation of the ordered system into the periodic classification in its image is a stepped but periodic system. It has all the form of the ordered table but it is also periodic. It also comes in 8, 18, and 32 columns as in the case of Mendeleev's periodic table.

This is how its 18-column classification would look like:

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
GRO	ROUPE B										UPE /	4					
III	IV	V	VI	VII	VIII			I	П	Ш	IV	V	VI	VII	VIII		
3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	1	2
										(Bloc	Ks "s	& p")			2	1	2
		(Bloc	Ks "f	& d)						5	6	7	8	9	10	3	4
										13	14	15	16	17	18	11	12
21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	19	20
39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	37	38
57*	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	55	56
89*	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	87	88

*	58	59	60	61	62	63	64	65	66	67	68	69	70	71
*	90	91	92	93	94	95	96	97	98	99	100	101	102	103

Graphs 1 and 2 depict the composition of the electron cloud elements in order and period respectively. The arrangement of the sublayers from 1s to 7p is practically done according to the increasing order of the energy level either by the ordinate " O_E " or by the abscissa "n" in the usual ordination of natural numbers.

The structure of an order according to the accepted values is (n-3) f(n-2) d(n-1) p ns and that of the period is ns (n-2) f (n-1) d np, it is remarkable that the order ends with the sublayer "ns" while it can start with any sublayer even with ns if the order is short as for the cases: n = 1 and 2. Practically the orders go from 1 to 8. For the periods, they start with "ns" and end with "np", the coefficients of the intermediate sublayers are counted from "n" to the limit values. The order is then different from the period, the periods go from 1 to 7. The coefficients of the sublayers in each case are linked to O_E and/or n, knowing that for the block "s" $O_E = n$. Thus, it is enough to be able to count from 1 to 7 and/or from 1 to 8 to develop the rule of stability.

Table 3: Group of sublayers according to the order of energy (Graph 1)

Orders « O _E »	1	2	3	4	5	6	7	8
Underlays	1s	2s	2p 3s	3p 4s	3d 4p 5s	4d 5p 6s	4f 5d 6p 7s	5f 6d 7p

Ondenays	10	23	2p 03	0 p 4 3	00 - p 03	-4 0p 03	+100 0p 73	01007p
	7	Table 4	: Group of	[:] sub-laye	rs according t	to the period ((Graph 2)	
				,	0			

Periods « n »	1	2	3	4	5	6	7
Underlays	1s	2s 2p	3s 3p	4s 3d 4p	5s 4d 5p	6s 4f 5d 6p	7s 5f 6d 7p

These two tables sufficiently express the difference between an "O_E" order and a "n" period, but these two criteria are well linked.

Table 5: Relationship between period and order (Graphs 1 & 2)

Periods « n »	1		2		3		4		5		6		7
Underlays	1s	2s	2р	Зs	Зр	4s	3d4p	5s	4d5p	6s	4f5d6p	7s	5f6d7p
Orders « O _E »	1	2	e.	3	4		5		6		7		8

These two very similar tables are different. For block "s", $O_F = n$, the order is identical to the period and for the others the orders are greater than the period by one unit. The condensed period form confirms all current periodic classifications of the elements. It is the summary of all the classic period processes developed according to the abscissa axis "n" whatever the two or three dimensional model, spiral or not.

The condensed ordered table is also another classification process leading to entirely new forms of tables which are far from being periodic. They are ordered or ordered tables of the elements, obtained along the "O_F" ordinate axis. These resulting paintings are presented in the form of a staircase also with 8, 18 and 32 columns. If for the periodic system the classification criterion is the period "n", the ordered tables are carried out around a new classification criterion, namely the order "O_E". Except that they are not periodic. The word "period" or its adjective "periodic" was not accidentally used by the founding father of the classification system. This choice expressed the behavior of the properties of the groups of elements which were born, evolved then diminished and canceled after a certain number of elements which make up said period. An order is a combination of the elements of two successive periods.

CONCLUSION

This new classification conforms to the electronic structures of the elements according to order and period. It can have historical and educational interest, because it is based on simple to understand concepts.

This work demonstrated that all classifications in the literature are not only periodic. The characteristic graph of the atom made it possible to confirm the existence of the two classification criteria leading to fairly classic periodic tables built around the period "n going from 1 to 7" and completely new ordered or ordered tables which carry out around the order of the increasing level of energy order of the sublayers "O_F starting from 1 to 8" following the ordinate of the graph. The ordered table corresponds to the transfer of the "s" block from the left of the periodic table of elements to its right with the tilting of the latter one level upwards from 7 stacks to 8 others.A third classification intermediate to the other two was envisaged by the reduction of the block "s" of the ordered form by one level downwards from 8 stacks to 7 others, hence a periodic classification in the image of the ordered form. A new illustration of the classification of elements.

We hope to soon provide a rigorous proof of atomistic terminology using the equation of the line, such as the deduction of valence electrons, the electronic transition and the relationship between the nucleus and the electron cloud of an atom.

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Screening of Therapeutic Potential and Compounds of Endemic *Nepeta pilinux* P.H. Davis in Kew Bull. from Şanlıurfa

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Abstract- Nepeta pilinux P.H. Davis in Kew Bull. was recently recorded as an endemic species in Birecik, Şanlıurfa. The essential oils were obtained from air-dried aerial parts by hydrodistillation and their composition was investigated using GC-FID and GC/MS. Determination of antioxidant capacity, and urease and xanthine oxidase inhibitions of the methanolic extracts were performed with HPLC-DAD and spectrophotometry. 34 compounds were identified constituting 89.2% of the total essential oil compounds. The major components were determined as T-cadinol (31.1%), γ -muurolene (14.4%) and 14-nor-cadin-5-en-4-one isomer A (11.0%) in the oil. Mainly rosmarinic acid, chlorogenic acid, and caffeic acid derivatives were quantified together with apigenin, luteolin and tangeretin derivatives in the extracts by HPLC-DAD. The total phenolics of the extract from leaf and flower parts, 50.81 mg GAE.g-1, was higher than the extract from stem part, and the radical scavenging activity of this extract was also stronger. While, the leaf and flower extract had significant urease and xanthine oxidase inhibitory activities (62.47 and 48.48 μ g.mL-1), stem extract had low inhibition on both enzymes.

Keywords: nepeta pilinux, essential oil, rosmarinic acid, GC-FID, GC/MS, HPLC-DAD.

GJSFR-B Classification: LCC: QK898.L42

SCREEN I NGOFTHERAPEUTI CPOTENTI A LAN DE OMPOUNDSOFENDEMI EN EPETAPI LI NUXPHDAVI SI NKEWBULLFROMSAN LI URFA

Strictly as per the compliance and regulations of:



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Abstract- Nepeta pilinux P.H. Davis in Kew Bull. was recently recorded as an endemic species in Birecik, Şanlıurfa. The essential oils were obtained from air-dried aerial parts by hydrodistillation and their composition was investigated using GC-FID and GC/MS. Determination of antioxidant capacity, and urease and xanthine oxidase inhibitions of the methanolic performed with HPLC-DAD extracts were and spectrophotometry. 34 compounds were identified constituting 89.2% of the total essential oil compounds. The major components were determined as T-cadinol (31.1%), ymuurolene (14.4%) and 14-nor-cadin-5-en-4-one isomer A (11.0%) in the oil. Mainly rosmarinic acid, chlorogenic acid, and caffeic acid derivatives were quantified together with apigenin, luteolin and tangeretin derivatives in the extracts by HPLC-DAD. The total phenolics of the extract from leaf and flower parts, 50.81 mg GAE.g¹, was higher than the extract from stem part, and the radical scavenging activity of this extract was also stronger. While, the leaf and flower extract had significant urease and xanthine oxidase inhibitory activities (62.47 and 48.48 μ g.mL⁻¹), stem extract had low inhibition on both enzymes.

Keywords: nepeta pilinux, essential oil, rosmarinic acid, GC-FID, GC/MS, HPLC-DAD.

I. INTRODUCTION

he genus *Nepeta* is distributed over a large part of Central and Southern Europe, and West, Central, and Southern Asia as a multi-regional genus of the Lamiaceae (labiatae or mint) family, consisting of approximately 300 taxa (1, 2). 34 species (40 taxa) of the genus Nepeta were recorded in the Flora of Turkey (3, 4). According to the revision of this genus, it is represented by 39 species (46 taxa) in Turkey with recent studies. Endemism rate on the basis of this species is 44% (5). *Nepeta* species are usually named as catmint or catnip due to the sedative effects on cats and they are commonly used as diuretic, spasmolytic, diaphoretic, bronchodilator, antitussive, anti-asthmatic and sedative agents in Turkey. Due to their antiseptic properties, they are used topically in the treatment of children with skin rashes, and in snake and scorpion bites as well. Nepeta caes area, an endemic species in Turkey, has folkloric uses in southern Anatolia and is used as a herbal tea to treat gastric disorders (6,7). Some Nepeta species have been known their feline attractant activity since they have nepetalactone and its derivatives which are responsible for attractant properties (1,7). Nepetalactones have been reported to effect on insects compare to DEET (N, N-diethyl-mtoluamide) (8-10). Nepetoideae is essential oil-rich genera of the Lamiaceae, therefore has potential economic interest (11).

According to the various studies on the essential oil composition of Nepeta species the essential oil composition depends on the species, place of cultivation, climatic conditions and method of analysis (12-26). The most comprehensive study on 22 Nepeta species was performed by Baser et al. (2000). They were classified into two groups according to of these species; composition of essential oil nepetalactone-containing and nepetalactone-less. Nepetalactone-containing species have $4a\alpha$ -7 α -7a α nepetalactone as the most frequently contained nepetalactone. Nepeta cadmea Boiss., Nepeta cataria L., Nepeta caesarea Boiss. and Nepeta pilinux P.H. Davis contained $4a\alpha$ - 7α - $7a\alpha$ nepetalactone while Nepeta racemosa Lam. contained $4a\alpha$ - 7α - $7a\beta$ -nepetalactone as major compound in their oils. Caryophyllene oxide or 1.8-cineole/linalool were identified in the essential oils of Nepetalactone-less species as the major components. As main compounds, β -pinene. a-terpineol, germacrene-D, and spathulenol were respectively determined in the oil of Nepeta phyllochlamys P.H. Davis, Nepeta viscida Boiss., Nepeta sorgerae Hedge et Lamond, and Nepeta trachonitica Post which are out of the two groups (1).

Being a type of the most popular antioxidant secondary metabolites, phenolic compounds were investigated in the *Nepeta* species as well. Rosmarinic

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acid, epicatechin, chlorogenic acid, caffeic acids, quercetin, rutin, ellagic acid, thymusin, luteolin, and apigenin which are well-known as antioxidant compounds were found in the extracts of *Nepeta cadmea* Boiss., *Nepeta nuda* subsp. *albiflora, Nepeta asterotricha* Rech., *Nepeta rtanjensis* Diklic & Milojević (27-30).

Ureases, a nickel-dependent metalloenzymes, are synthesized by plants, some bacteria, algae and fungi. The jack bean urease (urea amidohydrolase EC 3.3.1.5) catalyzes the hydrolysis of urea to form ammonia and carbon dioxide (31). *Helicobacter pylori*, a gram-negative microaerophilic pathogen survivable in a limited pH (4.0–8.2) range. This pathogen can successfully colonize and persist in the mucous layer of the human stomach since its urease activity which produces ammonia to reduce stomach acidity. Since antibiotic-resistant strains of H. pylori can emerge against antibiotics, it is believed that plant-derived urease inhibitors would be more beneficial against gastroduodenal disease associated with this pathogen (32, 33).

Xanthine oxidase (EC 1.2.3.2.) produces hydrogen peroxide and superoxide anion, which are reactive oxygen species (ROS) during the oxidation of hypoxanthine and xanthine to uric acid. Under physiological conditions, ROS is kept at a low level by the antioxidant system. Disruption of the balance of ROS and antioxidants due to some diseases causes tissue and DNA damage due to the increase in metabolism of ROS. Inhibition of xanthine oxidase reduces the amount of uric acid and ROS in the bloodstream to prevent both hyperuricemia and oxidation stress (34). Inhibitory effects of some flavonoids such as diosmetin, luteolin, chrysoeriol, apigenin, kaempferol on xanthine oxidase have been reported in various in vitro studies (35-37).

Nepeta pilinux P. H. Davis in Kew Bull. is an endemic species of Nepeta genus growing in the Southwestern Anatolia (Antalya: Alanya) (3). However, in recent flora research, Nepeta pilinux was encountered in Şanlıurfa, Birecik district and was recorded as new endemic species for Şanlıurfa flora. Nepeta pilinux is named as 'top pisik otu'. In Şanlıurfa, the fresh aerial parts of the plant are used to heal mouth sores (38).

There is no knowledge about the essential oil and phenolic composition of *Nepeta pilinux* endemic species from Şanlıurfa. Chemical compositions of the polar and apolar extracts from *Nepeta pilinux* were determined by HPLC-DAD and GC/MS respectively for the first time in this study. Radical scavenging activities against DPPH and ABTS radicals and enzyme inhibition activities on urease and xanthine oxidase were investigated in vitro to elucidate the bioactivities which may have developed by this species depending on its chemical composition.

II. Experimental

a) Materials

Nepeta pilinux was collected from Sanliurfa: Birecik, Kelaynak area in Turkey in 5 May 2018. The voucher specimen has been deposited at the Herbarium in the Recep Tayyip Erdoğan University (RTEUB 6079), Rize, Turkey (Voucher specimen no: FABAK 1702). The plant material was identified by Prof. Dr. Vagif ATAMOV (Recep Tayyip Erdogan University, Faculty of Science and Literature, Department of Biology, Rize, Turkey). All standards of phenolic compounds were obtained from Sigma-Aldrich (St. Louis, MO, USA) but guercetin from Fluka Chemie GmbH (Switzerland). Na₂CO₃ and K₂S₂O₈ were provided from Sigma-Aldrich (St. Louis, MO, USA. HPLC grade acetonitrile, methanol, acetic acid, Folin ciocalteau, 2,2-Diphenyl-1-picrylhydrazyl (DPPH), and 2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) diammonium salt (ABTS) were supplied by Merck (Darmstadt, Germany).

b) Isolation of the Essential Oil

Aerial parts of the plant were water distilled for 3 h using a Clevenger-type apparatus. The essential oil was stored at 4°C in the dark until analyzed.

c) Extraction of Phenolic Compounds

The aerial parts of the plant were divided into two parts. The leaves and flowers were combined in one part and the stems were separated into the other part. The part consisting of leaves and flowers was called NP-LF for short, and the part consisting of stem is called NP-S. These parts were finely ground and 0.5 grams of each were defatted by using 10 mL of hexane. The plant residues were dried at 40 °C for 30 min after removing the hexane extract. 20 mL methanol were added to these residues for the extraction of phenolic compounds. Extraction was continued overnight at 37 °C in the dry thermo-shaker cabinet at 352 rpm, then the extracts were centrifuged at 5000 rpm and supernatants transferred into the falcon tube. This procedure was repeated by adding 10 mL of methanol in the residue. All extract was concentrated until 5 mL by using rotary evaporator at 35 °C. 500 and 1500 μ L of these extracts were stored at -20 °C for HPLC-DAD analysis, antioxidant and enzyme inhibition tests. Remaining 3 mL of extract was evaporated to calculate the concentration of the extracts. NP-LF and NP-S extracts concentration were 23.17 and 23.50 mg.mL⁻¹ respectively.

d) Gas Chromatography (GC) and Gas Chromatography-Mass Spectrometry (GC/MS) analysis of essential oil

The oil was analyzed by capillary GC and GC/MS using an Agilent GC-MSD system.

i. GC/MS analysis Conditions of Essential Oils

The oil was analyzed by capillary GC/MS using an Agilent GC-MSD system (Agilent Technologies Inc., Santa Clara, CA). HP-Innowax FSC column (Hewlett-Packard-HP, U.S.A.) (60 m × 0.25 mm i.d., with 0.25 μ m film thickness) was used for separation of components in the oil and helium as a carrier gas (0.8 mL/min). The GC oven temperature was kept at 60°C for 10 min and programmed to 220°C at a rate of 4°C/min, and kept constant at 220°C for 10 min and then programmed to 240°C at a rate of 1°C/min. The split flow was adjusted at 40 mL min⁻¹ with 40:1 split ratio. The injector temperature was set at 250 °C. Mass spectra were taken at 70 eV with the mass range *m/z* 35-450.

ii. GC Analysis Conditions of Essential Oils

The GC analysis was done with Agilent 6890N GC system fitted with a FID detector set at a temperature of 300°C. To obtain the same elution order with GC/MS, simultaneous auto-injection was done on a duplicate of the same column applying the same operational conditions. Relative percentage amounts of the separated compounds were calculated from FID chromatogram.

iii. Identification of Essential Oils

Identification of essential oil components were performed by comparison of their mass spectra with those in the Baser Library of Essential Oil Constituents, Wiley GC/MS Library, Adams Library, MassFinder Library (39-41) and confirmed by comparison of their retention indices. A homologous series of *n*-alkanes were used as the reference points in calculation of relative retention indices (RRI) (42). The relative percentages of the separated compounds were calculated from FID chromatograms. The analysis results are expressed as mean percentage as listed in Table 1.

e) HPLC-DAD Analysis Conditions of Methanol Extracts

The chromatographic analyses were performed using a Dionex (Thermo scientific, Germering, Germany) Ultimate 3000 high performance liquid chromatography (HPLC) system equipped with an Ultimate 3000 diode array detector (DAD).

A Thermo acclaim C30 column (150mm. 3mm id. 3µm pd) was used with Macherey Nagel (3mm id) guard column. Gradient elution was used with mobil phases; A: 2% acetic acid in water and B: 70% acetonitrile-30% water. Flow rate was 0.37 mL/min and injection volume was 10 µL. Column temperature was 25°C. Following 24 phenolic standards were used to calibration and validation of HPLC-DAD analysis method: Gallic acid, protocatechuic acid, p-hydroxy benzoic acid (p-OH benzoic acid), vanillic acid, catechin, chlorogenic acid, caffeic acid, syringic acid, vanillin, epigallocatechin gallate (EGCG), epicatechin, pcoumaric acid, ferulic acid, chicoric acid, rutin, luteolin-7-glycoside, hesperidin, apigenin-7-glycoside, rosmarinic acid, luteolin, quercetin, hesperetin, apigenin, and tangeretin. They were diluted from their stock

solution into nine different concentration at 0.3125; 0.625; 1.25; 5.0; 10.0; 25.0; 40.0 mg.L⁻¹ in 1:1 methanolwater solution. External calibration method was used and their regression coefficient were found at least 0.999. Repeatability of the retention time and peak areas were measured as coefficient of variation (CV) which was under 0.93 for retention times and 6.02 for areas of the peaks. Limit of detection and quantification values of the peaks were under 0.11 and 0.37 μ g.mL⁻¹ for all standards. Chromatograms were processed at 254, 280, 315, and 370 nm with DAD which operated 200-400 nm. The identification of the peaks was carried out by comparing the retention times and UV spectra with those of standard phenolic compounds. Some peaks had the same or very similar UV spectra as some standards, but with different retention times. They were defined as derivatives of standards with similar UV spectrum and quantified as equivalent of those standards.

f) Determination of Total Phenolic Content

Total phenolic content was determined by using the yellow colored Folin–Ciocalteu's phenol reagent, which was reduced to its blue complex in the presence of reducing agent such as phenolic compounds (43). Gallic acid and quercetin were used as phenolic standards to generate standard curves in a range of 0.0156 and 0.500 mg/mL at 6 concentration levels ($r^2 =$ 0.998). The optical density of the extracts with phenol reagent in the alkaline solution was measured at 760 nm with a UV–Vis detector (Thermo Scientific Multiskan Go, USA). The results were expressed in mg of gallic acid (GAE) and quercetin equivalent (QE) per gram of extracts. All concentration point of the extracts was analyzed in triplicate.

g) DPPH Free-Radical Scavenging Activity Assay

The free-radical scavenging activity was determined based on the reduction of the purple colored 2,2-diphenyl-1-picrylhydrazyl (DPPH•) radical to the yellow colored DPPH-H form by the effect of an antioxidant species such as phenolic compounds in the extracts. It was spectrophometrically performed at 517 nm (44). Briefly, 0.15 mL of plant extract was mixed with 0.15 mL 0.1 mM daily prepared DPPH in methanol and incubated for 30 min in the dark. Gallic acid and quercetin were used as standards to compare with the methanol extracts. Results are reported as SC₅₀ values, demonstrating the concentration of extract (µg extract per mL methanol) necessary to scavenge 50% of DPPH•. All concentration point of the extracts was analyzed in triplicate.

h) ABTS Radical Scavenging Assay

7 mM of ABTS solution and 2.4 mM of potassium persulfate solution were mixed in equal quantities and allowing to oxidation reaction of ABTS by $K_2S_2O_8$ for 18 h at room temperature in the dark to form

the ABTS^{•+} radical. Obtained radical solution was then diluted with methanol 25 times to obtain an ABTS^{•+} solution has optical density of 0.700 \pm 0.01 at 734 nm (45). 50 μ L plant extracts were allowed to react with 250 μ L of the ABTS^{•+} radical solution and the absorbance was measured at 734 nm after 30 min using a spectrophotometer. The ABTS^{•+} scavenging capacity of the extracts were compared with that of gallic acid and quercetin and reported with SC₅₀ values (μ g extract/ mL methanol). All concentration point of the extracts was analyzed in triplicate.

i) Urease Inhibitory Assay

Urease inhibition of the extracts were performed according to the phenol-hypochlorite method developed by Weatherburn (1967) (46). Jack bean urease was used as a model enzyme. Optical density of the resulting blue-navy colored mixture at 625 nm were recorded on a spectrophotometer (1601UV-Shimadzu, Australia). To calculate the IC_{50} values of the polar extracts, different concentrations of the extracts or inhibitory compounds were prepared. Acetohydroxamic acid, well-known inhibitor of urease, were used as positive control.

) In Vitro Anti-Xanthine Oxidase Assay

The inhibition of xanthine oxidase was measured by UV spectroscopy technique at 295 nm which is attributed to the released uric acid from xanthine. The inhibitory activity of the extract was determined using a slight modification of the reference methods (34). Briefly, the reaction mixture consisted of 500 mL of the extract solution, (diluted in DMSO), 770 mL of phosphate buffer (pH 7.8) and 70 mL of bovine milk xanthine oxidase (0.4 U/mL, Sigma Aldrich, St. Louis, USA) was prepared. The reaction was initiated by the addition 660 mL of xanthine solution (0.4 mM) into the mixture after incubation at 25°C for 15 min. The assay mixture was incubated at 25°C for 15 min again. The reaction was stopped by adding 200 mL of 0.5 N HCl and the absorbance was measured at 295 nm using UV/vis spectrophotometer (1601UVShimadzu, Australia). A well-known XO inhibitor (XOI), allopurinol (Sigma Aldrich, St. Louis, USA) was used as a positive control. XO activity was expressed as percent inhibition of xanthine oxidase, calculate as (1-B/A) x 100, where A is the change in absorbance of the assay without the test samples. (Δ abs with enzyme - Δ abs without enzyme),

and B is the change in absorbance of the assay with the test sample (Δ abs with enzyme - Δ abs without enzyme). The assay was done in triplicate. The IC₅₀ value was determined as the concentration of the extract that gave 50% inhibition of maximal activity.

III. Results and Discussion

a) GC And GC/MS Analysis of Essential Oils

Essential oil yield in the sample was calculated as 0.26%. Thirty-four compounds comprising about 89.2% of the essential oil were identified. Identified essential oil components were compared with literature polar column retention times. The major components were determined as T-cadinol (31.1%), γ -muurolene (14.4%) and 14-nor-cadin-5-en-4-one isomer A (11.0%) in the oil. *Nepeta pilinux* essential oil has oxygenated sesquiterpenes (44.6%), sesquiterpenes hydrocarbons (17.3%), oxygenated monoterpenes (10.1%), monoterpene hydrocarbons (2.3%) and others (14.9%). The analysis results were shown in Table 1.

Our results were not similar to the classification of *Nepeta* species reported by Baser et al. (2000) (1). *Nepeta pilinux* which was collected from Antalya, Alanya district was in the group containing nepetalactone, an iridoid monoterpene according to their report. Although 89.2% of the essential oil was determined in *Nepeta pilinux* from Şanlıurfa, nepetalactone was not detected. The determination of different essential oil compositions from *Nepeta pilinux* from different regions may have resulted from the difference in locality.

The major constituent of water-distilled essential oils of Nepeta heliotropifolia and Nepeta congesta subsp. cryptantha was determined by GC/MS and GC-FID and found to be germacrene D (36.7% and 38.5%, respectively). Their main aroma component was determined as eucalyptol (48.0% and 24.7%, respectively) (47). Although the major essential oil and aroma compounds of these two species were quite different from the main essential oils of Nepeta pilinux, the compositions of these three species had lots of common compounds such as α and β -pinene, γ -muurolene, myrtenal, pinocarveol, caryophyllene oxide, cubenol, T and a-cadinol ect. The presence of more or less components may be due to differences between species, as well as environmental conditions and the harvesting time of the plant.

RRIª	RRI⁵	Components	%	IM
1032	1032 ^c	α-Pinene	2.3	t _R , MS
1118	1118 ^c	β-Pinene	tr	t _R , MS
1213	1213 ^d	1,8-Cineole	tr	t _R , MS
1376		trans-Muurola-3,5-diene	0.3	MS
1499	1499 ^c	α-Campholenal	0.7	MS
1535	1535 ^e	Pinocamphone	0.1	MS

Table 1: Composition of the Essential Oil of Nepeta Pilinux

Screening of Therapeutic Potential and Compounds of Endemic Nepeta Pilinux P.H. Davis in Kew Bull. from Sanliurfa

1553	1553 ^d	Linalool	2.3	t _R , MS
1577	1577 ^{e,g}	α-Cedrene	0.3	MS
1586	1586 ^c	Pinocarvone	0.7	MS
1617	1613 ^g	β-Cedrene	tr	MS
1648	1648 ^e	Myrtenal	0.6	MS
1670	1670 ^d	trans-Pinocarveol	1.3	t _R , MS
1684	1684 ^e	trans-Verbenol	4.4	MS
1694	1693 ^g	β-Acoradiene	tr	MS
1704	1704 ^{c,d}	γ-Muurolene	14.4	MS
1706	1706 ^d	α-Terpineol	tr	t _R , MS
1726	1725 ^e	Verbenone	tr	t _R , MS
1747	1740 ^e	p-Mentha-1,5-dien-8-ol	tr	MS
1751	1751 ^d	Carvone	tr	t _R , MS
1797	1804 ^d	Myrtenol	tr	MS
1845	1845 ^d	trans-Carveol	tr	t _R , MS
1853	1849 ^d	cis-Calamenene	1.8	MS
2008	2008 ^c	Caryophyllene oxide	0.6	t _R , MS
2050	2050 ^e	(E)-Nerolidol	0.4	t _R , MS
2080	2080 ^d	Cubenol	3.7	MS
0000		6-Methyl-5 (3-methyl phenyl)-2-	0	MO
2089		heptanone	0.8	IVIS
2187	2187 ^d	T-Cadinol	31.3	MS
2256		<i>epi-</i> α-Bisabolol	0.3	t _R , MS
0057	2233 ^f	Cadalana	0.5	MO
2237	2256 ^e	Cadalene	0.5	1012
2259	2219 ^c		1 /	+ MC
2200	2255 ^d	u-Caullioi	1.4	ι _R , ivis
2264	2264 ^e	4,7-dimethyl-1-tetralone	0.8	MS
2320	2324°	14-Nor-cadin-5-en-4-one isomer A	11.0	MS
2349	2349 ^e	Cadina-4, 10 (15)-dien-3-one	6.1	MS
	2931°			
2931	2913 ^f	Hexadecanoic acid	3.1	MS
		Monoterpene hydrocarbons	2.3	
		Oxygenated monoterpenes	10.1	1
		Sesquiterpenes hydrocarbons	17.3	
		Oxygenated sesquiterpenes	44.6	
		Others	14.9	
		Total %	89.2	
	1		00.1	1

RRI Relative retention indices experimentally calculated against *n*-alkanes; RRI^b RRI from literature [c (48); d (49); e (50); f (51); g (52)] for polar column values; % calculated from FID data; tr; Trace (<0.1 %); Identification Method (IM): $t_{\rm B}$, Identification based on comparison with co-injected with standards on a HP Innowax column; MS, identified on the basis of computer matching of the mass spectra with those of the libraries.

b) HPLC-DAD Analysis of Phenolic Compounds in the Methanol Extracts

RRI^a:

The list of standard phenolic compounds has a wide range of phenolic standards such as 5 benzoic acids (gallic, protocatechuic, p-OH benzoic, vanilic, and syringic acid), 1 hydroxybenzaldehyde (vanillin), 5 cinnamic acids (caffeic, p-coumaric, ferulic, chicoric and chlorogenic acid), 3 flavanols (catechin EGCG, epicatechin), 1 flavonol with its 1 glycoside (quercetin and rutin), 3 flavones with 2 sugar attached derivatives, (luteolin, apigenin, tangeretin, luteolin-7-glucoside, apigenin-7-glucoside), 1 flavanone with its 1 glycoside (hesperidin and hesperetin), and rosmarinic acid which is a caffeic acid ester (supp. Table 1).

A total 6.9 g and 2.1 g of phenolic compounds per 100 g of NP-LF and NP-S, respectively, were quantified by HPLC-DAD (Table 2). The stem part of the plant had at least 3 times less amount of phenolic compounds than the leaf and flower parts (NP-LF). Rosmarinic acid (RA) was found to be the major compound at 3.1 g per 100 g of NP-LF (3.1%) as expected from Lamiaceae family members. The peaks which had same UV spectrum as RA but eluted earlier were identified as RA derivatives (der.) and guantified as equivalent of RA (Fig. 1). They could be the sugar or other functional groups attached to the RA causing this early elution. Total ratio of RA derivatives in the extracts were 4.3% for NP-LF and 0.3% for NP-S (*Table 2*). Caffeic acid and derivatives including RA covered 4.5% of the NP-LF.



Figure 1: HPLC-DAD chromatogram of the extract from leaf and flower parts of Nepeta pilinux

Protocatechuic, chlorogenic, caffeic, *p*coumaric, chicoric acids and luteolin-7-glycoside were also determined in the extracts by comparing the retention time and UV spectrum of the peaks with those of the standard phenolics. There was extremely large peak among others in the chromatograms at 254 nm which was eluted at around 25.9 min and had overlaying two compounds with 243 and 257 nm maximum wavelengths. These peaks were identified as *p*-OH benzoic acid der. and quantified as its equivalent (*Fig. 1 and Table 2*). The peak eluting at 21.4 min. and having max. absorbance at 293 nm was identified as *p*-OH benzaldehyde (4-hydroxybenzaldehyde) since its elution order and UV spectrum were consistent with this phenolic aldehyde (53). The regression equation of vanillin which is the 4-hydroxy-3-methoxybenzaldehyde was used to quantified *p*-OH-benzaldehyde due to their structural and spectral similarity. Four tangeretin derivatives were identified tentatively by comparing UV spectrum of these peaks with that of tangeretin. Since tangeretin has 5 metoxy groups without hydroxyl group, it is eluting in the last part of the chromatogram from

Table 2: The amount and spectral details of phenolic compounds determined in the extracts by HPLC-DAD

	R	Г	Maximum Wavelength		mg/100g) extract
	NP-LF	NP-S	k max	Compounds	NP-LF	NP-S
1	ND	8.6	259-292	protocatechuic acid	ND	5.5
2	15.8	15.3	232-284	catechin der.	83.8	57.7
3	20.4	20.3	330-300sh-232	rosmarinic acid der.	1055.9	164.9
4	21.4	21.2	293	<i>p</i> -OH benzaldehyde	2.8	16.2
6	22.0	21.7	326-295sh-237	chlorogenic acid	86.9	6.7
7	22.9	ND	328-295-242	caffeic acid	75.9	ND
8	ND	22.9	315-230	p-coumaric acid der.	ND	14.2
9	23.7	ND	333-243	chicoric acid der.	2.8	ND
10	ND	23.9	281-229	syringic acid der.	ND	5.4
11	24.5	24.3	329-302sh-249	rosmarinic acid der.	84.3	31.9
12	25.2	25.0	330-300sh-243	rosmarinic acid der.	29.0	6.7
13	25.9	25.7	243 and 257 mix	p-OH benzoic acid equivalent	2204.7	1605.3
14	26.8	26.6	229-310	p-coumaric acid	3.7	2.4
15	27.1	26.8	352-255	luteolin-7-gycoside der.	12.7	5.7
16	28.9	ND	312-295-228	p-coumaric acid der.	11.6	ND
17	29.2	29.0	329-305sh-243	chicoric acid	3.7	2.1
18	ND	30.8	329-233	rosmarinic acid der.	ND	3.7
19	31.5	31.2	338-253-225	apigenin-glycoside der.	54.4	3.9
20	ND	31.4	342-246-226	luteolin-glycoside der.	ND	2.5
21	32.4	ND	327-228 and 347-270	ferulic acid der. and luteolin-7-glycoside	mix	ND
22	ND	32.4	320-244	caffeic acid der.	ND	1.2
23	33.2	33.0	228-278	syringic acid der.	31.4	20.1

Screening of Therapeutic Potential and	Compounds of Endemic N	Jepeta pilinux P.H. Davis in	Kew Bull. from
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24	33.4	33.2	228-278	syringic acid der.	17.8	17.3
25	ND	33.4	331-275	apigenin-glycoside der.	ND	3.2
26	ND	34.1	338-275	apigenin-glycoside der.	ND	5.6
27	35.4	35.3	330-290sh-229	rosmarinic acid	3099.7	59.1
28	ND	35.7	314-230	p-coumaric acid der.	ND	2.7
29	36.6	ND	328-300sh-234	caffeic acid der.	21.6	ND
30	ND	38.3	319-230-295sh	ferulic acid der.	ND	6.8
31	38.7	38.6	328-277-226	tangeretine der.	10.7	43.6
32	39.4	39.2	329-277	chicoric acid der.	6.3	1.6
33	ND	39.4	338-276	apigenin der.	ND	5.6
34	39.8	ND	327-300sh-239	chlorogenic acid der.	8.4	ND
35	ND	41.6	267-340	apigenin der.	ND	6.4
36	ND	42.0	324-270	tangeretin der.	ND	1.6
38	ND	45.1	328-277-214	tangeretin der.	ND	1.4
39	ND	49.8	327-275-223	tangeretin der.	ND	7.4
40	ND	50.3	316	p-coumaric acid der.	ND	16.2
41	50.4	50.6	315-214	p-coumaric acid der.	11.5	0.9
				total	6919.7	2135.2
				total rosmarinic acids	4269.0	266.3

Since Nepeta pilinux is an endemic species and newly recorded in Sanliurfa flora, there is not any scientific report for its phenoreverse phase chromatography (Table 2). Tangeretin derivatives may differ from each other in the number of hydroxyl and methoxyl groups and the presence of sugar or other functional groups. Thymusin which is the 5,6,4'-Trihydroxy-7,8-dimethoxyflavone (5,6,4' demethyltangeretin) and isothymusin were determined in Nepeta asterotricha Rech. (29). Therefore, two of the tangeretin der. could be thymusin and isothymusin. Apigenin and luteolin derivatives, determined in the extracts based on their UV spectra, were identified and quantified as apigenin, apigenin glycoside der. or luteolin glycoside der. based on their elution orders.

lic composition. On the other hand, there are some reports for chemical composition of the other species of Nepeta genus to compare with. For instance, epicatechin, caffeic acids, chlorogenic acid, quercetin and ellagic acid, well-known as antioxidant compounds were found in the ethanol extract of Nepeta cadmea Boiss. by HPLC-DAD (27). They couldn't have determined RA because it was not in their standard compound list. Unlike their results, epicatechin, guercetin and ellagic acid were not detected in Nepeta pilinux extracts (Table 2). Observation of similar but different phenolic compounds in species shows that differences between species also cause differences in metabolic product synthesis. In addition, different derivatives of phenolic compounds can be observed in samples of the same species grown in different geographical conditions. RA, apigenin, and quercetin were determined as major compounds in the ethanol

extract of *Nepeta nuda* subsp. *albiflora* (28). Isolation of some iridoid glycosides such as nepetamoside, nepetaracemoside B, nepetonic acid and some polyphenol and flavonoid components such as RA and its methyle ester, thymusin, luteolin and apigenin from *Nepeta asterotricha* Rech. were performed by Goldansaz et al. (2019) (29). Methanolic extracts from *Nepeta rtanjensis* Diklic & Milojević which is an endemic perennial plant, in a very limited area in Southeast Serbia were investigated in a scientific study (30). The presence of high levels of chlorogenic acid, RA and rutin in these extracts was thought to be the reason for their antigenotoxicity.

RA, which is the main component in these extracts, is synthesized by the phenylpropanoid pathway starting with L-phenylalanine and L-tyrosine. From Lphenylalanine, t-cinnamic acid, p-coumaric acid and pcoumaroyl-CoA are successively produced, while phydroxyphenylpyruvic acid, p-hydroxyphenyllactic acid are successively produced from L-tyrosine. Then, the production of p-coumaroyl-p-hydroxy-phenyllactic acid condensation of p-coumaroyl-CoA and by phydroxyphenyllactic acid is followed by RA synthesis (54). Chlorogenic acid is synthesized by the condensation of p-coumaroyl-CoA with guinic acid (55). These metabolites which were found in the Nepeta pilinux extracts, support the survival of plants against harsh environmental conditions, therefore they have been seen by humans as a remedy for various diseases for many years. It has been reported that RA, chlorogenic acid and its metabolite caffeic acid have a neuroprotective effect due to their antioxidant capacity (56, 57).

c) Total Antioxidant Characterization of the Methanol Extracts

Total phenolic content (TPC) and radical scavenging activities (RSA) against DPPH and ABTS radical were evaluated to characterize the antioxidant capacity of the extracts and results were presented in Table 3. TPC of the NP-LF and NP-S extract from Nepeta pilinux was 50.81 and 13.37 mg GAE/g respectively. Consistent with the TPC of NP-LF which was found around fourfold of NP-S, RSA of NP-LF were higher (433.18 and 82.29 μ g / mL for DPPH and ABTS respectively) than NP-S as expected (Table 3). The antioxidant capacity of this species was found to be quite consistent comparing with other species in the literature reports. Antioxidant activity of methanolic extracts from flower and leaf parts of Nepeta rtanjensis were reported by Bošnjak-Neumüller et al. (2017). They have found that the leaf extract had higher antioxidant capacity with TPC of 62.73 mg of GAE/g and IC₅₀ value

of RSA against DPPH as 112.59 µg/mL than those of flower extract. The antioxidant activities of ethanol, methanol, acetone, and water extracts from Nepeta cadmea were presented by Kaska et al. (2018). They have found that the water extract had the highest RSA (IC₅₀ value of DPPH as 25.54 μ g/mL and ABTS, 14.51 μ g/mL) in these four extracts. Highest TPC with 79.84 mg GAE/g, was found in the methanol extract while the highest total flavonoids with 77.09 mgQE/g was in the acetone extract. It was reported by Teber and Bursal (2020) that ethanol and water extracts of Nepeta nuda subsp. albiflora had strong antioxidant effects with IC₅₀ values of DPPH as 54.4 and 113.0 µg/mL, respectively. The TPC of flowers, leaves and roots methanol extracts of Nepeta humulis were found as 123.18, 66.20 and 54.77 mg GAE/g extract, respectively. Flower extract which had the highest TPC displayed best RCA with IC₅₀ of 1290 and 350 µg.mL⁻¹ against DPPH and ABTS respectively (58).

Table 3:	Total phenolic	content, radica	I scavenging a	activity and	enzyme inhib	ition of the extracts
	•		0 0		5	

	TPCª		RSA ^b		Enzyme inhibition	
Samples and standards	mgGAE/g	mgQE/g	DPPH SC50	ABTS SC50	urease IC ₅₀	XO IC ₅₀
NP-LF	50.81±1.50	35.40 ± 1.06	433.18±12.74	82.29±0.98	62.47±0.10	48.48±0.10
NP-S	13.37±0.34	9.03±0.24	523.49±5.82	381.58±5.15	230.59±0.23	222.67±0.13
Gallic Acid Quercetin			1.52±0.06 5.87±0.13	3.29±0.09 8.52±0.16		
Acetohydroxamic acid					24.56±0.29	
Allopurinol						$0.54 {\pm} 0.04$

GAE, Gallic acid equivalent; QE, quercetin equivalent; SC₅₀, value of the concentration of extract required to scavenge 50% of DPPH and ABTS radicals (μ g extract per mL methanol); IC₅₀, value of the concentration of extract required to inhibite 50% of Jack bean urease and bovine milk xanthine oxidase enzymes (μ g extract per mL methanol). ^aTotal phenolic contents are expressed in mg GAE/g extract and mg QE/g extract. ^bRadical Scavenging Activity

d) Urease and Xanthine oxidase inhibitions of the methanol extracts

The urease enzyme inhibition of the NP-LF with IC_{50} value of 62.47 μ g.mL⁻¹ was only three times lower than the inhibition of acetohydroxamic acid (24.56 μ g.mL⁻¹) which is standard medicine (*Table 3*). NP-S had low inhibition against to this enzyme with IC₅₀ value of 230.59 μ g.mL⁻¹. The xanthine oxidase inhibition of the NP-LF and NP-S extracts, 48.48 and 222.67 μ g.mL⁻¹ respectively, were quite lower comparing to the inhibition of allopurinol (0.54 μ g.mL⁻¹) which is standard medicine reducing the production of uric acid in the body caused by certain cancer medications and kidney stones. Akdeniz et al. (2020) were screened urease inhibition effect of the essential oils and ethanolic extracts of Nepeta heliotropifolia and Nepeta congesta subsp. cryptantha comparing with the standard thiourea. They reported that none of them exhibited urease inhibitory activity (47). In another study, ethyl acetate sub fraction of Nepeta praetervisa showed significant urease inhibitory activity (68%) (59). The structureactivity relationship revealed that the planar flavones and flavonols with a 7-hydroxyl group such as chrysin, luteolin, kaempferol, quercetin, myricetin, and isorhamnetin inhibited xanthine oxidase activity at low concentrations, while the nonplanar flavonoids, isoflavones and anthocyanidins were less inhibitory (35). Although the inhibitory effects of the extracts cannot compete with acetohydroxamic acid and allopurinol, the fact that they are a natural herbal inhibitor source shows that these extracts are more suitable for use.

IV. CONCLUSION

This was the first investigation on the chemical composition and bioactivities of the Nepeta pilinux. The polar and apolar extracts of aerial parts had high amount of phenolic and essential oil compounds and demonstrated the potential antioxidant capacities. The polar extract with 3.1% rosmarinic acid had urease and xanthine oxidase inhibition as well. This comprehensive evaluation of *Nepeta pilinux* revealed that this endemic plant could be the source of valuable therapeutic compounds. Besides, this report would be the incentive for further works on this plant's metabolites.

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Disclosure Statement

No potential conflict of interest was reported by the authors

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Recurrent Relationships as an Important Class of Mathematical Equations in Chemistry

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Abstract- The unique potential and different areas of applications of recurrent (synonymous: recursive) relationships in chemistry and chromatography are considered. Recurrent relations can be used in two forms: as functions of integer arguments, y(x + 1) = ay(x) + b, and as functions of equidistant argument values, $A(x + \Delta x) = aA(x) + b$, $\Delta x = \text{const.}$ The first form applies to all physicochemical properties of homologs in organic chemistry, because the number of carbon (and other) atoms in a molecule can be integer only. The second one applies to chemical variables depending on temperature, pressure, concentrations, etc., when the chemists should provide equal "steps" of their variations.

Recurrent relations combine the properties of arithmetic and geometric progressions, which accounts for their unique approximation abilities. This was illustrated by approximating the number of isomers of alkanes, the boiling points of homologs (nonlinear dependencies), the melting points of homologs (alternation effects), the temperature dependence of the solubility of inorganic salts in water, and by revealing the anomalies of gas chromatographic retention indices and retention times in reversed-phase high performance liquid chromatography.

Keywords: recurrent relations, chemistry, chromatographic retention, chemical variables, number of isomers, physicochemical properties, linear approximation.

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Recurrent Relationships as an Important Class of Mathematical Equations in Chemistry

Igor G. Zenkevich

Abstract- The unique potential and different areas of applications of recurrent (synonymous: recursive) relationships in chemistry and chromatography are considered. Recurrent relations can be used in two forms: as functions of integer arguments, y(x + 1) = ay(x) + b, and as functions of equidistant argument values, $A(x + \Delta x) = aA(x) + b$, $\Delta x =$ const. The first form applies to all physicochemical properties of homologs in organic chemistry, because the number of carbon (and other) atoms in a molecule can be integer only. The second one applies to chemical variables depending on temperature, pressure, concentrations, etc., when the chemists should provide equal "steps" of their variations.

Recurrent relations combine the properties of arithmetic and geometric progressions, which accounts for their unique approximation abilities. This was illustrated by approximating the number of isomers of alkanes, the boiling points of homologs (nonlinear dependencies), the melting points of homologs (alternation effects), the temperature dependence of the solubility of inorganic salts in water, and by revealing the anomalies of gas chromatographic retention indices and retention times in reversed-phase high performance liquid chromatography. In the latter case, specific anomalies in the recurrent approximation of retention times allow detecting the reversible hydration of analytes.

The restricted text cannot represent all the examples of the applications of recurrent relations in chemistry and chromatography. However, even this short review allows us to conclude that the potential of recurrences should not be underestimated.

The author confirms that the objective difficulties of the perception of recurrences seem to be the main reason for their rare use in chemistry. In such a situation, any examples of demonstrating their capabilities are significant. Consequently, the author is very grateful to the Editorial Board of GJSFR for the kind offer to submit a manuscript on this topic.

Keywords: recurrent relations, chemistry, chromatographic retention, chemical variables, number of isomers, physicochemical properties, linear approximation.

I. INTRODUCTION

Athematics is an essential tool for data processing in all natural sciences, including chemistry [1–4]. However, in all of them, there are several stereotypes of presenting different mathematical equations. Most of them are the following: in equations of the form y = f(x, ...) the argument(s) (x, ...) are typically on the right side, while the function is on the left. The number of possible examples is so large that it is difficult to estimate it. For instance, the well known Antoine equation which has many chemical applications relates the absolute temperature (*T*, argument), and the vapor pressure of pure liquids (*P*, function): log P = a/T + b (the coefficients *a* and *b* should be precalculated). If necessary, the argument and function can be swapped and this relationship can be used to calculate the temperature from the pressure. Besides that numerous analogues of Antoine equation are used in different areas. For instance, the dependence of retention times vs. temperature of gas chromatographic column is described with Antoine-like equation $\log(t_{\rm B}') = a/T + b$.

Numerous other forms of representing dependences (e.g., parametric) are known for mathematicians, but in chemistry they are perceived as rather unusual.

Recurrent equations are precisely such unusual forms of mathematical relationships. The main feature of them is the absence of arguments in their writing: recurrent equations relate the current value of a function to its previous value(s). The rare use of these relations is confirmed by the fact that many contemporary manuals have no information about them. Recurrent (recursive) progressions are mentioned in [2, P. 174] only as a particular case for comparing with other progressions (arithmetic, geometric, etc.). Such an attitude towards recurrences cannot be accepted, since the properties of recurrences make them indispensable mathematical object in chemistry. The main reason of that is, probably, just unusual mathematical form of recurrences, namely the absence of arguments.

The purpose of this article is to illustrate the application of recurrences with examples from various areas of chemistry and chromatography.

Generally, the most straightforward (first-order) linear recurrences can be represented in two ways. *The first kind of recurrence* combines the values of functions of discrete integer arguments (n + 1 and n) in relation (1):

$$y(n + 1) = ay(n) + b$$
 (1)

The coefficients *a* and *b* should be precalculated by the least-squares method using data for a preselected training set.

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At first glance, the scope of application of such relations with integer arguments seems to be somewhat limited. However, it is not; in organic chemistry many properties of organic compounds are considered as the functions of their positions in the corresponding homologous series, or, in other words, of the number of carbon (or other elements) atoms in the molecule. Meanwhile, the number of atoms of any element in a molecule can take integer values only. It is the principal reason for the applying of recurrent dependences to various properties of organic compounds. One can only be surprised that this was not been done before 2006, but it may only be explained by the unfamiliarity of these relations.

The second form of recurrent dependences applies to any functions for which it is possible to provide the equidistant argument values, namely, in the form $(x + \Delta x)$, where $\Delta x = \text{const}$ (relation 2). In other words, the experimenter must select appropriate argument values from their possible multitude:

$$y(x + \Delta x) = ay(x) + b, \quad \Delta x = \text{const}$$
 (2)

Such an expansion of the domain of definition allows applications of recurrences to functions of continuous variables such as temperature (T), pressure (P), concentrations (C), etc., i.e., not only to properties of homologs in organic chemistry, but also to characteristics of chemical systems in physical, analytical, inorganic chemistry, etc.

The general feature of recurrences is the absence of argument values in both parts (left and right) of equalities. Every point on the plots of recurrences is specified by two values of functions: current and previous, or current and subsequent ones. The second feature of equation (2) is the appearance of the additional (artificial) variable Δx : the shape of the graph begins to depend on this value.

As a result, both forms of recurrences are somewhat difficult to visualize, which explains in some way their rare use in scientific practice. To illustrate this, let us consider as an example the recurrent representation of any well-known function v(x). Let us select 36 points within the domain of its definition with the increment $\Delta x = 10$ and build a plot corresponding to the equation y(x + 10) = ay(x) + b, shown in Fig. 1.



Fig. 1: Recurrent representation of a well-known mathematical function in the form y(x + 10) = ay(x) + b. It is not so easy to recognize it

We obtain some kind of closed ellipsoidal curve, which has no analogues among the plots of commonly used functions. If we change the value Δx and select $\Delta x = 1$ (or less) instead of $\Delta x = 10$, the ellipsoidal curve will turn into a practically straight segment located between points (-1, -1) and (+1, +1). So as not to continue the intrigue, let us note that this function is $y = \sin x$, but it is rather difficult to "recognize" it from the recurrent graph without special preparation.

First-order recurrent relations like (1) or (2) look like simple mathematical expressions only at a first glance. These equations have the non-recurrent algebraic solution that is an unusual type of polynomial of variable degree, because the values of the argument are the power:

$$y(x) = y(0)a^{x} + b(a^{x} - 1) / (a - 1)$$
(3)

This solution can be easily found using MAPLE software. The value of the auxiliary parameter y(0)should be predetermined, for example, using the value y(1), because at x = 1 y(1) = y(0)a + b, and y(0) = [y(1)]-b]/a.

This solution is the row, because $(a^x - 1) / (a -$ 1) = $a^{x-1} + a^{x-2} + \dots + 1$. The variable degree of this polynomial is a crucial explanation of the usefulness of applying recurrences to approximation of chemical variables. This fact deserves special comments.

Example 1: Let us try to approximate the nonlinear dependence of normal boiling points of 1-chloroalkanes $C_nH_{2n+1}Cl$ on the number of carbon atoms in a molecule: $46.2^{\circ}C$ (n = 3), $78.4^{\circ}C$ (n = 4), $107.8^{\circ}C$ (n = 5), $135.6^{\circ}C$ (n = 6), $160.6^{\circ}C$ (n = 7), $183.8^{\circ}C$ (n = 8), $205.4^{\circ}C$ (n = 9), and so on. The commonly used way to do this is choosing the fixed-degree polynomial (e.g., second or third). At the same time, the polynomial of variable degree (3) will have the degree x = 3 for propyl chloride (n = 3), x = 4 for butyl chloride (n = 4), x = 5 for pentyl chloride (n = 5), etc.

In addition, a noticeable consequence of the application of this approach can be derived from this solution. If $a \equiv 1$ and $b \neq 0$, equation (3) transforms into the relation for a simple arithmetic progression, y(x) = k + bx. At the same time, if $0 < a \neq 1$ and $b \equiv 0$, this equation transforms into the expression for geometric progression, $y(x) = ka^x$. Hence, in the general case (at arbitrary values of coefficients *a* and *b*), the recurrent equation (1) and/or (2) combines the mathematical properties of both kinds of progressions in variable proportions. It explains us the excellent approximating "power" of recurrences for numerous chemical variables. Thus, we must conclude that recurrences are not the particular case of other progressions [2] but are their generalization.

The second noticeable consequence from equation (3) is the behavior of function y(x) or $A(n_c)$ with a hypothetical increase in the number of carbon atoms in a molecule, $x \rightarrow \infty$, or $n_c \rightarrow \infty$. If the value of coefficient *a* obeys the inequality a < 1, the limit of polynomial (3) exists, and it is equal to:

$$\lim A(n_{\rm C})|_{\rm nC\to\infty} = b/(1-a)$$
 (4)

If a > 1, the initial numerical sequence has no limit and tends to infinity.

The author cannot cite all his papers devoted to the properties and chemical applications of recurrences due to the necessity to restrict the self-citation. Only six of them published within the period from 2009 to 2024 are mentioned as references [5–10].

To conclude the Introduction, it should be noted that, probably, the most famous and widely mentioned examples of recurrences are so-called Fibonacci numbers, F_n , known since ancient times [11–13]:

$$F_{n} = F_{n-2} + F_{n-1}$$
, $(F_{0} = F_{1} = 1)$ (5)

A special journal named *The Fibonacci Quarterly* has been published since 1963 (see ref. [11]).

II. Experimental

This semi-review article does not imply specific experimental conditions. The necessary data are

available from the references cited. In general, recurrent relationships can be applied to any non-linear set of equidistant variables not only in chemistry or chromatography. For example, the temperatures of the cooling kettle, $T(^{\circ}C) = f(t, \min)$ are: 72(0), 62(5), 55(10), 50(15), 46(20), etc. All these data correspond to the linear recurrent dependence $T(t + 5 \min) = aT(t) + b$ with the following parameters: $a = 0.72 \pm 0.01$. $b = 10.0 \pm 0.8$, R = 0.9996, $S_0 = 0.2$.

Another important example is the approximation of the carbon dioxide content in the atmosphere of Earth (data from site http://climate.gov),(CO₂ concentration (ppm)) = f(year): 324(1970), 337(1980), 353(1990), 369(2000), 390(2010), and 412(2020). Parameters of the recurrence C(year + 10) = aC(year) + b are: a = 1.14 ± 0.01 , $b = -33 \pm 8$, R = 0.9994, $S_0 = 1.2$. After that we can easily (because the dependence is linear) estimate the concentration of CO₂ in the atmosphere at 2030: it should be $1.14 \times 412 - 33 = 437$ ppm.

The values of different physicochemical properties of organic compounds (boiling points, melting points, etc.) were taken from all available sources of reference information, e.g., [14], as well as from Internet. The necessary stage of processing the initial information was the rejection of outliers, followed by averaging the data. It is required because recurrences are very "sensitive" even to minor errors in data sets.

The simplicity of recurrent calculations requires special comments. The use of, e.g., Origin software implies the following stages. First, the initial data on the function of integer (equation 1) or equidistant (equation 2) arguments are entered into column A[X]. The next step is copying all data from this column excluding the value from the first line into column B[Y] with shifting one line up. Just such shifting is the sense of recurrences. After that, we can operate with the obtained twodimensional data array in the usual way, i.e., plot this dependence and calculate all the parameters of the linear regression y(x+1) = ay(x) + b. For example, let us make the recurrent approximation of the squares of natural numbers from 3 to 10. Hence, the first column (Origin or Excel software) contains the values 9, 16, 25, 36, 49, 64, 81, and 100 (eight values). The second column (shifted) in the same lines contains the values 16, 25, 36, 49, 64, 81, and 100. In the result we obtain seven pairs of numbers for calculating the parameters of linear regression by least squares method: $a = 1.16 \pm$ 0.01, $b = 4.5 \pm 0.5$, R = 0.9998, $S_0 = 0.7$. The plot of this dependence looks like "ideal" straight line. Such calculations are very simple, and the application of no special functions is required.

RESULTS AND DISCUSSION III.

Chemical Applications a)

i. Number of Isomers

In this section, it seems advisable to avoid considering the simplest artificial examples and start directly with real chemical problems. The first is the dependence of the number of structural isomers (N) of homologs on the number of carbon atoms in their molecules $(n_{\rm c})$. The number of such isomers for *n*alkanes C_5-C_{18} is listed in Table 1; it is the rapidly increasing function [15]. It is noteworthy that estimates of the number of isomeric alkanes with $n_{\rm C}$ = 18 and above, obtained by different methods, appeared to be slightly different, e.g., 60524 vs. 60523 for $C_{18}H_{38}$. However, such small differences for many practical purposes are negligible.

Number of Carbon Atoms, <i>N</i> c	Number of Structural Isomers, <i>N</i>	Log <i>n</i>
5	3	0.477
6	5	0.699
7	9	0.954
8	18	1.255
9	35	1.544
10	75	1.875
11	159	2.201
12	305	2.550
13	802	2.904
14	1858	3.269
15	4347	3.638
16	10359	4.015
17	24894	4.396
18	60524 (60523)*	4.782

Table 1: Number of structural isomers of alkanes C_nH_{2n+2}

The plot of this function $N = f(n_c)$ for alkanes with $5 \le n_{\rm C} \le 17$ is presented in Fig. 2(a). It seems rather challenging to precalculate the number of isomeric alkanes with a higher number of carbon atoms because we do not know the type of this function (it should be evaluated or assumed preliminarily).

However, if we transform this dependence to the linear form, the problem's solution enormously simplifies. because we can evaluate other values using the simplest linear extrapolation. Just recurrent approximations allow us to implement this approach.



Fig. 2: (a) Plot of the number of structural isomers of alkanes C_nH_{2n+2} (N) vs. the number of carbon atoms in the molecule (n_c) ; (b) plot of the recurrent dependence $N(n_c + 1) = aN(n_c) + b$; parameters of the linear regression: a = 2.429 ± 0.004, b = -62 ± 30 , R = 0.99999, S₀ = 98; (c) plot of the recurrent dependence of logarithms, logN(n_c + 1) = $a\log N(n_c) + b$; parameters of the linear regression: $a = 1.036 \pm 0.005$, $b = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.998, $S_0 = 0.248 \pm 0.012$, R = 0.998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.9998, $S_0 = 0.248 \pm 0.012$, R = 0.012, R0.02

Approximating the data set in the form $N(n_{\rm C} +$ 1) vs. $N(n_c)$ gives the plot presented in Fig. 2(b). It is a linear dependence with the correlation coefficient R =0.99999. Other parameters of the linear regression are indicated in the footnote to Fig. 2. If the correlation exceed 0.999, the plots coefficients of such dependencies are visually perceived as absolutely straight lines. It is an essential moment in the
characterization of recurrent dependencies. For numerous chemical variables, these dependencies are linear, but if not, the reasons of deviations from linearity should be investigated. If we will use this result for evaluating the number of isomers of alkane $C_{18}H_{38}$ using the number of isomers for the previous alkane $C_{17}H_{36}$, we should make the following simple calculations:

$24894 \times 2.429 - 62 \approx 60406$

For the example considered the evaluation of uncertainty appeared to be approximately \pm 98. The correct answer is 60524 or 60523 isomers (Table 1). Hence, the relative error is approximately 0.19%. Such accuracy seems to be enough for many practical purposes.

This example illustrates another feature of recurrent approximation. If it is linear for variable x, it remains linear for any monotonic functions of this variable, e.g., x^n , log(x), exp(x), etc. Hence, we can approximate not the number of isomeric alkanes (*N*) but

the log *N* values. Fig. 2(c) illustrates the linear recurrent dependence for logarithms of isomers' number.

The number of isomers of homologs of other series [16, 17] can be approximated using recurrent relations without any restrictions.

ii. Approximation of Boiling Points of Homologs

To illustrate the potential of recurrent relations in form (1) (functions of integer argument), let us consider the approximation of values of any physicochemical properties of homologs. The simplest example is normal (at the atmospheric pressure) boiling points (T_b) of *n*alkanes C_nH_{2n+2} within the set of compounds C_2-C_{14} (if necessary, this range can be expanded); the reference data are readily available [14]. The plot in Fig. 3(a) illustrates the initial nonlinear dependence $T_b(n_c)$; the linear recurrent approximation $T_b(n_c + 1) = aT_b(n_c) + b$ is illustrated in Fig. 3(b). The parameters of the linear regression are listed in the footnote to this figure.



Fig. 3: (a) Dependence of the normal boiling points of n-alkanes C_1-C_{14} (°C) on the number of carbon atoms in the molecule; (b) recurrent approximation of the normal boiling points (equation 1); parameters of linear regression: $a = 0.916 \pm 0.003$, $b = 36.9 \pm 0.5$, R = 0.99993, $S_0 = 1.2$; (c) recurrent approximation of the normal boiling points for all alkanes (including branched isomers); parameters of the linear regression: $a = 0.922 \pm 0.003$, $b = 35.6 \pm 0.5$, R = 0.9997, $S_0 = 2.0$; (d) recurrent approximation of the normal boiling points for perfluoro-1-alkanes C_nF_{2n+2} ; parameters of the linear regression: $a = 0.912 \pm 0.001$, $b = 32.5 \pm 1.4$, R = 0.9991, $S_0 = 4.2$

The next exciting feature of recurrences does not follow from their mathematical properties but has a critical chemical sense. Namely, if the recurrent linearization is valid for normal alkanes (with nonbranched carbon skeleton), the equation with practically the same coefficients will be valid for all isomeric structures of the same class, i.e., for alkanes with branched carbon skeletons. It is illustrated by Fig. 3(c) (39 points for iso-alkanes were added). Comparing the coefficients of recurrences (b) and (c) confirm their identity: 0.916 and 0.922 (values of coefficient a), as well as 36.9 and 35.6 (values of coefficients b). Moreover, combining the data for 189 homologs of 10 different homologous series confirms the existence of a single linear relationship for all of them with the following parameters: $a = 0.930 \pm 0.002$, $b = 33.5 \pm 0.3$, R =0.9995, $S_0 = 2.4$ [7]. It allows us to conclude that coefficients a and b or recurrent regressions are close to each other for all homologous series with the same homologous differences, at first CH₂,

The following example illustrates the applicability of recurrences not to alkanes but compounds of other chemical nature. The plot of the recurrent approximation of boiling points of perfluoro-*n*-alkanes C_1-C_{14} is presented in Fig. 3(d).. The correlation coefficient of this dependence exceeds 0.999.

The linearity of recurrent relations makes it possible to evaluate the boiling points of any next homologs using the data for previous ones.

Example 2: Let us evaluate the normal boiling point of 1chlorodecane using $T_{\rm b}$ value for 1-chlorononane (205.4°C, see Example no. 1). The values of recurrent coefficients for the series of chloroalkanes are a = 0.923 ± 0.002 , $b = 35.7 \pm 0.3$, R = 0.99999, and $S_0 = 0.3$. The next step of calculations is simple: 205.4 $\times 0.923$ + 35.7 = 225.3°C (the reference value is 225.8°C). *Example 3:* Let us evaluate the normal boiling point of *n*butylisocyanate using $T_{\rm b}$ value for *n*-propylisocyanate (88°C). As the data for the series of alkyl isocyanates are insufficient to calculate the coefficients of equation (1), let us use the "universal" values of these coefficients for any homologous series (see above). The result is 88 × 0.930 +33.5 = 115.6°C (the reference value is 115°C).

Recurrent relations provide linear approximations of not only normal boiling points but also the values of other physicochemical properties of organic compounds within homologous series, $A(n_{\rm c})$. These include, for example, relative density (d_4^{20}) , refractive index (n_D^{20}) , dynamic viscosity (µ), surface tension (σ), ionization potential (*I*), acidity constant (pK_a), dielectric permittivity (ϵ), water solubility (w, pS), and many others. The fundamental requirement to the dependences $A(n_c)$ is the same: these functions should be monotonic (the signs of their first derivatives should constant). The example of nonmonotonic be dependencies is, for instance, the temperature dependence of the solubility of organic compounds in water.

Due to the problem of the monotony of changes in the values of physicochemical properties of organic compounds, it is interesting to consider another form of recurrent relations, namely, second-order recurrences:

$$y(n+2) = ay(n) + b$$
 (5)

iii. Melting Points of Homologs

Similarly to the previous ones, the coefficients a and b of equation (5) should be precalculated by the least-squares method. The algebraic solution of equation (5) can be found in the same way as the solution of equation (1) using MAPLE software. Surprisingly, it turned out to be much more complex than the solution of (1):

$$y(x) = \frac{(y(0)a - y(1))\sqrt{a}(-\sqrt{a})\uparrow x - (y(1)\sqrt{a} - ay(0))(\sqrt{a})\uparrow x}{2a} - \frac{b}{a - 1} - \frac{b(\sqrt{a} - a)(-\sqrt{a})\uparrow x + b(\sqrt{a} + a)(\sqrt{a})\uparrow x}{2a(a - 1)}$$
(6)

Relation (6) obviously cannot be recommended for any calculations due to its complexity, but it allows us to make important conclusions. First, this solution contains negative terms to the *x* power, namely (sqr(*a*)) 1x. It means that these terms have different signs depending on whether the *x*-values are even or odd. Therefore, the presence of terms with alternating signs in the equation makes it applicable for approximation of variables with pronounced alternation effects. The most important of these variables are the melting points of homologs with even and odd numbers of carbon atoms in the molecules. The alternation of the melting point is well known for *n*-alkanes, *n*-alkenes, 1-alkanols, carboxylic acids, 1-alkylamines, numerous 1, ∞ - difunctionalized alkanes, etc. Fig. 4 illustrates different kinds of the recurrent data approximation for 1alkylamines. Plot (a) represents the set of initial data, "distorted" with alternation effects. The first-order linear recurrent approximation (equation 1) converts this plot to two nonparallel straight lines (b). Finally, the application of equation (5) to this data set gives a single straight line with the correlation coefficient R = 0.9996(no less than the *R*-values in the other cases mentioned above), but with irregular arrangement of points (c). The parameters of linear regression (c) are listed in the footnote.



Fig. 4: (a) Dependence of the melting points of n-alkylamines C_2-C_{18} (°C) on the number of carbon atoms in the molecule; the alternation effect is easily noticeable; (b) recurrent approximation of the melting points (equation 1); all points belong to two non-parallel lines; (c) recurrent approximation of the melting points for n-alkyl amines (equation 4); parameters of the linear regression: $a = 0.786 \pm 0.006$, $b = 16.1 \pm 0.2$, R = 0.9996, $S_0 = 0.9$

Secondly, despite the complexity of equation (6), it is easy to conclude that the limiting value of y(x) at $x \rightarrow \infty$ is the same as that of the polynomial (3), namely, b/(1 - a). Hence, we can evaluate the melting point for hypothetical 1-alkylamine with $n_c = \infty$, i.e., $16.1 \times (1 - 0.786) \approx 75^{\circ}$ C. The highest-molecular-mass 1-alkylamine for which the melting point has been determined is 1-octadecylamine with T_m 53°C [14].

Another example with alternating melting points of homologous *n*-alkane carboxylic acids is considered in [5].

iv. Temperature Dependence of Solubility of Inorganic Salts in Water

The next possible application of recurrences is approximating the temperature dependence of the solubility of inorganic salts in water, w(T), g/100 g of water. It is known that numerous inorganic and some

organic compounds form hydrates in aqueous solutions. Using the recurrences allows us to detect important features of the data sets on solubility, namely, the formation of hydrates. If the dissolved compound exists in the single chemical form (either hydrated or non-hydrated) at different temperatures of the solution, the recurrent approximation of w(T) dependence has no anomalies; i.e., it is linear. This case can be illustrated with plots for the solubility of copper sulfate within the temperature range $0 \le w(T) \le 100^{\circ}$ C presented in Fig. 5. Despite the nonlinearity of the initial dependence w(T)(a), its recurrent approximation has the correlation coefficient R = 0.9998 (b).



Fig. 5: (a) Nonlinear temperature dependence of the solubility of copper sulfate pentahydrate (CuSO₄:5H₂O); (b) linear recurrent approximation of this dependence, $w(T + 20^{\circ}C) = aw(T) + b$; parameters of the linear regression: a $= 1.36 \pm 0.02, b = 0.4 \pm 0.6, R = 0.9998, S_0 = 0.5$

Another example is the solubility of potassium acetate (Fig. 6). Considering the initial w(T) dependence (a) does not allow us to decide whether the leftmost point is weird or belongs to a general nonlinear dependence. The red line in plot 6(a) results from approximation w(T) by a fourth-degree polynomial. However, if we present the same data in the recurrent form, we obtain the linear dependence for all the points (R = 0.9998), excluding the leftmost one, which is an obvious outlier. It follows from this fact that, at low temperatures (e.g., 0°C), this salt exists in another form, less soluble than we can expect from the dependence w(T) at other temperatures.

b) Applications of Recurrences in Chromatography

i. Revealing the anomalies of gas chromatographic retention indices of chlorinated benzenes

Gas chromatographic retention indices (RI) are important chromatographic characteristics of analytes and are informative for their GC-MS identification. Let us consider the sequence of GC retention indices (RI) for benzene and its chlorinated derivatives with chlorine atoms in unhindered positions (the series of so-called congeners). It includes data for only four compounds. namely, unsubstituted benzene (I), chlorobenzene (II), 1,3-dichlorobenzene (III), and 1,3,5-trichlorobenzene (IV) (all chlorine atoms are in meta-positions relative to each other). It should be noted that all RI values were taken from an independent source of reference information, namely, the NIST RI Database [18]:



RI values on standard non-polar polydimethylsiloxane stationary phases [18]:

989 ± 10

$$654 \pm 7$$
 839 ± 7

 1117 ± 14

 1322 ± 11

Adding the next chlorine atom (1,2,3,5tetrachlorobenzene, V) will cause it to be in an orthoposition to two other chlorine atoms, which gives rise to steric tension in the molecule. However, it is difficult to identify this effect directly from the dependence RI = $f(n_{\rm Cl})$ (Fig. 7(a)). At first glance, only the second point from the right on this plot seems to be located slightly © 2024 Global Journals

below the regression line, which is not understandable. At the same time, plotting the recurrent dependence $RI(n_{CI} + 1) = aRI(n_{CI}) + b$ (Fig. 7(b)) immediately reveals the anomaly just for the rightmost point corresponding to tetrachlorobenzene (V). The steric hindrance in this tetrachlorobenzene increases its retention index above the value extrapolated from data for other (nonhindered) congeners. Similar effects are also observed for retention indices of bromo- and methyl-substituted benzenes.



Fig. 6: (a) Nonlinear temperature dependence of the solubility of potassium acetate (CH₃CO₂K) in water; (b) recurrent approximation of this dependence, $w(T + 10^{\circ}C) = aw(T) + b$; parameters of the linear regression without the leftmost point: $a = 0.882 \pm 0.005$, $b = 10.1 \pm 0.4$, R = 0.9998, $S_0 = 0.5$.



Fig. 7: (a) Dependence of the retention indices of benzene and its four chloroderivatives (see text) on the number of chlorine atoms (standard nonpolar polydimethylsiloxane stationary phase); (b) recurrent approximation of this dependence, $RI(n_{CI} + 1) = aRI(n_{CI}) + b$; parameters of the linear regression without the rightmost point: $a = 0.83 \pm$ 0.01, b = 296 ± 10 , R = 0.9998, S₀ = 2.9

The last example illustrates the next important feature of recurrences. In many cases, the deviations of recurrent approximations from the linearity seem to be no less informative than the linearity of these dependencies. The explanation of this fact appears to be relatively simple: Recurrent relations (1) or (2) allow approximations linear of numerous monotonic dependencies, but preferably if they can be described with a single equation. If these dependencies are distorted by additional effects, the high "linearization ability" of recurrences appeared to be insufficient for their linearization. This feature allowed us to detect the steric effects of the fourth chlorine atom in RI values of chlorinated benzenes.

ii. Revealing the anomalies of retention parameters in reversed-phase high-performance liquid chromatography

Considering the results of the last example, let us discuss the unusual applications of recurrent relations in reversed-phase high-performance liquid chromatography (RP HPLC). It is the contemporary analytical method based on the different distribution of smallest amounts of analyzed compounds (analytes) between moving liquid phase (eluent) and stationary

phase fixed in a chromatographic column (modified silica gels).

The dependences of retention times of analytes on the content of organic solvent in an eluent, $t_{R}(C)$, should be considered first. Establishing the regularities of these dependencies seems to be the general approach in characterization both analytes and sorbents in chromatographic columns [19-22].

First, let us illustrate the general statement mentioned in the previous subsection: If the dependence $t_{\rm R}(C)$ has no anomalies, its recurrent approximation is linear with a high *R*-value. The absence of anomalies means the single law of variations of retention times vs. concentration of organic modifier in an eluent throughout the concentration range under consideration. It can be illustrated with data for acetophenone (methanol-water eluents). Chromatographic analyses were performed using a Shimadzu LC-20 Prominence liquid chromatograph equipped with a diode-array detector and Phenomenex C18 columns 250 mm long and 4.6 mm i. d. with a sorbent particle size of 5 µm. Water-methanol mobile phases were used in several isocratic modes with 5 or 10% concentration steps of methanol at an eluent flow rate of 1.0 mL min⁻¹ and column temperature of 30°C. The samples were injected using an SIL-20A/AC autosampler; the sample volume was 20 µL. To prepare eluents, we used deionized water (resistivity 18.2 M Ω cm) prepared using a Milli-Q device (Millipore, USA), and methanol (analytical grade, Kriokhrom, St. Petersburg). The pH values of the eluent were 6.2–6.3. The number of replicate injections of each sample was 2–3. The interinjection variations of the retention times in all the cases did not exceed 0.01–0.02 min.

The plot in Fig. 8(a) presents the initial nonlinear (close to exponential or hyperbolic) dependence of retention times of acetophenone on the methanol content in the eluent. The recurrent approximation of these data (Fig. 8(b)) is linear with R = 0.99999. Such *R*-value means that 99.999% of this function is a linear component, and only 0.001% is its distortion.



Fig. 8: (a) Plot of the acetophenone retention time (min) vs. methanol concentration in the eluent, $t_R(C)$: 32.464(30), 18.244(40), 11.230(50), 7.774(60), 6.052(70), and 5.226(80); (b) recurrent approximation of these data: $t_R(C + 10\%) = at_R(C) + b$. Parameters of the linear regression: $a = 0.4932 \pm 0.0005$, $b = 2.231 \pm 0.008$, R = 0.99999, $S_0 = 0.01$ (no outliers)

Let us combine two theses. *First*, some organic compounds form hydrates and some of these hydrates are, for example:

Compound	CAS no. of anhydrous form	CAS no. of hydrate
Glycine	56-40-6	130769-54-9
Oxalic acid	144-62-7	856335-90-5 (mono) 6153-56-6 (di)
Toluene	108-88-3	112270-21-0
Phenol	108-52-2	217182-78-0 144796-97-4
Nicotinamide	98-92-0	917925-73-6
Caffeine	58-08-2	5743-12-4
Methane (!)	74-82-8	14476-19-8

CAS numbers obviously are different both for anhydrous and hydrated forms of the compounds mentioned. Such a number for a hydrate does not confirm its stability or isolation; it is the confirmation of interest in this hydrated form. It is interesting that this list even includes the simplest hydrocarbon, methane. Second, the separation of analytes in RP HPLC takes place in water-containing eluents, when the probability of hydrate formation is high.

To identify the area in which anomalies should be expected, let us look again at Fig. 6. So far as the formation of hydrate of potassium acetate is most probable at low temperature of the aqueous solution, one anomalous leftmost point corresponds to the minimal values of solubility, w(T). In HPLC eluents, the hydrate formation is most probable at the maximal water content of an eluent. Hence, the weird point in the $t_{\rm R}(C)$ plot should be located on the right. The plot in Fig. 9 represents the recurrent approximation of the set of retention times $t_{\rm B}(C)$ for 2-methoxybenzaldehyde oxime at retention times within the range of approx. 5-40 min. Four points belong to the single straight line with the correlation coefficient R = 1.000, but the rightmost point deviates downward from the regression line [10]. The formation of hydrates was confirmed experimentally for some oximes of aromatic carbonyl compounds [23, 24].



Fig. 9: The plot illustrating the anomaly (the rightmost point below the regression line) in the recurrent approximation of the retention times for 2-methoxybenzaldehyde oxime. Parameters of the linear regression (without outlier): a = 0.4763 ± 0.0006 , b = 2.144 ± 0.007, R = 1.000, S₀ = 0.006.

It is worth noting that deviations of the rightmost points in the plots of recurrent approximations of retention times in RP HPLC were detected for the first time for complex organic compounds, namely, synthetic antitumor sulfonamide drugs containing the polar functional group -SO₂-NRR' [9]. Most sulfonamides form stable hydrates, which can be isolated in a solid form [25-28].

Thus, the recurrent approximation of retention times in RP HPLC allows us to detect the reversible formation of hydrates of analytes. The possibility is all the more important considering that detecting hydrates using other methods is much more difficult.

IV. CONCLUSION

This manuscript provides a brief exploration of the applications of recurrent relations in chemistry and chromatography, acknowledging that it cannot encompass all possible examples. Nonetheless, the review highlights the significant potential of recurrent

relations in simplifying calculations through the linearization of diverse chemical dependencies.

The linearization achieved by recurrent relations not only streamlines mathematical manipulations but also reveals valuable insights from deviations in chemical dependencies. For instance, in reversedphase high performance liquid chromatography (RP HPLC), non-linear variations in retention times can indicate hydration phenomena during analyte separation.

Formally, two alternative conclusions emerge from our discussions. Firstly, a broad hypothesis posits that many monotonic chemical variables follow firstorder recurrent dependencies - a potential general law in chemistry. However, a more modest yet preferable conclusion emphasizes the exceptional extrapolation capabilities of recurrent relations, which justify their widespread adoption in approximating a wide array of chemical variables.

In conclusion, to make a reader smile, one humorous application of recurrences should be mentioned. There is a very nonlinear dependence of the time of drop falling of liquid (including wine) from an empty bottle, t(n), on the serial number of the drop (n). No idea concerning the analytical form of this function can be proposed (or, at least, it is rather difficult). However, the recurrent representation of this dependence, t(n+1) = at(n) + b, demonstrates the excellent linearity [5].

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Medicinal Mushroom Biotechnology

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Abstract- Since most of medicinal mushrooms are rare in nature production of fungal fruiting bodies using artificial cultivation in a form of farming has been intensively established during the last 40 years. Solid state cultivation of various medicinal mushroom mycelia in various types of bioreactors, suitable for veterinary use, appears slightly in last few decades. Developing submerged technologies, using stirred tank and air lift bioreactors, are the most promising technologies for fast and large cultivation of medicinal pharmaceutically active products for human need. This potential initiates the development of new drugs and some of the most attractive over the counter human and veterinary remedies. This article is an overview of the engineering achievements in comprehensive medicinal mushroom mycelia cultivation.

Keywords: medicinal mushrooms, cultivation techniques, bioreactors.

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Medicinal Mushroom Biotechnology

Marin Berovic ^a, Bojana Boh ^o & Andrej Gregori ^e

Abstract- Since most of medicinal mushrooms are rare in nature production of fungal fruiting bodies using artificial cultivation in a form of farming has been intensively established during the last 40 years. Solid state cultivation of various medicinal mushroom mycelia in various types of bioreactors, suitable for veterinary use, appears slightly in last few decades. Developing submerged technologies, using stirred tank and air lift bioreactors, are the most promising technologies for fast and large cultivation of medicinal pharmaceutically active products for human need. This potential initiates the development of new drugs and some of the most attractive over the counter human and veterinary remedies. This article is an overview of the engineering achievements in comprehensive medicinal mushroom mycelia cultivation.

Keywords: medicinal mushrooms, cultivation techniques, bioreactors.

I. INTRODUCTION

total of more than 200 medicinal functions are thought to be produced by medicinal mushrooms (MM) and fungi including antitumor, immunomodulating, antioxidant, radical scavenging, cardiovascular. cholesterol-lowering. antiviral. antibacterial, anti-parasitic, antifungal, detoxification, hepatoprotective, anti-diabetic, anti-obesity, neuroprotective, neuroregenerative, and other effects. Also, substances derived from MM can be used as painkillers or analgetics [4, 5, 6]. The best implementation of MM drugs and dietary supplements has been in preventing immune disorders and maintaining a good quality of life, especially in immunodeficient and immuno-depressed patients, patients under chemotherapy or radiotherapy, patients with different types of cancers, chronic bloodborne viral infections of Hepatitis B. C. and D. different types of anemia, the human immunodeficiency virus/acquired immunodeficiency syndrome (HIV/AIDS), herpes simplex virus (HSV), Epstein Bar virus, Influenza viruses A and B, H5N1 [7], COVID-19 [8-10], West Nile virus, chronic fatigue syndrome, patients with chronic gastritis and gastric ulcers caused by Helicobacter pylori, and people suffering from dementia (especially Alzheimer's disease) [1, 2, 3, 5].

To combat these threats, humankind is focusing more and more attention on mushrooms and mushroom products. Mushrooms themselves are consumed regularly as part of the human diet and are treated as healthy or functional foods. On the other hand, the term mushroom nutriceuticals or dietary supplements has been applied to products derived from medicinal mushrooms that are taken to enhance general health and fitness but are not a regular food but a dietary food supplements [1].

II. Main Pharmaceutically Active Compounds

The main components of these supplements are polysaccharides, triterpenes and immunomodulatory proteins. Polysaccharide components, in particular, have been widely investigated as a source of anti-tumor and immunostimulating agents. They are widely distributed in mushrooms, with over 660 species from 183 genera reported to contain pharmacologically active polysaccharides. About 77% of all medicinal mushroom products are derived from the fruiting bodies, which have been either commercially farmed or collected from the wild, 21% from culture mycelium and 2% from culture broths. Precisely how these products work is still a matter of conjecture, but numerous laboratory animal tests as well as human clinical trials have shown them to be effective. In some cases, attention has focused on a bioactive mushroom component and its single effectiveness in treating specific disease conditions, much like a pharmaceutical. In the case of nutriceuticals/dietary supplements, emphasis has been placed on a combination of components that collectively impact on an individual's overall health and guality of life [2].

Many such products are currently available, and their market value worldwide increased from 1.2 billion in 1991 to 3.6 billion USD in 1994. The combined market value of medicinal mushrooms, mushroom extracts and derived products in 1999 was estimated to be 6.0 billion USD. That year, the United States nutraceutical market alone was valued at 35 million USD. Since then, demand has increased between 20% and 40% annually depending on the species, with *Ganoderma*-based dietary supplements alone valued at 1.6 billion USD. The MM industry has grown from small-scale (cottagebased) operations aimed at supplementing household incomes, to medium and mega-sized industrial

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ventures. This review examines the past, present and future of MM development, and includes a pyramid model addressing key issues [3].

They are of different chemical composition, such as polysaccharides, glycopeptide-protein complexes, proteoglycans, proteins and triterpenoids, with most scientific attention focussed to the group of non-cellulosic β -glucans with β -(1-3) linkages in the main chain of the glucan, and additional β -(1-6) branch points, that are characteristic for the antitumor and immuno-stimulating action. Mushroom polysaccharides do not attack cancer cells directly, but produce their antitumor effects by activating different immune responses in the host. Their mechanisms of action involve them being recognized by several immune cells receptors as non-self molecules, so the immune system is stimulated by their presence. Structurally different βglucans have different affinities toward receptors and thus generate different host responses [12].

Immunomodulating and antitumor activities of these metabolites are related to immune cells such as hematpoietic stem cells, lymphocytes, macrophages, T cells, dendritic cells, and natural killer cells, involved in the innate and adaptive immunity, resulting in the production of biologic response modifiers [13]. Clinical evidence for antitumor and other medicinal activities come primarily from some commercialised purified polysaccharides, such as lentinan from Shiitake -*Lentinula edodes*, krestin from *Coriolus versicolor*, grifolan from *Grifola frondosa*, and schizophyllan from *Schizophyllum commune* [12,14], but polysaccharide preparations of some other medicinal mushrooms also show promising results.

III. Wood Degrading *Basidiomycetes* with Pharmacological Effects

Almost unknown in Western Scientific Research only three decades ago, some of the wood degrading *Basidiomycetes* became intensely and systematically studied due to their promising pharmacological effects (Figures 1 A-E). Among them, *Ganoderma lucidum* – Ling zhi or Reishi, *Grifola frondosa* – Maitake, *Hericium erinaceus* – Lion mane and *Cordyceps militaris* have been known from the traditional Asian medicine, and *Trametes versicolor* Turkey tale, (previous *Coriolus versicolor*) are the subject of this review.



Figure 1: Ganoderma lucidum (A), Trametes versicolor (B) (previous Coriolus versicolor), Grifola frondosa (C), Hericium erinaceus (D) and Cordyceps militaris (E)

Technological possibilities for commercial cultivation gave rise to the number of patents, which are protecting inventions related to new methods and technologies for cultivation of fruit bodies and/or mycelium biomass, methods of active compounds isolation, and development of new commercial formulations and products. *Ganoderma lucidum* has been identified as a medicinal mushroom with the largest number of patented inventions [11].

a) Ganoderma Lucidum

Ganoderma (W. Curt.: Fr.) Lloyd is a white rot wood Basidiomycete that degrades lignin and possess hard fruiting body. G. lucidum is from Ganoderma sp. the most often reported as a source of various medicinal compounds. In Asian traditional medicine, the fruiting body of G. lucidum, named Ling-zhi in Chinese and Reishi in Japanese language, has been used for treatment of several diseases for thousands of years, as reported in Shen Nong's Materia Medica (Leung et.al., 2002; Kim and Kim, 2002; Lin, 2009).

Modern uses of *Ganoderma* include treatment of coronary heart diseases, arteriosclerosis, hepatitis, arthritis, nephritis, bronchitis, asthma, hypertension, cancer and gastric ulcer [15,16] Publications also report on *Ganoderma* antiallergenic constituents [17], immunomodulatory action [18-20], antitumor activity [21-23] cardiovascular effects [24], liver protection and detoxification, and effects on nervous system [22,25]. New reports emphasize its potential in treatment of viral, especially HIV infections [26, 27].

Pharmaceutically active compounds from Ganoderma lucidum include triterpenoids, polysaccharides $(1,6-\beta-D-glucans and 1,3-\beta-D-glucans)$, proteins, proteoglycans, steroids, alkaloids, nucleotides, lactones and fatty acids, amino acids, nucleotides, alkaloids, steroids, lactones and enzymes. Over 100 triterpenoids were found in Ganoderma spp., such as ganoderic (highly oxygenated C₃₀ lanostane-type triterpenoids), lucidenic, ganodermic, ganoderenic and ganolucidic acids, lucidones, ganoderals and ganoderols [11].

A large and diverse spectrum of chemical compounds with a pharmacological activity has been isolated from the mycelium, fruiting bodies and sclerotia mushrooms: of Ganoderma triterpenoids, polysaccharides, proteins, amino acids, nucleotides, alkaloids, steroids, lactones, fatty acids and enzymes [28, is 29]. There abundant evidence that polysaccharides isolated from G. lucidum are immunomodulatory effective [4, 30-34,]. Studies have shown that the most active immunomodulatory polysaccharides are $(1\rightarrow 3)$ - β -D and $(1\rightarrow 6)$ - β -D glucans, that can be precipitated by ethanol. Their prevailing structure is β -1.3 D-glucopyronan with 1–15 units of β -1,6 monoglucosyl side chains. Their 1.3-linked backbone, relatively small side chains, and an organized helical structure are beneficial for the immunostimulation. Although they are chemically heterogeneous, these polysaccharides are usually termed as β -glucans [30-32, 36, 37].

Bioactive polysaccharides have been isolated from different sources of *G. lucidum*: basidiocarps, spores and from the mycelial biomass cultivated in liquid culture. Few have been isolated from the culture medium [37]. Though different models of the fungal cell wall differ somewhat, scientists agree that β -glucan is not located on the surface of the wall but is more or less immersed in the wall material. Generally, the cell wall of most fungi contains five main components: $(1\rightarrow3)$ - β -D glucan, $(1\rightarrow6)$ - β -D glucan, chitin, and glycoproteins. β glucan forms 9–46 % of cell wall mass [36, 37]. A large number of studies have shown that polysaccharides of *G. lucidum*, especially β -D glucan, can modulate the functions of many components of the immune system such as the antigen-presenting cells, T and B lymphocytes, NK cells, neutrophil granulocytes, dendritic cells and, on cytokine production [30, 32, 38, 39].

b) Grifola Frondosa

Grifola frondosa also known as Maitake is white rot lignin degrading Basidiomycete with excellent nutritional and medicinal properties. Grifola frondosa active compounds primarilly belong to the group of polysaccharides (especially $1,6-\beta$ -D-glucans and $1,3-\beta$ -D-glucans), glycoproteins, and proteins. This products have been used for treatment of a series of diseases, including hepatitis, arthritis, nephritis, bronchitis, coronary heart diseases, asthma, arteriosclerosis, hypertension, cancer and gastric ulcer. Newer investigations report on Grifola frondosa antiallergenic constituents, immunomodulatory action and treatment of HIV infections, antitumor and cardiovascular effects, liver protection and detoxification and effects on nervous system [40].

G. frondosa has gained in popularity among consumers, not only because as a gastronomic delight of its taste and flavor, but also because of its reported medicinal value. Its active compounds primarily belong to the group of polysaccharides (especially 1,6-β-Dglucans and $1,3-\beta$ -D-glucans), glycoproteins, and proteins. Medicinal effects of G. frondosa are numerous, including anti-cancer activity [41, 42], immune system stimulation [43,44], effects on angiogenesis [45], reduction of benign prostatic hyperplasia [46]. antibacterial [47], and antiviral effects [48], effects on lipid metabolism and hypertension [41], antidiabetic activity [49], vitality and performance enhancement [50], antioxidant effects, and beneficial cosmetic effects on skin [51]. According to Shen (2001), more than 20 antitumor polysaccharides have been isolated and purified from G. frondosa [52].

c) Hericium Erinaceus

In *H. erinaceus* various pharmaceutically active substances were found. Phytosterols (β -sitosterol and ergosterol), lower the content of low density lipoproteins (LDL) and triglycerides that operate anticarcinogenic as well they reduce the metabolism of fats [53]. In *H. erinaceus* fruitbody numerous constituents such as are polysaccharides, proteins, lectins, phenols, hericenones, erinacines and terpenoids were identified. They strengthen the immune system, relieve gastritis and gastrointestinal infections, reflux and upset stomach due to stress [54].

H. erinaceus water-soluble polysaccharides increased activity of macrophages and other immune cells in the fight against cancer cells, but also demonstrated the reduction of formation of metastases. The most outstanding activity of the extract of *H. erinaceus* is that it strengthens the immune system and activate the synthesis of nerve growth factor [55]. Due to the increased proliferation of T and B-lymphocytes it strengthens the immune system and strengthens the body's natural defences Thus, *H. erinaceus* expresses very positive effects on prolongation of quality of life of the cancer patients [53].

Among the compounds isolated from fruiting bodies and cultured mycelia of *H. erinaceum*, most interesting are the low-molecular-weight compounds belonging to a group of cyathin diterpenoids (erinacines A–K, P, and Q). Several of them, i.e. erinacines A–H, are known to have a potent stimulating effect on nerve growth factor (NGF) synthesis *in vitro* [56-61].

H. erinaceus polysaccharides (HEP) derived from fruiting bodies and mycelium severed as effective therapeutic agents in liver damage-associated diseases. A study demonstrated that the serum levels of aspartate aminotransferase and glutamic pyruvic transaminase activities in carbon tetrachloride-induced hepatic injury were decreased by administration with extracellular and intracellular HEP (200, 400, and 800 mg/kg/day) from mycelium, but the blood lipid levels in the serum of mice were enhanced [62]. Moreover, Kim et al. found that 10 mg/kg/day HEP markedly alleviated Salmonella typhimurium-induced liver damage and reduced infected mice mortality [63]. Zhang et al. revealed that endo-HEP potent hepatoprotective effect in vivo, which may be due to its powerful antioxidant capacity. Taken together, the HEP could be exploited as a supplement in the prevention of hepatic diseases [64].

d) Trametes Versicolor

Trametes versicolor, previous *Coriolus versicolor*, also known as Turkey tail mushroom is one of the most attractive medical fungi. It is known for its use and success as a remedy in Asian traditional medicine [65-67]. *T. versicolor* pharmaceutical activities include immunomodulation, antibody production, activation of apoptosis etc.

The two most prominent products of T. versicolor are polysaccharide Krestin (PSK) and polysaccharide peptide (PSP) both potentially highly active pharmaceutical substances in complementary cancer treatments. PSP has a variety of physiological as immunological effects. such enhancement, antitumor, liver protecting, oxidation resistance, and reducing blood fat. PSP has been clinically used in treating cancers, hepatitis, hyperlipidemia, chronic bronchitis, and other diseases. The clinical data also demonstrate that PSP has diverse functions such as improving the quality of patients' life, enhancing learning and memory, and antiulcer effects [70-73].

Current studies support PSP as an immunotherapeutic. PSP activates and enhances the function and recognition ability of immune cells, strengthens the phagocytosis of macrophages, increases the expressions cytokines of

and chemokines such as tumor necrosis factor-a (TNF- α), interleukins (IL-1 β and IL-6), histamine. and prostaglandin E, stimulates the filtration of both dendritic cells and T-cells into tumors, and ameliorates the adverse events associated with chemotherapy. In recent years, immunotherapy has been widely used in cancer However, use PSP treatment. to as an immunotherapeutic at world stage, further chemical, biochemical and pharmacological studies of PSP are needed [66].

In vitro and *in vivo* studies have shown that the mixture of PSP and PSK has a synergistic action that highly affects immune cell proliferation and highly expresses antitumor activities [65, 67, 68, 74-76].

e) Cordyceps Militaris

The usage of natural/herbal medicines over the synthetic ones has seen an upward trend in the recent past. *Cordyceps* being an ancient medicinal mushroom used as a crude drug for the welfare of mankind in old civilization is now a matter of great concern because of its unexplored potentials obtained by various culture techniques and being an excellent source of bioactive metabolites with more than 21 clinically approved benefits on human health [77-78].

The studies by many researchers in the past on *Cordyceps* have demonstrated that it has antibacterial, anti-fungal, larvicidal, anti-inflammatory, antidiabetic, anti-oxidant, anti-tumor, pro-sexual, apoptotic, immunomodulatory, anti-HIV, remarkable clinical health effects including action on hepatic, renal, cardiovascular, respiratory, nervous, besides having anti-cancer, anti-oxidant, anti-inflammatory and antimicrobial activities [79-82].

Cordycepin has received much attention due to its broad-spectrum biological activity. It is known to interfere with various biochemical and molecular processes including purine biosynthesis [83, 84], DNA/RNA synthesis [85] and mTOR (mammalian target of rapamycin) signaling transduction [86]. *Cordyceps* has been included as one of the growing numbers of fungal traditional Chinese medicine (FTCM) used as cures for modern diseases with many products available commercially.

Great potential of pharmaceutical active compounds and Cordyceps militaris extract contains many biological bioactive materials, such as the terpenoids cordvcepin and cordvcepic acid. polysaccharides, sterols and other compounds [87]. Cordyceps militaris main active component is terpenoid cordycepin that inhibit the development of cancer cells including antitumor, anti-metastatic, insecticidal, antiproliferative, anti-bacterial properties, anti-fungal, larvicidal, anti-inflammatory, anti-diabetic, anti-oxidant, pro-sexual, apoptotic, immunomodulatory, anti-HIV, remarkable clinical health effects including action on hepatic, renal, cardiovascular, respiratory, nervous, besides having anti-cancer, anti-oxidant, antiinflammatory and anti-microbial activities, anti-leukemia and antimalarial activities [77-82].

The second main active components are polysaccharides, which research have shown to be effective in regulating blood sugar and also have antimetastatic and antitumor properties [87-89]. The most outstanding active substance of *C. militaris* is Cordycepin. The structure of Cordycepin is very much similar with cellular nucleoside, adenosine. Cordycepin, i.e., 3'-deoxyadenosine, is the main active constituent which is most widely studied for its medicinal value having a broad spectrum biological activity and acts like a nucleoside analogue [90].

Cordycepin alone has been widely explored for its anti-cancer/anti-oxidant activities, thus, holding a strong pharmacological and therapeutic potential to cure many dreadful diseases in future. Further investigations need to be focused on to study the mechanistic insight into the mysterious potential of this medicinal mushroom on human health and promoting its cultivation strategies for commercialization and ethno-pharmacological use of this wonderful herb [91, 92].

IV. Cultivation Technologies

Since medicinal mushrooms are scarce in nature, cultivation of fruit bodies on artificial media has been introduced. Traditional cultivation of fruit bodies on wood logs has been known for centuries. With time, cultivation methods have diversified, modified and developed (Figure 2). [11].

Besides on wood logs, fruit bodies are being produced on sawdust substrates in trays or beds, and in sterilised plastic bags or in bottles. In addition, production of fungal mycelia has been developed in bioreactors, utilizing submerged cultivation in liquid media, or solid state cultivation on various secondary wastes substrates from wood and agricultural industry [93].





a) Farming Fruit Bodies Cultivation

In the wild, wood degrading mushrooms grow primarily on hardwood of trees. However, under artificial cultivation conditions, they thrive on various other substrates containing lignin and cellulose, and therefore have a high potential for recycling different types of organic waste materials from wood and agriculture industry. As lignocellulose containing wastes are produced worldwide in large quantities, and in many instances they pose a threat to the environment. Cultivation of edible and medicinal wood degrading fungi on lignocellulosic substrates offers almost unlimited possibilities and economically viable potentials for large scale commercial cultivation on a World scale [94].

In farming fruit bodies cultivation substrate cultivation methods are divided mostly into bottle cultivation and bag cultivation, sometimes also called synthetic logs. Bag cultivation has more advantages, such as the use of more substrate, a strong body and convenient manipulation, so it is more widely used. Both production processes include the following main steps: raw material preparation, mixing, bagging (bottling) and sterilization, inoculation, spawning, embedding in soil (or transfer to mushroom house), fruiting body development, management and harvesting [94].



Figure 3: Farming production of Ganoderma lucidum fruit boddies (Photo A.Gregori)

For a production of G. lucidum fruit bodies supplemented sawdust is performed in heat-resistant polypropylene bottles or bags. Sawdust can be supplemented with rice bran (10%) and $CaCO_3$ (3%), moistened with water and filled (700 g) into plastic bags. A plastic collar is then fitted onto each bag and stopped with a cotton plug. After 5h of heat treatment (95-100°C) and cooling, substrate is inoculated with grain or sawdust spawn, and incubated for 3 to 4 weeks (or until the spawn fully colonizes the substrate) [95].

b) Solid State Cultivation

Solid state cultivation (SSC) is a three-phase heterogeneous process taking place in various bioreactors, comprising solid, liquid and gaseous phases, which offers potential benefits for the microbial cultivation for bioprocesses and products development.

Microbial growth on solid state substrate particles is very close to the growth of fungi in the natural environment. Main source of water, carbohydrates, phosphorous, nitrogen and sulphur are intrapartically bounded, therefore the microbial culture applied have to posses the abilities to access the water and essential element sources out the solid matrix Concerning that on the tips of young growing hiphae fungal polysaccharides are produced. Polysaccharide gel has actually two functions. Primary it serve as a gel media where from lignocellulitic enzymes are secreted in the solid wooden matrix and secondary as a sticky material used for anchoraging of hyphae and for moving on the solid surface. Produced fungal polysaccharides have also a whole palette of their pharmaceutical activities that are used in traditional Eastern medicine already for centuries [96].



Figure 4: Growth of *Hericium erinaceus* on solid substrate 124 h (D), 145 h (A), 234h (B) and 280 h (C), (Photo M. Vittori)

SSC in bioreactors involves the growth and metabolism of microorganisms in fully control environment. Mycrobial growth in takes place in aerated beds of moist solid materials in which the interparticle spaces contain a continuous gas phase and little or no free water. The upper limit of moisture content for solid state cultivations is determined by the absorbancy of the solid, which varies between substrates, although for most substrates a free water becomes apparent before 80% moisture level is reached. Although fungal mycelia growth in SSC is very close to the growth in nature habitats fungal fruit boddies are not produced in this technology [93]. An example of SSC mycelia growth in presented in Figure 4.

Over the last two decades, SSC has gained significant attention for the development of industrial

bioprocesses, particularly due to lower energy requirement associated with higher product yields and less wastewater production with lesser risk of bacterial contamination.

An important advantage of solid state cultivation over other techniques is that a concentrated product can be obtained from a cheap substrate, such as wood and agricultural secondary residue with little pretreatment or enrichment. For this reason, solid state cultivation seems to be most appropriate for the production of pharmaceutically active animal feed supplements, for which the whole fermented substrate can be used as the product [97]. Results of solid state medicinal mushroom cultivation on various substrates are presented in Table 1.

Table 1: Solid state cultivation of various medicinal mushroom in 15 L horizontal stirred tank bioreactor (HSTR). Own results

Fungus	Substrate	Biomass (mg/g)	Intracellular IPS (mg/g)
Ganoderma lucidum*	Beech saw-dust	68	7.45
Grifola frondosa	Beech saw-dust	54	4.70
Trametes versicolor	Corn straw	83.5	5.95
Hericium erinaceus	Husked and paddy millet	350	3.07
Cordyceps militaris	Husked paddy millet+ rice	236	10.42

*European strain MZKI G93

An important advantage of solid state cultivation over other techniques is that a concentrated product can be obtained from a cheap substrate, such as an agricultural residue with little pretreatment or enrichment. On the other hand, the use of an undefined medium, such as sawdust, might make the product purification more difficult. For this reason, solid state cultivation seems to be most appropriate for the production of pharmaceutically active animal feed supplements, for which the whole fermented substrate can be used as the product [98].

In recent years, substantial credibility in employing solid state cultivation (SSC) technique has been witnessed owing to its numerous advantages over submerged bioprocessing (SC). In spite of enormous advantages, true potential of SSC technology has not yet been fully realized at industrial scale [96].

C) Submeged Cultivation

Submerged cultivation of mushrooms represent the best and the fastest technology for a large scale production of medicinal mushroom mycelia and their products for a human use. In recent years, its submerged cultivation has received great interest in Asian countries as a promising alternative for efficient production of medicinal mushroom mycelia and its valuable metabolites [99].



Figure 5: Growth of Hericium erinaceus in submerged cultivation. (A) fungal mycelia (14 h)(magnification 1000x); (B) mycelia clumps (218 h) (400x), (C) mycelia pellets (262 h) (100x), (D) mycelia pellets (346 h) (100x) (Photo M.Vittori)

Mycelial growth and the results of submerged medicinal mushroom cultivation of five species on various substrates are presented in Figure 5 and Table 2.

The problems in submerged cultivation of fungal biomass increase with increase the broth viscosity during cultivation because of changes in the fungal morphology, biomass and extracellular polysaccharide (EPS) production. Therefore one of the most important factors of large-scale submerged cultures in bioreactors is related to the heath and mass transfer liquid phase oxygen supply. It is necessary to characterize the variations that occur during the

submerged cultivation in bioreactors and their effects on growth and product formation [100,101].

Species	Substrate	Bioreactor	Products	
Grifola frondosa	Optimized medium for mycelial biomass: 45.2 gL^{-1} glucose, 2.97 gL ⁻¹ KH ₂ PO ₄ , 6.58 gL ⁻¹ peptone Optimized medium for extracellular polysaccarhides 58.6 gL ⁻¹ glucose, 4.06 gL ⁻¹ KH ₂ PO ₄ 3.79 gL ⁻¹ peptone	15 L STR bioreactor innoculum 10% (v/v), T 25 °C, initial pH 5.5, aeration rate 8.0 vvm, agitation speed 80 rpm	Biomass 22.50 gL ⁻¹ Extracellular polysaccarhides 1.32 gL ¹	[103]
Ganoderma lucidum	Potato dextrose 101.2 gL ⁻¹ glucose 2 % olive oil	10 L STR bioreactor innoculum 10% (v/v) T 30 °C, initial pH 5.8, aeration rate 1.0 vvm, agitation speed 300 rpm	Biomass 15.9 gL ⁻¹ Extracellular polysaccarhides 9.6 gL ⁻¹ Intracellular polysaccarhides 6.3 gL ⁻¹	[104]
Hericium erinaceus	30 gL ⁻¹ corn flour 10 gL ⁻¹ glucose 3.0 gL ⁻¹ yeast extract 1.0 gL ⁻¹ KH ₂ PO _{4,} 0.5 gL ⁻¹ CaCO ₃ 15 mL of corn steep liquor	15 L STR bioreactor innoculum 10% (v/v), T 28 °C initial pH 5.7 aeration rate 0.8 vvm, agitation speed 80 rpm	Biomass 20.5 gL ⁻¹ Extracellular polysaccarhides 4.25 gL ⁻¹	[105]
Trametes versicolor	35 gL ⁻¹ glucose 0.5 gL ⁻¹ yeast extract, 5.0 gL ⁻¹ pepton 1.0 gL ⁻¹ KH ₂ PO ₄ 0.5 gL ⁻¹ MgSO ₄ x 7 H ₂ O 0,05 gL ⁻¹ tiamin	10 L STR inoculum 10% (v/v) , T 28 °C initial pH 5.7 aeration rate 1.0 vvm, agitation speed 400 rpm	Biomass 18,5 gL ⁻¹ Extracellular polysaccharide 3.8 gL ⁻¹	[106]
Cordyceps militaris	80 gL ⁻¹ glucose, 10 gL ⁻¹ yeast extract, 0.5 gL ⁻¹ MgSO ₄ ·7H ₂ O 0.5 gL ⁻¹ KH ₂ PO ₄	5 L STR bioreactor T 24 °C, pH 5.8, agitation 200 rpm, aeration 1.5 vvm,	Biomass 40.60 gL ⁻¹ Extracellular polysaccharide 6.74 gL ⁻¹	[107]

Table 2: Results of submerged cultivation of five species of medicinal mushrooms in bioreactors

d) Submerged Cultivation of the other Species in Bioreactors

Besides the cultures present in this chapter, there are also some other medicinal mushrooms that were submerge cultivated in mostly in lab scale bioreactors.

Inonotus levis and *Agaricus nevoi* cultivation was proceed in a 10 L stirred-tank bioreactor using substrate based on glucose and corn steep liquor at pH 5.5. Agitation speed gradually increased from 50 to 300 rpm and T 28 °C. *I.levis* developed very rapidly and after 5 days of cultivation the culture reached the stationary phase of growth with a high level of mycelia biomass of 16 g/L at level of EPS concentration 4.2 g/L. *A. nevoi* was distinguished by a much lower growth rate and entered the stationary growth phase on day 10 with mushroom biomass 12 g/L and EPS of 3.9 g/L [102].

Submerged cultivation of *Agaricus brasiliensis* was studied in 1 L stirred tank reactor. Sucrose was found to be most effective for EPS production. Yeast extract was the best for EPS among the inorganic and organic nitrogen sources tested. The factorial experiment demonstrated that a temperature of 30 °C and a pH of 6.1 were the best for the EPS production.

Glucose 10 g/L, yeast extract 3 g/L , $K_{2}HPO_{4}$ 0.6 g/L and MgSO_{4} 0.3 g/L [108].

Armillaria mellea was cultivated on glucose 40 g/L and yeast extract based substrate in 5 L stirred tank reactor at 22°C; the two-stage aeration rate strategy (1.2 \rightarrow 0.6 vvm); 150 rpm, controlled pH 4.0, 6.65 g/L fungal biomass and 233.2 m g/L of extracellular polysaccharides with antioxidant properties were obtained [109].

Pleurotus pulmonarius was studied in submerged cultivation in a 2 L stirred-tank reactor. Substrate was composed by 20 g/L of brown sugar, 4 g/L rice bran, 4 g/L malt extract, and 4 g/L of yeast extract with initial pH of 5.5 Incubated at 28°C with agitation speed of 250 rpm and oxygen partial pressure of 30–40%. Maximum *P. pulmonarius* dry biomass production of 11.8 g/L was achieved after 3 days of cultivation [110].

Pleurotus saca was submerged cultivated on substrate consist of beer worth substrate (batch mode in 10-L stirred tank reactor. Agitation speed was 500 rpm and aeration 5 Lmin⁻¹ and pH 6.2, up to 48.5 g/L of dry biomass was obtained [111].

Pleurotus ostreatus was cultivated in a 20 L stirred tank bioreactor in a submerged process with enhanced glucan and dietary fibres content, using 57 g/L xylose and 37 g/L corn steep liquor. High yields 39.2 g/L of dry biomass was obtained [112,113].

Ganoderma lingzhi were studied in 5 L stirred bioreactor. The optimum conditions were an initial pH of 5.9, 20.0% DO and T 29° C. These conditions resulted in a triterpene acids (TA) of 0.31 g/L. Furthermore this optimized conditions were then successfully scaled up to a production scale of 200 L, and maximum TA production and productivity of 0.29 g/L and 0.05 g/L day⁻¹ were achieved. [114].

Differences between Solid State and Submerged e) Cultivation

Main difference and benefits of solid state and submerged medicinal mushroom cultivation are presented in Table 3.

Table 3: Comparison of solid-state and submerged cultivation [96]

Solid state cultivation	Submerged cultivation		
Some products can only be produced well under low moisture conditions. For other products, if the producing organisms require free water, solid state cultivation cannot be used.	A wide range of products can be produced, from a wide range of micoorganisms and fungi. Many products are produced best under submerged cultivation.		
The medium is relatively simple (eg. grain) and unrefined. It may contain all nutrients necessary for growth, or only require wetting with a mineral solution. Pretreatment can be as simple as cooking or grinding. However, the substrate composition and characteristics can be variable.	The medium often contains more highly processed ingredients and is therefore more expensive. Unprocessed ingredients may need processing to extract and solubilize the nutrients. With defined media good reproducibility is possible		
The low water availability helps to select and protect against growth of contaminants.	The water activity is usually very high and many contaminants can grow well.		
Media are concentrated and smaller bioreactors can be used, leading to higher volumetric productivities, even when growth rates and yields are lower.	Media are dilute and therefore occupy larger volumes, leading to lower volumetric productivities.		
High substrate concentrations can enable high product concentrations.	High substrate concentrations can cause rheological problems. Substrate feeding systems may be required.		
Aeration requires less power since pressures are lower. Gas transfer is easier since the particles have a large surface area.	High air pressures can be required. Gas transfer from the gas to liquid phase is slow and can be limiting.		
Mixing within particles is not possible, and growth can be limited by the diffusion of nutrients.	Vigorous mixing can be used, and diffusion of nutrients is usually not limiting.		
Ability to remove metabolic heat is restricted, leading to overheating problems.	High water content and more dilute nature makes temperature control easier.		
Process control can be difficult due to difficulties in making on-line measurements, and in measuring biomass. The addition of substances during the process is difficult.	Many on-line sensors are available and more are being developed. Additions of substances can be made to control the process.		
Downstream processing may be simpler since products are more concentrated. However, extracts can be contaminated with substrate components.	Downstream processing requires removal of large volumes of water, and is more expensive. However, with defined media, product purification may be easier.		
Liquid waste is not produced.	Usually large volumes of liquid wastes are produced.		
Growth kinetics and transport phenomena have received little attention and are poorly characterized.	Much kinetic and transport information is available in literature, which can guide reactor design and operation.		
Research results and information from the solid-state cultivation can be scaled-up, or even transferred and applied in liquid-state cultivation.	In scaling up fungal submerged cultivation processes, various technical problems need to be solved, such as an increased broth viscosity and oxygen supply.		
Solid-state cultivation of fungal mycelia is less labor intensive.	Submerged cultivation is more demanding and labor intensive.		

V. Down Stream Processing

Disruption of medicinal mushrooms by mechanical, chemical or enzymatic methods is greatly required for the efficient extraction of active compounds from them. In addition, ultrafine powder of medicinal mushroom by mechanical method can be used for functional food or dietary supplement. Other products used for pharmaceuticals are produced at the stages of extraction, fractionation and purification by varying techniques including microwave assisted extraction, membrane separation, adsorptive separation and chromatograpy. Recrystallization, lyophilisation, drying and formulation are used as final product treatments [115].

Current disruption methods can be classified into mechanical, chemical and enzymatic methods in terms of their principles and characteristics. Mechanical methods are often preferred due to short residence time and lower operating costs [116]. The most common mechanical means for disruption are bead mill and homogenizer [117]. Another most frequently used sample disruption method is air jet milling which uses high velocity jets of gas to impart energy to particles for size reduction. The main features of air jet mills include [118]. Extraction is the first step to separate the desired products from the raw materials. Nearly 80–85 % of all medicinal mushroom products are extracted from their fruiting bodies while only 15 % are derived from mycelium culture [119].

Solvent is one of the most important parameters for a successful extraction. Selectivity, solubility, cost and safety should be considered in selection of solvents. Alcohols (ethanol and methanol) are universal solvents for the extraction of natural products although their low selectivity.

In recent years, advanced and greener extraction methods such as supercritical fuid extraction, pressurized liquid extraction, ultrasound assisted extraction, microwave assisted extraction, pulsed electric field extraction and enzyme assisted extraction have also been applied for extraction of natural products, and they offer some advantages such as lower organic solvent consumption, shorter extraction time and higher selectivity. In particular, supercritical fluid extraction gains increasing attention due to its higher efficiency and greener characteristics.

A brief summary of the various extraction methods used for medicinal mushroom products is shown in Table 4 [120].

Applied Method	Solvent	Volume of consumed solvent	Temperature	Time	Pressure	Polarity of extracted products
Maceration	Various solvent	Large	Room temperature	Atmospheric	Long	Dependent on the solvent
Percolation	Various solvent	Large	Room temperature, sometimes under heat	Atmospheric	Long	Dependent on the solvent
Decoration	Water	None	Under heat	Atmospheric	Moderate	Polar compounds
Reflux extraction	Various solvent	Moderate	Under heat	Atmospheric	Moderate	Dependent on the solvent
Soxhlet extraction	Organic solvent	Moderate	Under heat	Atmospheric	Long	Dependent on the solvent
Pressurized liquid extraction	Organic solvent	Small	Under heat	High	Short	Dependent on the solvent
Supercritical fluid extraction	Supercritical fluid (usually CO ₂)	None or small	Near room temperature	High	Short	None polar or moderate polar
Ultrasound assisted extraction	Various solvent	Moderate	Room temperature or under heat	Atmospheric	Short	Compounds dependent on the solvent
Microwave assisted extraction	Various solvent	Moderate	Room temperature or under heat	Atmospheric	Short	Dependent on the solvent
Pulsed electric field extraction	Various solvent	Moderate	Room temperature or under heat	Atmospheric	Short	Dependent on the solvent

Table 4: A brief summary of various extraction method used for medicinal mushroom product adapted from [120]

Enzyme assisted extraction	Various solvent	Moderate	Room temperature or under heat	Atmospheric	Moderate	Dependent on the solvent
Distillation	Water	Moderate	Under heat	Atmospheric	Long	Essential oil

The obtained fraction containing the desired products via fractionation steps described above may consist of several compounds with highly similar chemical and physical properties. Sometimes these compounds are analoges or even isomers as already stated, presenting a huge challenge for the purification. Column chromatography packed with separation media with small particle size is the most prevailing technique to implement the task, because it can offer very high number of theoretical plates and thus high resolution. However, in industrial scale, the particle size of the packed material is much larger because no pump can generate the pressure in manufacturing scale as that in a HPLC system [120].

VI. Conclusion

Various pharmaceutically active substances from medicinal mushrooms represent effective potential in human life. Great demand for medicinal fungi biomass production could be fulfilled using various cultivatiom techniques. Medicinal fungi biomass in a present time is mostly covered by farming. Farming cultivation represents cheap but long time consuming technology. Using cultivation on a wooden locks a few year coughing time is need. Cultivation on sawdust substrates in trays or beds and in sterilised plastic bags or in bottles represents an advance and much faster production of fungal fruit boddies than conventional farming.

Solid state a few week time cultivation is a comprehensive, well controlled technology that enables much faster medium scale technology for medicinal mushroom mycelia production. In this technology various secondary waste from wood and agriculture industry are successfully used. No fungal fruit boddies are produced. Final product delignolized, wooden material is overgrown by medicinal fungi biomass in two to four weeks, enreach with proteins and various pharmaceutically products need to be dryed and pulverized and in such form it could be directly used in a veterinary need. Solid state cultivation of medicinal mushroom biomass is cheap in non time consuming technology perfectlly suitable in veterinary use.

Submerged cultivation of medicinal fungi biomass represents fast and comprehensive technology method. Fungal biomass is in its final state from 10 to 28 days. The main benefit of cultivation of medicinal mushroom biomass in bioreactors is in using of higher sterility, comprehensive technology and bioprocess control, for large scale production of various pharmaceutically active compounds as are fungal polysaccharides, terpenoides or proteinoglucans in shorter cultivation time. Reports much on pharmacological activity of extracts, partly purified preparations and isolated compounds from biomass of G.lucidum, G.frondosa, T.versicolor, H.erinaceus and C.militaris and the other reported species in laboratory and pilot scale, are very convincing. Production of medicinal fungi polysaccharides enhanced by fed-batch or two-stage cultivation strategy was found very useful for improving the production. Some of present results of lab scale research of various medicinal fungi are already transferred to pilot and lower industrial scale and they represents suitable starting platform for development of medicinal fungi biomass in large scale pharmaceutical industrial production. Comparing the economy of the same product production solid state processing is 30 % less than those of the one produced in submerged cultivation.

Isolation of medicinal fungi pharmaceutically compounds in all three technologies is is based on precipitation with hot water and ethanol. Crude extracts often show equal or stronger pharmacological activity as purified compounds, which suggests potential synergistic effects of several naturally occurring compounds. In the future, more precise capture (or enrichment) and separation techniques such as affinity separation, and more integrated bioprocesses for medicinal mushroom products should be developed which would enable a higher product yield and better process performance.

As pointed out submerged cultivation of medicinal mushrooms has significant large scale industrial potential, but its success on a commercial scale depends on existing field-cultivation technology as well as pharmaceutical market economy. Production of various medicinal mushrooms products, terpenoids, polysaccharides and proteinoglucans represents great business in Asiatic space where it is traditional. Great demand of this active ingredients definitively need to include new and fast large scale industrial production technologies as are solid state and submerged cultivations. In opposite, Western Pharmaceutical Industry has no tradition in natural isolates from herbal and fungal sources. Unfortunately, it is too convinient and much basesed on classical pharmaco-chemical biosynthesis including its all side effects.

However, from the viewpoint of Western Science, pharmaceutical legislation and regulations might be one of the main obstacles hindering the introduction of medicinal mushrooms products as registered pharmaceuticals. In any case, further research is needed to fully understand all mechanisms of pharmaceutical effects of medicinal mushrooms products and to identify their potential side effects.

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17. *Never copy others' work:* Never copy others' work and give it your name because if the evaluator has seen it anywhere, you will be in trouble. Take proper rest and food: No matter how many hours you spend on your research activity, if you are not taking care of your health, then all your efforts will have been in vain. For quality research, take proper rest and food.

18. Go to seminars: Attend seminars if the topic is relevant to your research area. Utilize all your resources.

19. Refresh your mind after intervals: Try to give your mind a rest by listening to soft music or sleeping in intervals. This will also improve your memory. Acquire colleagues: Always try to acquire colleagues. No matter how sharp you are, if you acquire colleagues, they can give you ideas which will be helpful to your research.

20. *Think technically:* Always think technically. If anything happens, search for its reasons, benefits, and demerits. Think and then print: When you go to print your paper, check that tables are not split, headings are not detached from their descriptions, and page sequence is maintained.

21. Adding unnecessary information: Do not add unnecessary information like "I have used MS Excel to draw graphs." Irrelevant and inappropriate material is superfluous. Foreign terminology and phrases are not apropos. One should never take a broad view. Analogy is like feathers on a snake. Use words properly, regardless of how others use them. Remove quotations. Puns are for kids, not grunt readers. Never oversimplify: When adding material to your research paper, never go for oversimplification; this will definitely irritate the evaluator. Be specific. Never use rhythmic redundancies. Contractions shouldn't be used in a research paper. Comparisons are as terrible as clichés. Give up ampersands, abbreviations, and so on. Remove commas that are not necessary. Parenthetical words should be between brackets or commas. Understatement is always the best way to put forward earth-shaking thoughts. Give a detailed literary review.

22. Report concluded results: Use concluded results. From raw data, filter the results, and then conclude your studies based on measurements and observations taken. An appropriate number of decimal places should be used. Parenthetical remarks are prohibited here. Proofread carefully at the final stage. At the end, give an outline to your arguments. Spot perspectives of further study of the subject. Justify your conclusion at the bottom sufficiently, which will probably include examples.

23. Upon conclusion: Once you have concluded your research, the next most important step is to present your findings. Presentation is extremely important as it is the definite medium though which your research is going to be in print for the rest of the crowd. Care should be taken to categorize your thoughts well and present them in a logical and neat manner. A good quality research paper format is essential because it serves to highlight your research paper and bring to light all necessary aspects of your research.

INFORMAL GUIDELINES OF RESEARCH PAPER WRITING

Key points to remember:

- Submit all work in its final form.
- Write your paper in the form which is presented in the guidelines using the template.
- Please note the criteria peer reviewers will use for grading the final paper.

Final points:

One purpose of organizing a research paper is to let people interpret your efforts selectively. The journal requires the following sections, submitted in the order listed, with each section starting on a new page:

The introduction: This will be compiled from reference matter and reflect the design processes or outline of basis that directed you to make a study. As you carry out the process of study, the method and process section will be constructed like that. The results segment will show related statistics in nearly sequential order and direct reviewers to similar intellectual paths throughout the data that you gathered to carry out your study.

The discussion section:

This will provide understanding of the data and projections as to the implications of the results. The use of good quality references throughout the paper will give the effort trustworthiness by representing an alertness to prior workings.

Writing a research paper is not an easy job, no matter how trouble-free the actual research or concept. Practice, excellent preparation, and controlled record-keeping are the only means to make straightforward progression.

General style:

Specific editorial column necessities for compliance of a manuscript will always take over from directions in these general guidelines.

To make a paper clear: Adhere to recommended page limits.



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Mistakes to avoid:

- Insertion of a title at the foot of a page with subsequent text on the next page.
- Separating a table, chart, or figure—confine each to a single page.
- Submitting a manuscript with pages out of sequence.
- In every section of your document, use standard writing style, including articles ("a" and "the").
- Keep paying attention to the topic of the paper.
- Use paragraphs to split each significant point (excluding the abstract).
- Align the primary line of each section.
- Present your points in sound order.
- Use present tense to report well-accepted matters.
- Use past tense to describe specific results.
- Do not use familiar wording; don't address the reviewer directly. Don't use slang or superlatives.
- Avoid use of extra pictures—include only those figures essential to presenting results.

Title page:

Choose a revealing title. It should be short and include the name(s) and address(es) of all authors. It should not have acronyms or abbreviations or exceed two printed lines.

Abstract: This summary should be two hundred words or less. It should clearly and briefly explain the key findings reported in the manuscript and must have precise statistics. It should not have acronyms or abbreviations. It should be logical in itself. Do not cite references at this point.

An abstract is a brief, distinct paragraph summary of finished work or work in development. In a minute or less, a reviewer can be taught the foundation behind the study, common approaches to the problem, relevant results, and significant conclusions or new questions.

Write your summary when your paper is completed because how can you write the summary of anything which is not yet written? Wealth of terminology is very essential in abstract. Use comprehensive sentences, and do not sacrifice readability for brevity; you can maintain it succinctly by phrasing sentences so that they provide more than a lone rationale. The author can at this moment go straight to shortening the outcome. Sum up the study with the subsequent elements in any summary. Try to limit the initial two items to no more than one line each.

Reason for writing the article-theory, overall issue, purpose.

- Fundamental goal.
- To-the-point depiction of the research.
- Consequences, including definite statistics—if the consequences are quantitative in nature, account for this; results of any numerical analysis should be reported. Significant conclusions or questions that emerge from the research.

Approach:

- Single section and succinct.
- An outline of the job done is always written in past tense.
- o Concentrate on shortening results—limit background information to a verdict or two.
- Exact spelling, clarity of sentences and phrases, and appropriate reporting of quantities (proper units, important statistics) are just as significant in an abstract as they are anywhere else.

Introduction:

The introduction should "introduce" the manuscript. The reviewer should be presented with sufficient background information to be capable of comprehending and calculating the purpose of your study without having to refer to other works. The basis for the study should be offered. Give the most important references, but avoid making a comprehensive appraisal of the topic. Describe the problem visibly. If the problem is not acknowledged in a logical, reasonable way, the reviewer will give no attention to your results. Speak in common terms about techniques used to explain the problem, if needed, but do not present any particulars about the protocols here.



The following approach can create a valuable beginning:

- Explain the value (significance) of the study.
- Defend the model—why did you employ this particular system or method? What is its compensation? Remark upon its appropriateness from an abstract point of view as well as pointing out sensible reasons for using it.
- Present a justification. State your particular theory(-ies) or aim(s), and describe the logic that led you to choose them.
- o Briefly explain the study's tentative purpose and how it meets the declared objectives.

Approach:

Use past tense except for when referring to recognized facts. After all, the manuscript will be submitted after the entire job is done. Sort out your thoughts; manufacture one key point for every section. If you make the four points listed above, you will need at least four paragraphs. Present surrounding information only when it is necessary to support a situation. The reviewer does not desire to read everything you know about a topic. Shape the theory specifically—do not take a broad view.

As always, give awareness to spelling, simplicity, and correctness of sentences and phrases.

Procedures (methods and materials):

This part is supposed to be the easiest to carve if you have good skills. A soundly written procedures segment allows a capable scientist to replicate your results. Present precise information about your supplies. The suppliers and clarity of reagents can be helpful bits of information. Present methods in sequential order, but linked methodologies can be grouped as a segment. Be concise when relating the protocols. Attempt to give the least amount of information that would permit another capable scientist to replicate your outcome, but be cautious that vital information is integrated. The use of subheadings is suggested and ought to be synchronized with the results section.

When a technique is used that has been well-described in another section, mention the specific item describing the way, but draw the basic principle while stating the situation. The purpose is to show all particular resources and broad procedures so that another person may use some or all of the methods in one more study or referee the scientific value of your work. It is not to be a step-by-step report of the whole thing you did, nor is a methods section a set of orders.

Materials:

Materials may be reported in part of a section or else they may be recognized along with your measures.

Methods:

- Report the method and not the particulars of each process that engaged the same methodology.
- o Describe the method entirely.
- To be succinct, present methods under headings dedicated to specific dealings or groups of measures.
- Simplify—detail how procedures were completed, not how they were performed on a particular day.
- o If well-known procedures were used, account for the procedure by name, possibly with a reference, and that's all.

Approach:

It is embarrassing to use vigorous voice when documenting methods without using first person, which would focus the reviewer's interest on the researcher rather than the job. As a result, when writing up the methods, most authors use third person passive voice.

Use standard style in this and every other part of the paper—avoid familiar lists, and use full sentences.

What to keep away from:

- Resources and methods are not a set of information.
- o Skip all descriptive information and surroundings—save it for the argument.
- Leave out information that is immaterial to a third party.



Results:

The principle of a results segment is to present and demonstrate your conclusion. Create this part as entirely objective details of the outcome, and save all understanding for the discussion.

The page length of this segment is set by the sum and types of data to be reported. Use statistics and tables, if suitable, to present consequences most efficiently.

You must clearly differentiate material which would usually be incorporated in a study editorial from any unprocessed data or additional appendix matter that would not be available. In fact, such matters should not be submitted at all except if requested by the instructor.

Content:

- o Sum up your conclusions in text and demonstrate them, if suitable, with figures and tables.
- o In the manuscript, explain each of your consequences, and point the reader to remarks that are most appropriate.
- Present a background, such as by describing the question that was addressed by creation of an exacting study.
- Explain results of control experiments and give remarks that are not accessible in a prescribed figure or table, if appropriate.
- Examine your data, then prepare the analyzed (transformed) data in the form of a figure (graph), table, or manuscript.

What to stay away from:

- o Do not discuss or infer your outcome, report surrounding information, or try to explain anything.
- Do not include raw data or intermediate calculations in a research manuscript.
- Do not present similar data more than once.
- o A manuscript should complement any figures or tables, not duplicate information.
- Never confuse figures with tables—there is a difference.

Approach:

As always, use past tense when you submit your results, and put the whole thing in a reasonable order.

Put figures and tables, appropriately numbered, in order at the end of the report.

If you desire, you may place your figures and tables properly within the text of your results section.

Figures and tables:

If you put figures and tables at the end of some details, make certain that they are visibly distinguished from any attached appendix materials, such as raw facts. Whatever the position, each table must be titled, numbered one after the other, and include a heading. All figures and tables must be divided from the text.

Discussion:

The discussion is expected to be the trickiest segment to write. A lot of papers submitted to the journal are discarded based on problems with the discussion. There is no rule for how long an argument should be.

Position your understanding of the outcome visibly to lead the reviewer through your conclusions, and then finish the paper with a summing up of the implications of the study. The purpose here is to offer an understanding of your results and support all of your conclusions, using facts from your research and generally accepted information, if suitable. The implication of results should be fully described.

Infer your data in the conversation in suitable depth. This means that when you clarify an observable fact, you must explain mechanisms that may account for the observation. If your results vary from your prospect, make clear why that may have happened. If your results agree, then explain the theory that the proof supported. It is never suitable to just state that the data approved the prospect, and let it drop at that. Make a decision as to whether each premise is supported or discarded or if you cannot make a conclusion with assurance. Do not just dismiss a study or part of a study as "uncertain."

Research papers are not acknowledged if the work is imperfect. Draw what conclusions you can based upon the results that you have, and take care of the study as a finished work.

- You may propose future guidelines, such as how an experiment might be personalized to accomplish a new idea.
- Give details of all of your remarks as much as possible, focusing on mechanisms.
- Make a decision as to whether the tentative design sufficiently addressed the theory and whether or not it was correctly restricted. Try to present substitute explanations if they are sensible alternatives.
- One piece of research will not counter an overall question, so maintain the large picture in mind. Where do you go next? The best studies unlock new avenues of study. What questions remain?
- o Recommendations for detailed papers will offer supplementary suggestions.

Approach:

When you refer to information, differentiate data generated by your own studies from other available information. Present work done by specific persons (including you) in past tense.

Describe generally acknowledged facts and main beliefs in present tense.

The Administration Rules

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Segment draft and final research paper: You have to strictly follow the template of a research paper, failing which your paper may get rejected. You are expected to write each part of the paper wholly on your own. The peer reviewers need to identify your own perspective of the concepts in your own terms. Please do not extract straight from any other source, and do not rephrase someone else's analysis. Do not allow anyone else to proofread your manuscript.

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Please note that following table is only a Grading of "Paper Compilation" and not on "Performed/Stated Research" whose grading solely depends on Individual Assigned Peer Reviewer and Editorial Board Member. These can be available only on request and after decision of Paper. This report will be the property of Global Journals.

Topics	Grades		
	А-В	C-D	E-F
Abstract	Clear and concise with appropriate content, Correct format. 200 words or below	Unclear summary and no specific data, Incorrect form Above 200 words	No specific data with ambiguous information Above 250 words
Introduction	Containing all background details with clear goal and appropriate details, flow specification, no grammar and spelling mistake, well organized sentence and paragraph, reference cited	Unclear and confusing data, appropriate format, grammar and spelling errors with unorganized matter	Out of place depth and content, hazy format
Methods and Procedures	Clear and to the point with well arranged paragraph, precision and accuracy of facts and figures, well organized subheads	Difficult to comprehend with embarrassed text, too much explanation but completed	Incorrect and unorganized structure with hazy meaning
Result	Well organized, Clear and specific, Correct units with precision, correct data, well structuring of paragraph, no grammar and spelling mistake	Complete and embarrassed text, difficult to comprehend	Irregular format with wrong facts and figures
Discussion	Well organized, meaningful specification, sound conclusion, logical and concise explanation, highly structured paragraph reference cited	Wordy, unclear conclusion, spurious	Conclusion is not cited, unorganized, difficult to comprehend
References	Complete and correct format, well organized	Beside the point, Incomplete	Wrong format and structuring

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